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C07D 231/22

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(22) Date of filing: **21.02.1990**

(54) **3-(Substituted phenyl)pyrazole derivatives, salts thereof, herbicides therefrom, and process for producing said derivatives or salts**

3-(Substituierte Phenyl)pyrazolderivate, deren Salze, deren Herbizide und Verfahren zur Herstellung dieser Derivate oder Salze

Dérivés de 3-(phényl substitué)-pyrazol, leurs sels, leurs herbicides, et les procédés de préparation de ces dérivés ou de leurs sels

(84) Designated Contracting States:
BE ES NL

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EP-A- 0 052 333 **EP-A- 0 361 114**
FR-A- 2 262 663 **GB-A- 2 073 172**

Remarks:

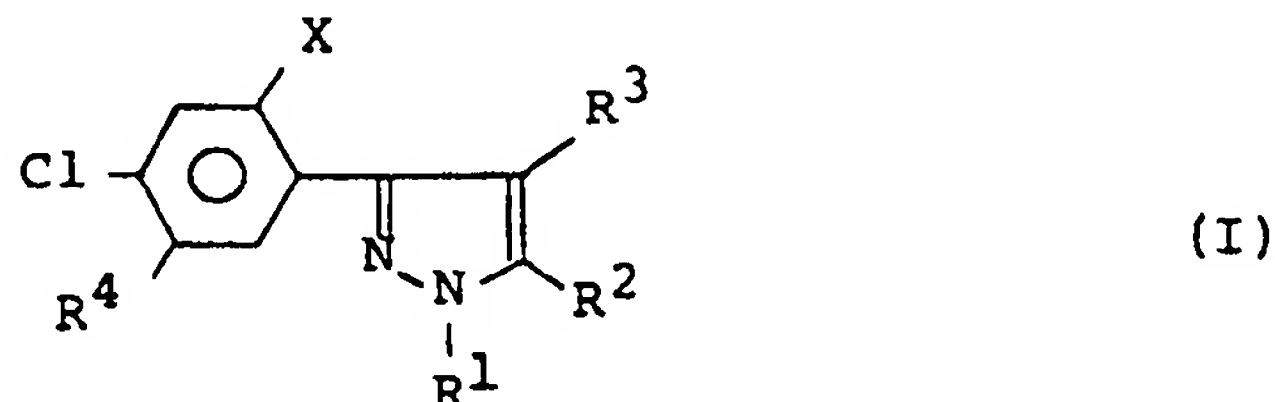
The file contains technical information submitted after the application was filed and not included in this specification

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Description

The present invention relates to 3-(substituted phenyl)pyrazole derivatives or salts thereof, a process for producing said derivatives or salts, and to herbicidal compositions comprising said derivatives or salts and methods for applying said herbicidal compositions. The 3-(substituted phenyl)pyrazole derivatives are represented by the general formula



wherein

X denotes halogen,

R¹ denotes CH₃,

R² denotes hydroxy, mercapto, alkylthio, haloalkoxy or haloalkylthio,

R³ denotes hydrogen or halogen, and

R⁴ denotes (i) formyl, (ii) nitro, (iii) -CO-B-R⁶ wherein B denotes -O-, -S-, or -NR⁷, R⁶ and R⁷ are the same or different and each denote hydrogen or cycloalkyl, and when B is -O-, R⁶ can be an alkali metal atom or -CO-B-R⁶

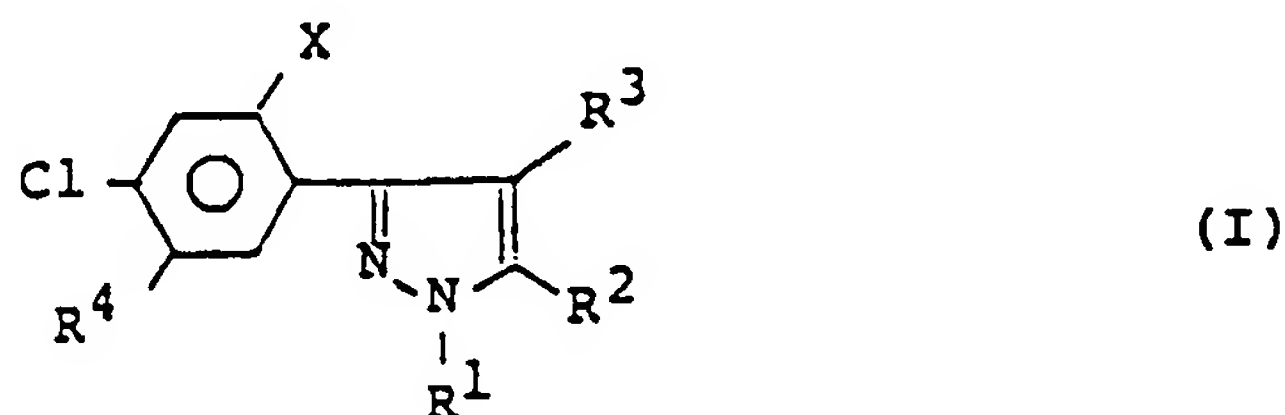
denotes methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, methoxycarbonylmethoxycarbonyl, ethoxycarbonylmethoxycarbonyl, propoxycarbonylmethoxycarbonyl, methoxycarbonylethoxycarbonyl, ethoxycarbonylethoxycarbonyl, or propoxycarbonylethoxycarbonyl, (iv) -D-R⁸ wherein D denotes -O-, -S(O)_n, n being an integer of 0 to 2, or -NR⁹, R⁸ and R⁹, which are the same or different, each denotes hydrogen; alkyl; haloalkyl;

aminosulfonyl; phenylalkyl or phenoxy-alkyl optionally having on the phenyl ring one or more halogen substituents which are the same or different; or -D-R⁸ denotes an alkenyloxy group containing propenyl, butenyl, or pentenyl;

an alkynyloxy group containing propynyl, butynyl or pentynyl; an alkenylamino group containing propenyl, butenyl or pentenyl; an alkoxy-carbonylalkoxy or an alkylthiocarbonylalkoxy group containing each methyl, ethyl, propyl or butyl; or (v) -(CHR¹⁰)_m-CO-E-R¹² wherein E denotes -O-, -S-, or -NR¹¹; R¹⁰ denotes hydrogen and R¹¹ and R¹², which are the same or different, each denotes hydrogen; alkyl; haloalkyl; phenyl optionally having on the phenyl ring one or more halogen substituents; phenylalkyl optionally having on the phenyl ring one or more halogen substituents; or R¹¹, jointly with R¹², forms a piperidino or morpholinyl group, and when E is -O-, R¹² can be an alkali metal atom;

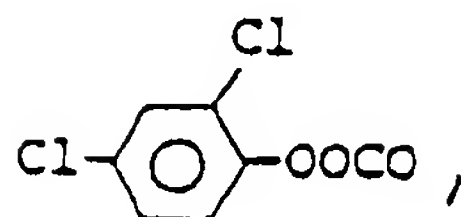
and m denotes an integer of 0 to 3.

Further, the present invention provides a 3-(substituted phenyl)-pyrazole derivative or a salt thereof, the derivative being represented by the general formula

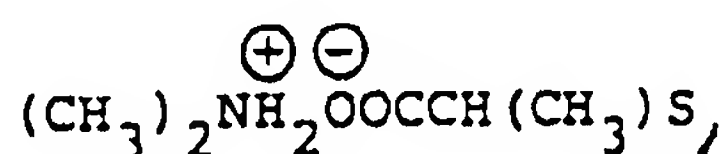
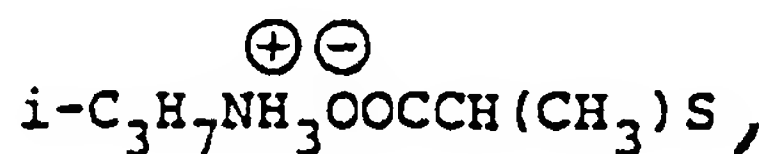


wherein

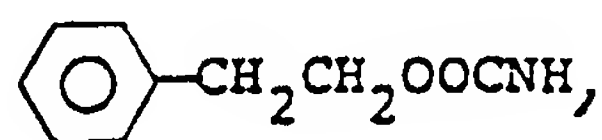
a) X is Cl, R¹ is CH₃, R² is CH₃S, R³ is Cl and R⁴ is selected from



HOOCCH(CH₃)O, CH₃OOCCH(CH₃)O, C₂H₅OOCCH(CH₃)O, i-C₃H₇OOCCH(CH₃)O, CH₂=CHCH₂OOCCH(CH₃)O, CH≡CCH₂OOCCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(Cl)CH₂S, CH≡CCH₂S, HO₂CCH(CH₃)S,



20 Cl(CH₂)₃OOCCH(CH₃)S, CH₂=CHCH₂OOCCH(CH₃)S, CH≡CCH₂OOCCH(CH₃)S, C₂H₅CH(CH₃)CH₂OOCCH(CH₃)S, (C₂H₅)₂NCOCH(CH₃)S, CH₃NHSO₂, (CH₃)₂NSO₂, n-C₃H₇NHSO₂, CH₃SO₂NHSO₂, Cl⁻H₃N⁺, CH≡CCH₂NH, CH₃OOCNH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃NHSO₂N(CH₃), (CH₃)₂NSO₂N(CH₃), i-C₃H₇NHSO₂NH, CH≡CCH₂N(CH₃)SO₂N(CH≡CCH₂), CH₃OCH₂CH₂OOCNH, BrCH₂CH₂OOCNH, NCCH₂CH₂OOCNH, CH₂=C(CH₃)CH₂OOCNH,



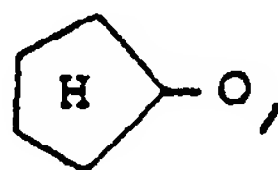
30 CH₂=CHCH₂OOCCH₂NH, CH≡CCH₂OOCCH₂NH, C₂H₅OOCCH(CH₃)NH, HOOCCH₂NH, CH₃OOCCH₂NH, HOOCCH(CH₃)NH and CH₃OOCCH(CH₃)NH;

35 b) X is Cl, R¹ is CH₃, R² is CH₃SO, R³ is Cl and R⁴ is selected from NO₂, CH₃SO₂NH, CH₃NHSO₂NH, i-C₃H₇NHSO₂NH and CH₃NHSO₂(CH≡CCH₂);

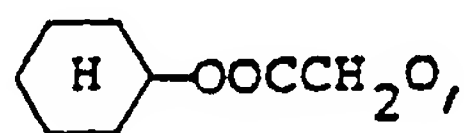
c) X is Cl, R¹ is CHF₂, R² is CH₃S, R³ is Cl and R⁴ is selected from NH₂, CH₃NHSO₂NH, CH₃NHSO₂N(CH≡CCH₂) and CH₃N(CH≡CCH₂)SO₂N(CH≡CCH₂);

40 d) X is Cl, R¹ is CHF₂, R² is CH₃SO, R³ is Cl and R⁴ is selected from NH₂ and NO₂;

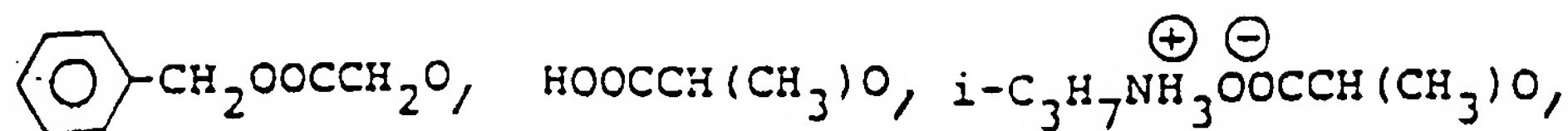
e) X is Cl, R¹ is CH₃, R² is CHF₂O, R³ is Cl and R⁴ is selected from



50 C₂H₅OCH₂CH₂O, NCCH₂O, CH₃OOCO, n-C₆H₁₃OOCCH₂O, CH₂=CHCH₂OOCCH₂O, CH≡CCH₂OOCCH₂O,



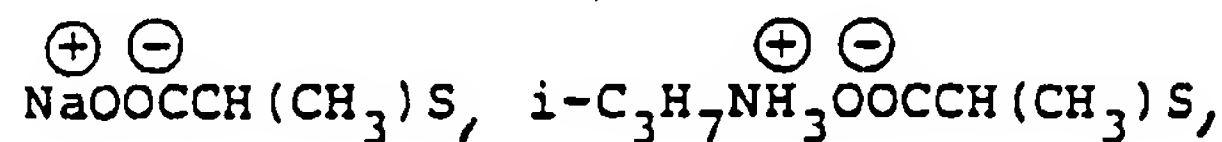
C₂H₅O(CH₂)₂OOCCH₂O, C₂H₅O(CH₂)₂O(CH₂)₂OOCCH₂O,



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$\text{CH}_2=\text{CHCH}_2\text{OOCCH}(\text{CH}_3)\text{O}$, $\text{CH}\equiv\text{CCH}_2\text{OOCCH}(\text{CH}_3)\text{O}$, $\text{CH}_3\text{NHCOCH}(\text{CH}_3)\text{O}$, $(\text{CH}_3)_2\text{NCOCH}(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{S}$, $\text{CH}_2=\text{CHCH}_2\text{SO}$, $\text{CH}\equiv\text{CCH}_2\text{S}$, $(\text{C}_2\text{H}_5\text{O})_2\text{P}(\text{O})\text{CH}_2\text{S}$, $\text{CH}_3\text{OOCCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{OOCCH}_2\text{S}$, $\text{CH}_2=\text{CHCH}_2\text{OOCCH}_2\text{S}$, $\text{HOOCCH}(\text{CH}_3)\text{S}$,

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$\text{CH}_3\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{C}_2\text{H}_5\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{C}_2\text{H}_5\text{OOCCH}(\text{CH}_3)\text{S}(\text{O})$, $\text{C}_2\text{H}_5\text{OOCCH}(\text{CH}_3)\text{SO}_2$, $\text{i-C}_3\text{H}_7\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{n-C}_4\text{H}_9\text{OOCCH}(\text{CH}_3)\text{S}$, $(\text{CH}_3)_2\text{CH}(\text{CH}_2)_2\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{CH}_2=\text{CHCH}_2\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{CH}\equiv\text{CCH}_2\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{O}(\text{CH}_2)_2\text{O}(\text{CH}_2)_2\text{OOCCH}(\text{CH}_3)\text{S}$, $(\text{C}_2\text{H}_5\text{O})_2\text{P}(\text{O})\text{CH}_2\text{OOCCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{NHCOCH}(\text{CH}_3)\text{S}$, $(\text{CH}_3)_2\text{NCOCH}(\text{CH}_3)\text{S}$, $(\text{C}_2\text{H}_5)_2\text{NCOCH}(\text{CH}_3)\text{S}$, $\text{i-C}_3\text{H}_7\text{NHCOCH}(\text{CH}_3)\text{S}$,

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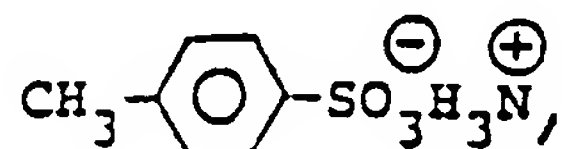
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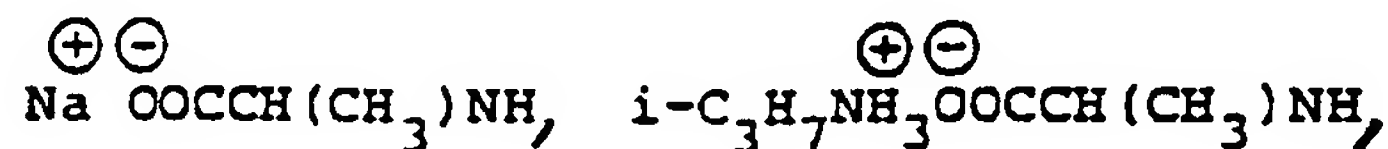
$\text{CH}_3\text{SO}_2\text{NH}$, $\text{CH}_3\text{NHSO}_2\text{NH}$, $\text{CH}_2=\text{CHCH}_2\text{NH}$, $\text{CH}\equiv\text{CCH}_2\text{NH}$, CH_3OOCNH , $\text{Cl}^-\text{H}_3\text{N}^+$,

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$\text{CH}_3\text{OOCCH}_2\text{NH}$, $\text{HOOCCH}(\text{CH}_3)\text{NH}$,

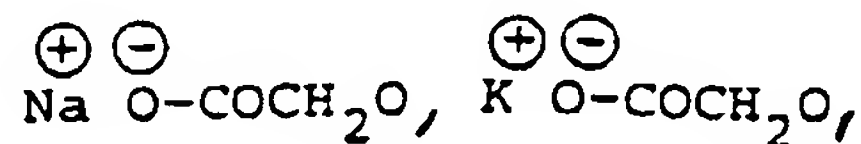
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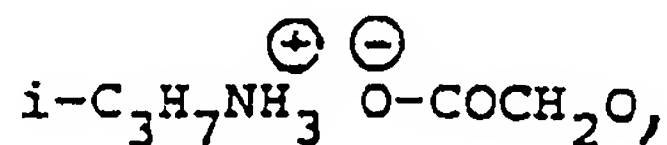
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$\text{C}_2\text{H}_5\text{OOCCH}(\text{CH}_3)\text{NH}$, $\text{CH}_3\text{NHOCCH}(\text{CH}_3)\text{NH}$, $(\text{CH}_3)_2\text{NOCCH}(\text{CH}_3)\text{NH}$, CH_3NHSO_2 , $(\text{CH}_3)_2\text{NSO}_2$, $(\text{C}_2\text{H}_5)_2\text{NCOS}$, $(\text{CH}_3)_3\text{SiCH}_2\text{S}$, $(\text{CH}_3)_3\text{SiCH}_2\text{O}$, $\text{HO-COCH}_2\text{O}$,

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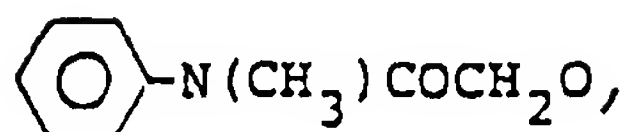
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$\text{CH}_2=\text{CHC}(\text{CH}_3)_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{C}\equiv\text{CCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{NHCOCH}_2\text{O}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{O}$, $\text{n-C}_4\text{H}_9\text{NHCOCH}_2\text{O}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{O}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{O}$,



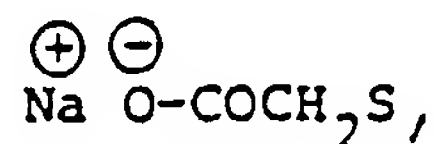
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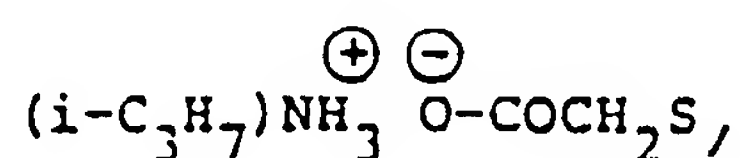
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$\text{CH}_3\text{SO}_2\text{NHCOCH}_2\text{O}$, $\text{CH}_3\text{SO}_2\text{NHCOCH}(\text{CH}_3)\text{O}$, $(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}(\text{CH}_3)\text{O}$,
 $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{S}$, $\text{CH}_2=\text{C}(\text{Cl})\text{CH}_2\text{S}$, $\text{ClCH}=\text{CHCH}_2\text{S}$, $\text{ClCH}=\text{C}(\text{Cl})\text{CH}_2\text{S}$, NCCH_2S ,

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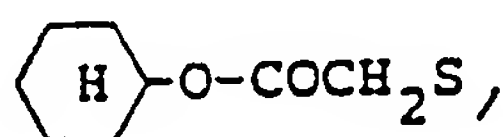
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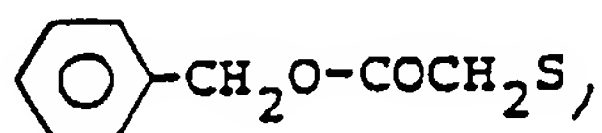
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$\text{HO-COCH}_2\text{S}$, $\text{i-C}_3\text{H}_7\text{O-COCH}_2\text{S}$, $\text{n-C}_4\text{H}_9\text{O-COCH}_2\text{S}$, $\text{C}_{12}\text{H}_{25}\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{S-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{S-COCH}_2\text{S}$, $\text{i-C}_3\text{H}_7\text{S-COCH}_2\text{S}$,
 $\text{CH}\equiv\text{CCH}_2\text{O-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}_2\text{S}$,

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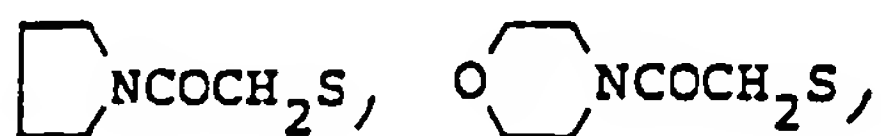


$(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{NHCOCH}_2\text{S}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{S}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{S}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{S}$,
 $\text{HC}\equiv\text{CCH}_2\text{NHCOCH}_2\text{S}$,

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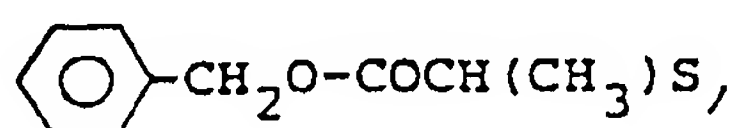


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$\text{i-C}_4\text{H}_9\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{t-C}_4\text{H}_9\text{CH}_2\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{n-C}_6\text{H}_{13}\text{O-COCH}(\text{CH}_3)\text{S}$,

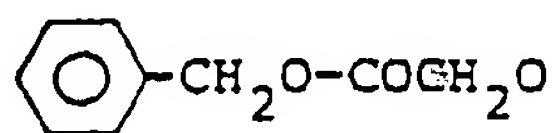
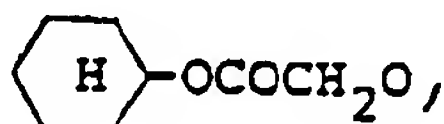


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$\text{C}_2\text{H}_5\text{S-COCH}(\text{CH}_3)\text{S}$, $\text{i-C}_3\text{H}_7\text{S-COCH}(\text{CH}_3)\text{S}$, $\text{NCCH}(\text{CH}_3)\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{SC}_2\text{H}_4\text{O-COCH}(\text{CH}_3)\text{S}$,

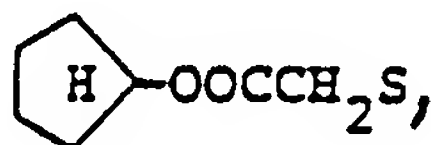


(CH₃)₃SiCH₂O-COCH(CH₃)S, CH₂=CHCH₂NHCOCH(CH₃)S, (H₂C=CHCH₂)₂NCOCH(CH₃)S, HC≡CCH₂NHCOCH(CH₃)S, CH₃CO-O-(CH₂)₂S, C₂H₅O-COC(CH₃)₂S, CH₃O-COCH(i-C₃H₇)S, C₂H₅O-COCH(CH₃)N(CH₃), CH₂=CHCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, CH≡CCH₂O-CO,

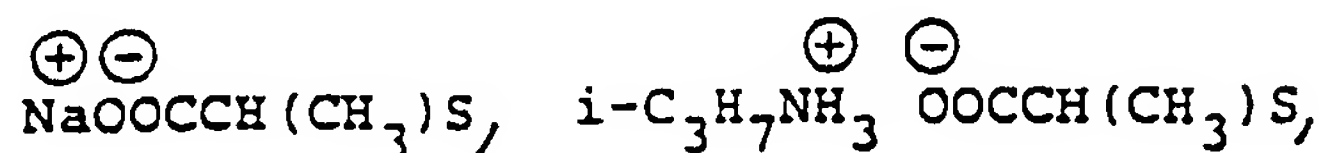


and CH₃O-CO-OCH₂S-CO;

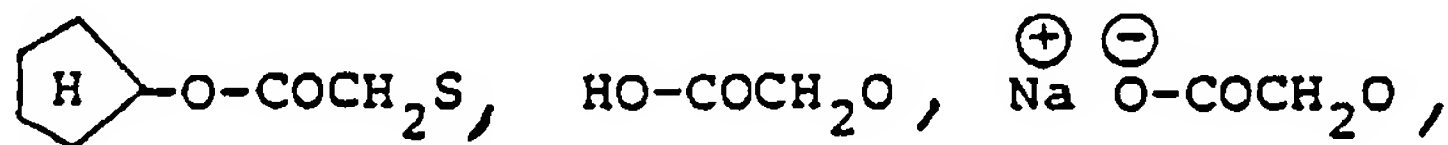
f) X is Cl, R¹ is CH₃, R² is CHF₂S, R³ is Cl and R⁴ is selected from CH₂=CHCH₂S, CH≡CCH₂S, C₂H₅OCH₂CH₂S, CH₃SCH₂S, NCCH₂S, C₂H₅OOCS, CH₃OOCH₂S, C₂H₅OOCH₂S,



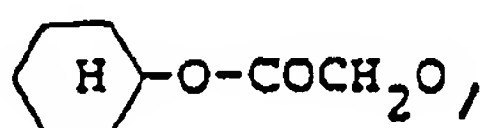
CH₃NHOCCH₂S, (CH₃)₂NOCCH₂S, HOOCCH(CH₃)S,



C₂H₅OOCH(CH₃)S, n-C₃H₇OOCH(CH₃)S, i-C₃H₇OOCH₂S, (CH₃)₂CHCH₂OOCH₂S, CH₂=CHCH₂OOCH(CH₃)S, C₂H₅OOCH(CH₂)₃S, CH₃NHOCCH(CH₃)S, (CH₃)₂NOCCH(CH₃)S, CH≡CCH₂NH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃OOCHNH, (C₂H₅)₂NCOS, NCCH₂O, ClCH=CHCH₂O, CH₃OCH₂O, C₂H₅OC₂H₄O, CH₃OCH₂S, ClCH=CHCH₂S, CH₂=C(Cl)CH₂S,

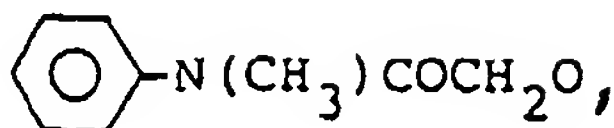


CF₃CH₂O-COCH₂O, CH₂=CHCH₂O-COCH₂O,



CH₃NHCOCH₂O, CH₂=CHCH₂O-COCH(CH₃)O, CH₃NHCOCH(CH₃)O, NCCH₂O-COCH₂O, CH₃O-COCH(CH₃)

S, NCCH₂O-COCH(CH₃)S, C₂H₅O-CO(CH₂)₃S,



HO-COCH₂NH, C₂H₅O-COCH₂NH and C₂H₅O-COCH(CH₃)NH;

10 g) X is F, R¹ is CH₃, R² is CH₃SO, R³ is Cl and R⁴ is 5-NO₂;

h) X is F, R¹ is CH₃, R² is CH₃S, R³ is Cl and R⁴ is selected from 5-NH₂, 5-SH, 5-CH≡CCH₂O, 5-CH≡CCH₂S and 5-C₂H₅OOCCH(CH₃)S;

15 i) X is F, R¹ is CH₃, R² is CHF₂O, R³ is Cl and R⁴ is selected from HO-COCH₂O, CH₃O-COCH₂O, CH₂=CHCH₂O-COCH₂O, CH≡CCH₂O-COCH₂O, HO-COCH(CH₃)O, CH₂=CHCH₂O-COCH(CH₃)O, CH≡CCH₂O-COCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(CH₃)CH₂S, CH≡CCH₂S, C₂H₅OC₂H₄S, HO-COCH₂S, CH₃O-COCH₂S, C₂H₅O-COCH₂S, C₂H₅O-COCH₂S(O), n-C₄H₉O-COCH₂S, CH₂=CHCH₂O-COCH₂S, CH₃SCH₂O-COCH₂S,



25 CH₃NHCOCH₂S, (CH₃)₂NCOCH₂S, HO-COCH(CH₃)S, CH₃O-COCH(CH₃)S, C₂H₅O-COCH(CH₃)S, n-C₄H₉O-COCH(CH₃)S, CH₂=CHCH₂O-COCH(CH₃)S, CH≡CCH₂O-COCH(CH₃)S, CH₃OC₂H₄OC₂H₄O-COCH(CH₃)S, CH₃SO₂NH, CH₃NHSO₂NH, C₂H₅O-COCH₂NH, C₂H₅O-COCH(CH₃)NH, CH₂=CHCH₂O-CO, CH≡CCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, (CH₃)₂NCO and

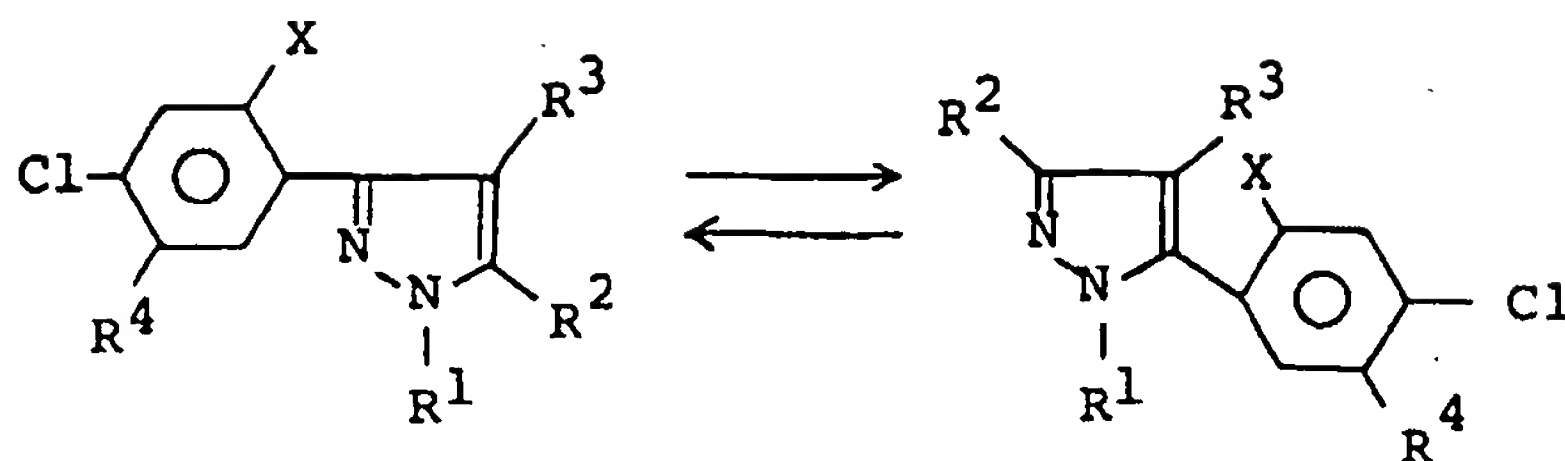


35 j) X is Cl, R¹ is CH₃, R² is CHF₂O, R³ is Br and R⁴ is BrCH₂CH(Br)CH₂O or BrC≡CCH₂O.

The present inventors made intensive studies in order to develop a novel herbicide and as a result have found that 3-(substituted phenyl)pyrazole derivatives represented by the above general formula (I) (hereinafter, simply referred to as formula (I)) and salts thereof are novel compounds, not yet written in literature, and they show excellent herbicidal effects on weeds even at lower dosages. Based on this finding, the present invention has been accomplished.

Prior to the present invention, compounds considered as analogous to the present inventive compounds were disclosed as herbicides in Japanese Patent Application Kokai Nos. Sho.50-117936 (corresponding to GB-A-1488285) Sho.52-91861, Sho.54-70270, and Sho.55-9062 and in other literature. However, the present inventive 3-(substituted phenyl)pyrazole derivative represented by formula (I) or salts thereof have never been disclosed and show superior herbicidal effects at lower dosages than those where the compounds disclosed in the above patent applications do.

The present inventive 3-(substituted phenyl)-pyrazole derivatives represented by formula (I) and salts thereof include structural isomers as shown below. These structural isomers form simultaneously in the production of 3-(substituted phenyl)pyrazole derivatives and each isomer can be isolated by a suitable separating method, e.g. recrystallization or column chromatography. The scope of the present invention also includes these structural isomers.



In this formula, R¹, R², R³, R⁴, and X are as defined above.

Certain compounds of the 3-(substituted phenyl)pyrazole derivatives represented by formula (I) have optical isomers, which are also included in the scope of the present invention.

The substituent R¹ in 3-(substituted phenyl)-pyrazole derivatives represented by formula (I) and salts thereof is methyl.

The substituent R² can be exemplified by hydroxy, mercapto alkylthio, haloalkoxy, and haloalkylthio groups, of which preferred are haloalkoxy groups and particularly preferred is a difluoromethoxy group.

The substituent R³ can be exemplified by hydrogen and halogen atoms, of which preferred are halogen atoms, particularly a chlorine atom.

The substituent R⁴ can be exemplified by formyl, nitro, -CO-B-R⁶ (wherein, B and R⁶ are as defined above) and -D-R⁸ (wherein, D and R⁸ are as defined above). Preferred examples of -CO-B-R⁶ are methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, methoxycarbonylmethoxycarbonyl, ethoxycarbonylmethoxycarbonyl, propoxycarbonylmethoxycarbonyl, methoxycarbonylethoxycarbonyl, ethoxycarbonylethoxycarbonyl, or propoxycarbonylethoxycarbonyl.

Preferred examples of -D-R⁸ are alkoxy and alkylthio groups containing each methyl, ethyl, propyl, or butyl; alkenyloxy groups containing each propenyl, butenyl, or pentenyl; alkynyloxy groups containing each propynyl, butynyl, or pentynyl; alkenylamino groups containing each propenyl, butenyl, or pentenyl; and alkoxy-carbonylalkoxy and alkylthiocarbonylalkoxy groups containing each methyl, ethyl, propyl, or butyl.

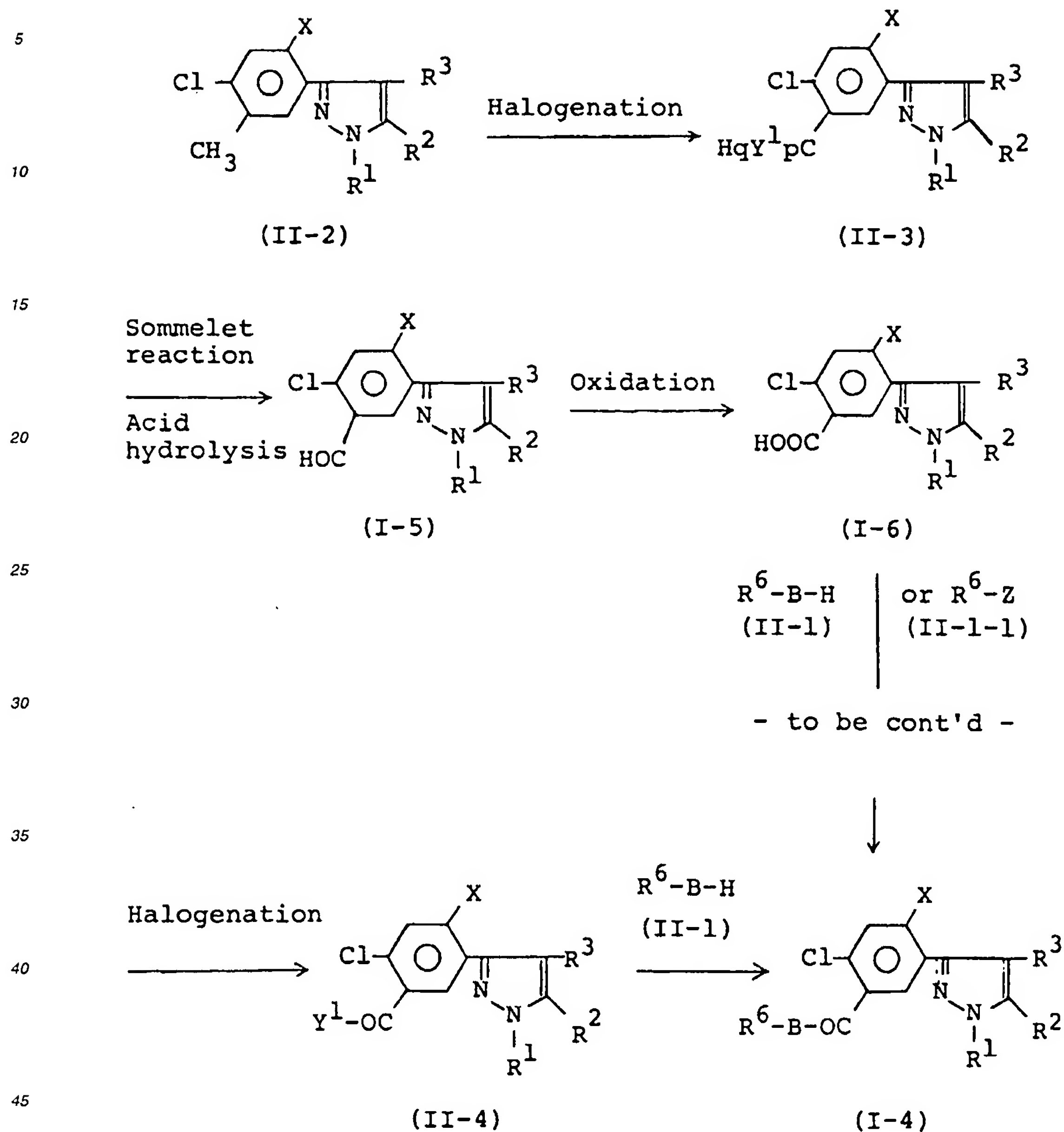
The substituent X is a halogen atom which can be exemplified by chlorine, bromine, fluorine, and iodine, of which preferred are chlorine and fluorine.

Salts of 3-(substituted phenyl)pyrazole derivatives represented by formula (I) are of mineral acids including, e.g. sulfuric acid and hydrochloric acid and of organic acids including, e.g. p-toluenesulfonic acid.

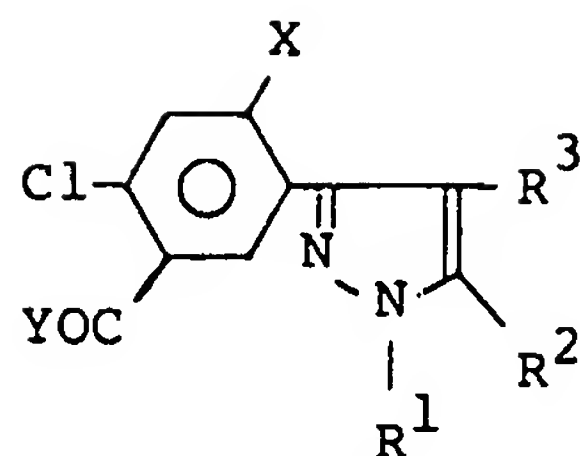
Salts of 3-(substituted phenyl)pyrazole derivatives represented by formula (I) can be produced by treating these derivatives with a suitable mineral acid or organic acid.

Typical processes for producing 3-(substituted phenyl)pyrazole derivatives of formula (I) or salts thereof can be illustrated by the following reaction schemes ①, ② and ③

Reaction Scheme (1)



In the above equation, X, R¹, R², R³, and -CO-B-R⁶ are as defined above, Y¹ and Z denote halogen, and p and q denote each an integer of 1 or 2 with the proviso that the sum of p and q is 3. That is, the 3-(substituted phenyl)pyrazole derivative represented by formula (I-4) can be produced as follows: A pyrazole derivative represented by formula (II-2) is reacted with a halogenating agent in an inert solvent to give a pyrazole derivative represented by formula (II-3), which in turn is subjected to Sommelet reaction with hexamethylenetetramine in an inert solvent and further hydrolyzed with acid to give a 3-(substituted phenyl)-pyrazole derivative represented by formula (I-5), which in turn is oxidized in an inert solvent to give (I-6), which in turn is halogenated in an inert solvent to give an acid halide represented by formula (II-4). Pyrazole derivatives represented by the following formula (I-3);



(I-3)

in which R^1 , R^2 , R^3 , R^6 , X and B are defined above, Y denotes halogen or OH, which consist of a derivative of formula (I-6) (corresponding to the case when Y is OH and a derivative of formula (II-4) (corresponding to the case when Y is halogen) are reacted with a compound represented by formula (II-1) or with a halide represented by formula (II-1-1) in an inert solvent in the presence or absence of a base to produce the 3-(substituted phenyl)pyrazole derivative represented by formula (I-4).

(1) Preparation of compound (II-3) from compound (II-2)

Any inert solvent may be used in this reaction that does not markedly hinder the reaction from proceeding. Such solvents include, though not limited to; halogenated hydrocarbons, e.g. methylene chloride, chloroform, and carbon tetrachloride; aromatic hydrocarbons, e.g. benzene, toluene, and xylene; nitriles, e.g. acetonitrile and benzonitrile; linear ethers, e.g. methyl Cellosolve and diethyl ether; cyclic ethers, e.g. dioxane and tetrahydrofuran; and sulforane, dimethyl sulfone and dimethylsulfoxide. These solvents may be used alone or in combination.

Halogenating agents suitable for use in the above reaction include, for example, N-bromosuccinimide (NBS), N-chlorosuccinimide (NCS), bromine, chlorine, t-butyl hypohalogenite, and trichloromethanesulfonyl halide. When NBS or NCS is used as a halogenating agent, it is preferable to use an organic peroxide such as benzoyl peroxide or light as a catalyst or exciter.

The amount of halogenating agent used may be chosen from the range of 1 mole to an excess of 1 mole per mole of the pyrazole derivative of formula (II-2).

The reaction temperature ranges from 0°C to the boiling point of the inert solvent used.

The reaction period is chosen from the range of several minutes to 48 hours so as to complete the reaction.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (II-3) can be obtained from the reaction product solution by an ordinary method, e.g. solvent extraction, and if necessary, by purification procedure such as recrystallization or column chromatography.

(2) Preparation of compound (I-5) from compound (II-3)

This reaction process consists of the first step of carrying out Sommelet reaction and the second step of carrying out acid hydrolysis. Inert solvents suitable for use in the first step include organic acids, e.g. glacial acetic acid, mineral acids, e.g. hydrochloric acid and sulfuric acid, water, and mixtures of these solvents.

Hexamethylenetetramine is used in an amount equimolar to the pyrazole derivative of formula (II-3). An excess of hexamethylenetetramine can also be used.

The reaction temperature ranges from 10°C to the boiling point of inert solvent used, preferably from 30 to 180°C.

After completion of the reaction, the resultant solution containing the objective compound is subjected to the acid hydrolysis in the second step.

To the acid hydrolysis, the reaction product solution from the first step is subjected as such or with a large excess of mineral acid such as hydrochloric acid or sulfuric acid added as occasion demands, the latter procedure being preferable.

The reaction temperature is in the range of 80 to 180°C and the reaction period is chosen from the range of several minutes to 48 hours so as to complete the reaction.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (I-5) can be obtained from the reaction product solution by an ordinary method, e.g. solvent extraction, and if necessary, by purification procedure such as recrystallization or column chromatography.

(3) Preparation of compound (I-6) from compound (I-5)

Inert solvents suitable for use in this reaction include aromatic hydrocarbons such as benzene, toluene, and xylene, and further pyridine and its derivatives and water. These solvents may be used alone or in combination.

Oxidants, e.g. potassium permanganate and potassium dichromate, can be used as oxidizing agents in amounts of 1 to 5 moles, preferably 1 to 2 moles, per mole of the 3-(substituted phenyl)pyrazole derivative of formula (I-5).

The reaction temperature is from 0°C to the boiling point of inert solvent used, preferably in the range of 30 to 180°C. The reaction period is chosen from the range of several minutes to 48 hours so as to complete the reaction.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (I-6) can be obtained from the reaction product solution by an ordinary method e.g. solvent extraction, and if necessary, by purification procedure such as recrystallization or column chromatography.

(4) Preparation of compound (II-4) from compound (I-6)

Inert solvents suitable for use in this acid halide-forming reaction include; halogenated hydrocarbons, e.g. methylene chloride, chloroform, and carbon tetrachloride; aromatic hydrocarbons, e.g. benzene, toluene, and xylene; linear ethers, e.g. methyl Cellosolve, diethyl ether, and diisopropyl ether; and cyclic ethers, e.g. dioxane and tetrahydrofuran.

Halogenating agents suitable for this reaction include, for example, thionyl chloride, phosphorus pentachloride, and phosphorus trichloride.

The halogenating agent may be used in amounts of 1 mole to an excess of 1 mole, preferably in such an excess, per mole of the 3-(substituted phenyl)pyrazole derivative of formula (I-6).

A catalytic amount of triethylamine, pyridine, dimethylformamide, or the like may be added for the purpose of promoting this reaction.

The reaction temperature is in the range of room temperature to the boiling point of solvent used. The reaction period ranges from several minutes to 48 hours depending upon the amounts of reactants used and the reaction temperature.

After completion of the reaction, the excess halogenating agent and the solvent are removed from the reaction product solution by evaporation to isolate the objective compound, which may be subjected to the next reaction after purification, if necessary, by recrystallization, distillation, or some other method.

(5) Preparation of compound (I-4) from compound (I-6) or from compound (II-4)

This reaction is esterification or amide-forming reaction. The esterification can be carried out by using an excess of the alcohol corresponding to the objective ester in the presence of a mineral acid, e.g. concentrated sulfuric acid, which acts as a reactant and a solvent. For the esterification, the 3-(substituted phenyl)pyrazole derivative of formula (I-6) can also be used in the form of alkali metal salt.

The esterification or the amide-forming reaction can also be carried out in the presence of an inert solvent and a dehalogenating agent. Inert solvents suitable for use in the acid halide-forming reaction can also be used in this case.

The alcohol or amine or the halide of formula (II-1-1) may be used in an equimolar amount or more.

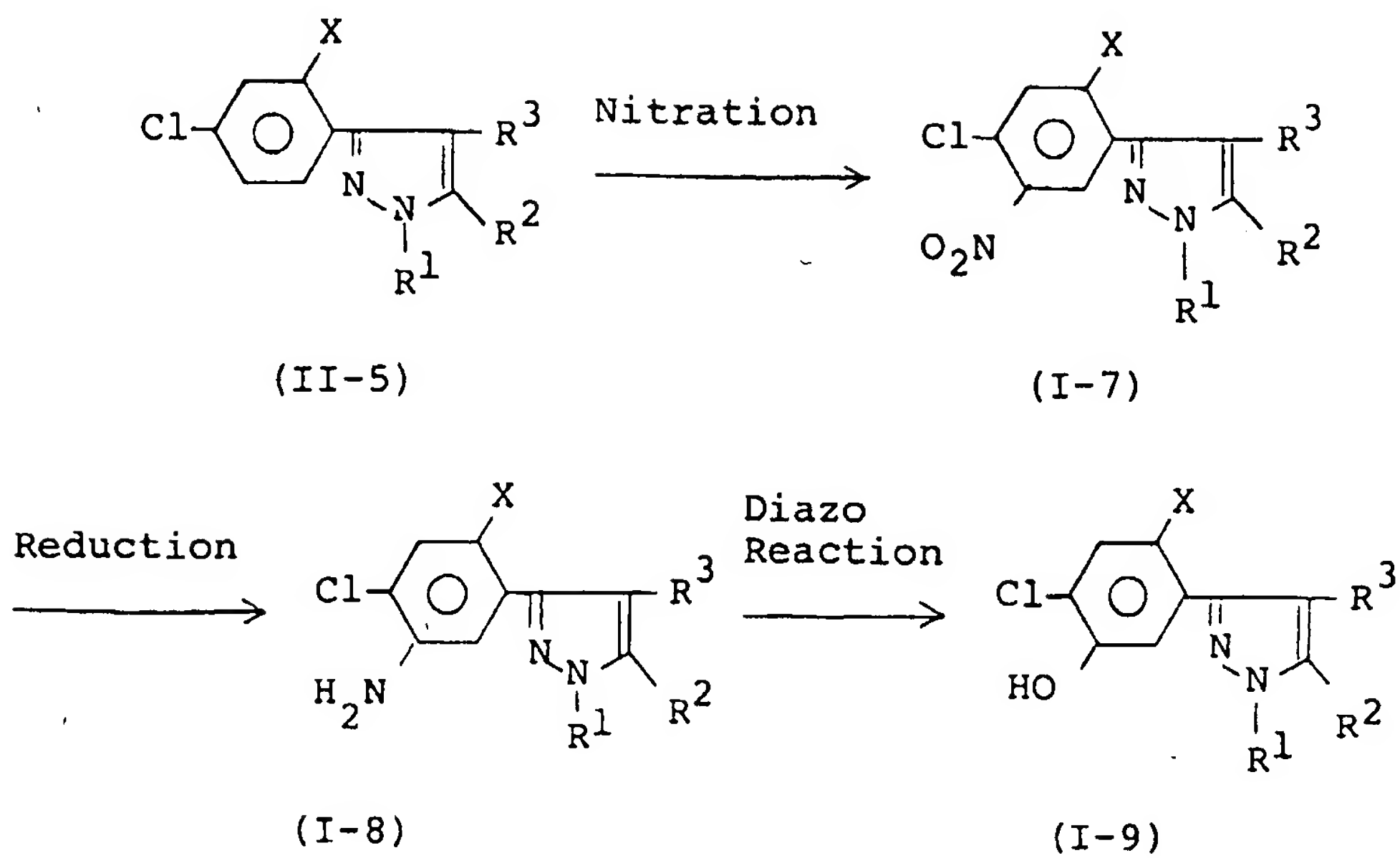
Dehalogenating agents suitable for use in this reaction are inorganic bases and organic bases. The inorganic bases include alkali metal salts, e.g. sodium hydroxide and potassium hydroxide and the organic bases include tertiary amines, e.g. triethylamine, and 4-dimethylaminopyridine, and 1,8-diazabicyclo-[5,4,0]-7-undecene (DBU).

The reaction temperature is in the range of 0°C to the boiling point of the solvent used, preferably in the range of 0 to 150°C. The reaction period ranges from several minutes to 48 hours depending upon the amounts of reactants used, the reaction temperature, etc.

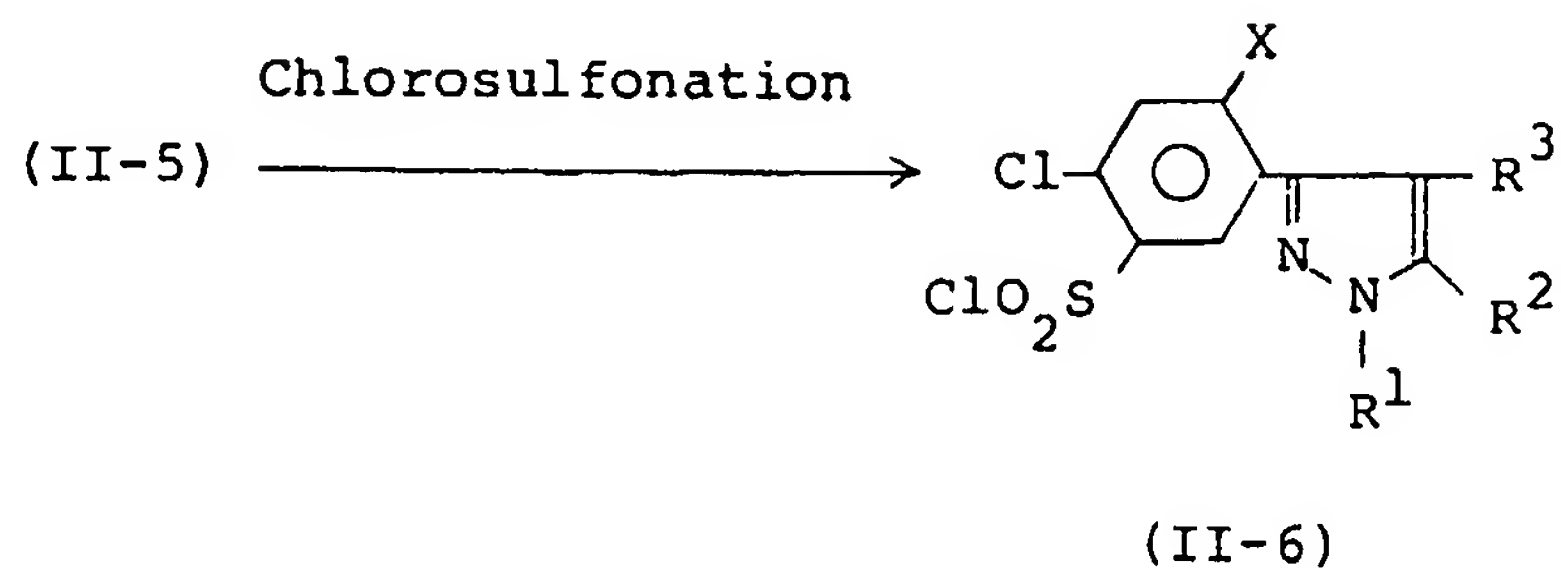
After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (I-4) can be obtained from the reaction product solution by an ordinary method, e.g. solvent extraction, and if necessary, by purification procedure such as recrystallization or column chromatography.

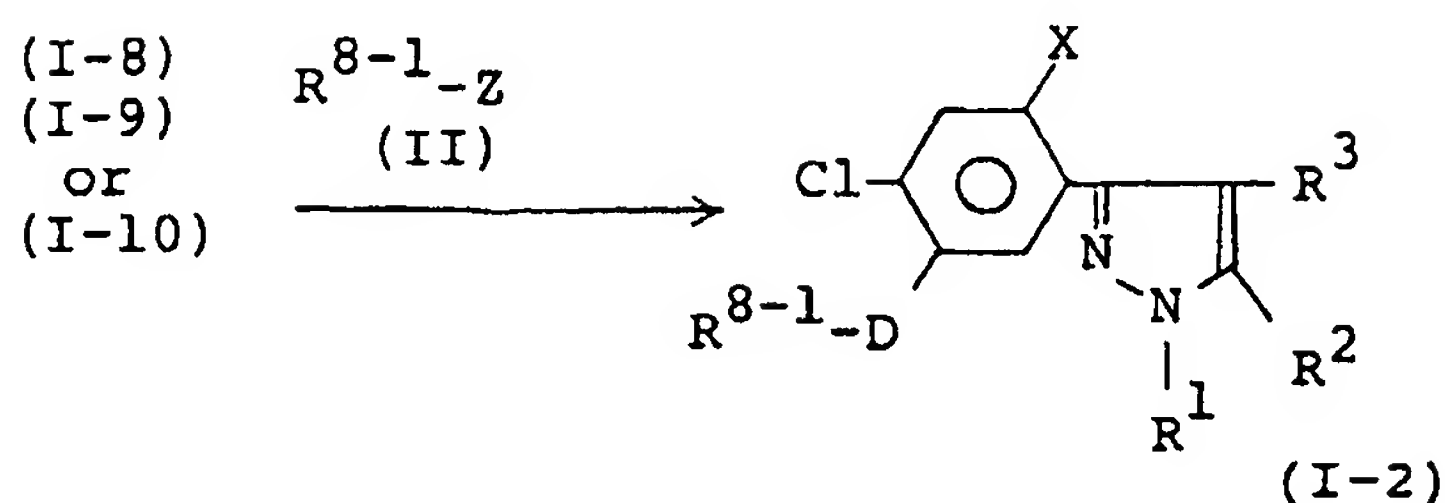
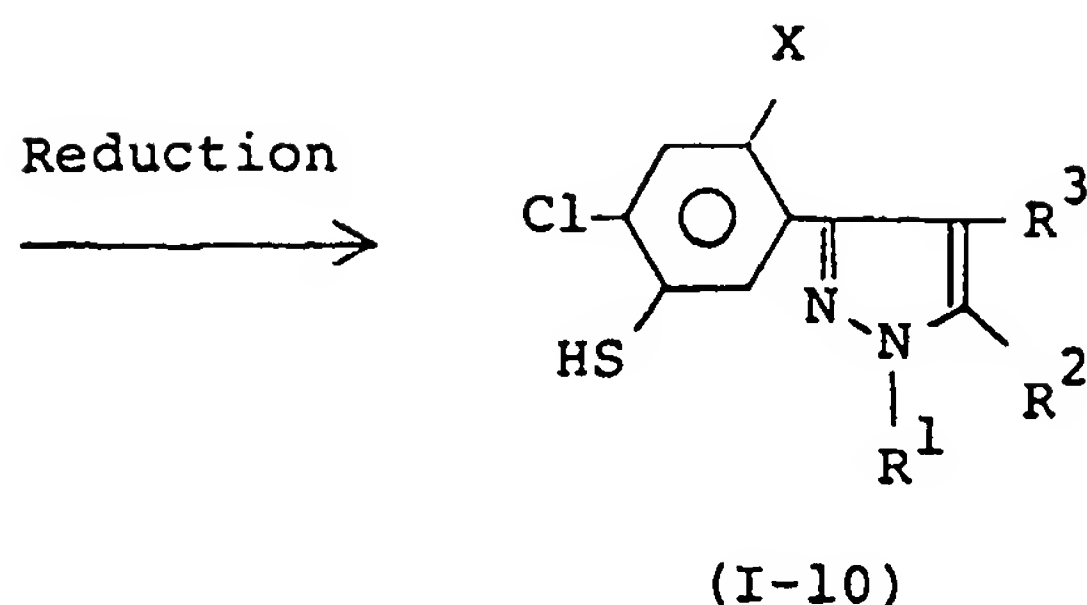
Reaction Scheme (2)

Process A



Process B

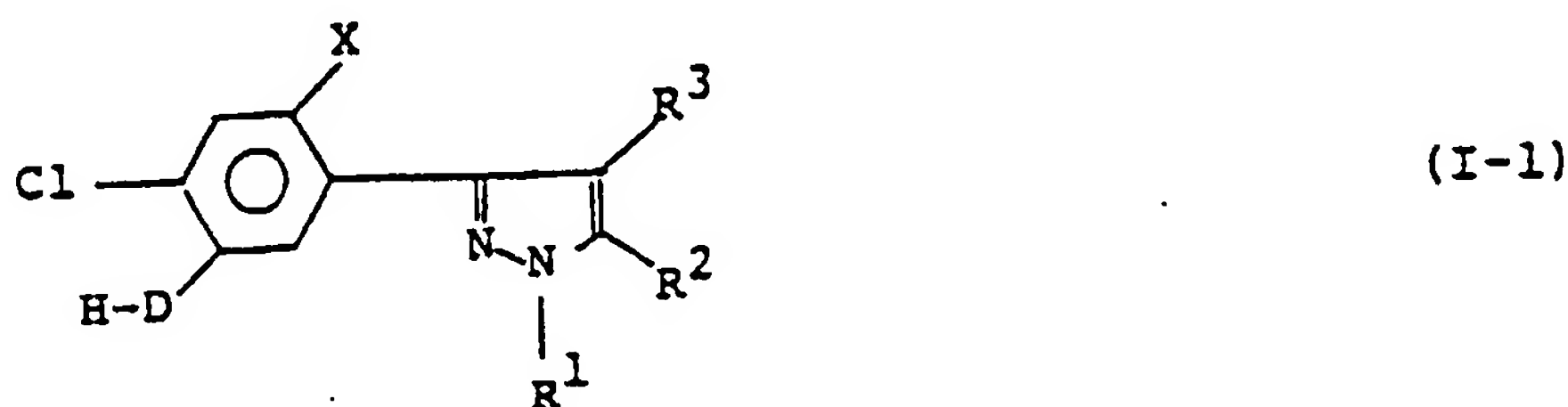




In the above equations, R^1 , R^2 , R^3 , R^{8-1} , D, X, and Z are as defined above.

That is, a pyrazole derivative of formula (II-5) is subjected to nitration with a nitrating agent in an inert solvent to give a 3-(substituted phenyl)-pyrazole derivative of formula (I-7). This derivative is then reduced in an inert solvent to give a 3-(substituted phenyl)pyrazole derivative of formula (I-8), which is further diazotized in an inert solvent and then decomposed, whereby a 3-(substituted phenyl)-pyrazole derivative of formula (I-9) can be obtained. On the other hand, a pyrazole derivative of formula (II-5) is subjected to chlorosulfonyl substitution in an inert solvent to give a pyrazole derivative of formula (II-6), which in turn is reduced, whereby a 3-(substituted phenyl)pyrazole derivative of formula (I-10) can be obtained.

Further, a 3-(substituted phenyl)pyrazole derivative of formula (I-2) can be produced from a 3-(substituted phenyl)pyrazole derivative represented by the general formula



wherein X, R^1 , R^2 , R^3 , and D are as defined above, especially of formula (I-8), (I-9), or (I-10) by reacting with a halide of formula (II) in an inert solvent in the presence of a base.

When R^{8-1} in the 3-(substituted phenyl)-pyrazole derivative of formula (I-2) is $-CH \cdot R^{10} \cdot CO \cdot E \cdot R^{12}$ (wherein, R^{10} , R^{12} , and E are as defined above), this 3-(substituted phenyl)pyrazole derivative can be produced by converting an acid represented by $-CHR^{10} \cdot CO \cdot E \cdot H$ to an acid halide in the same manner as in the above reaction process (1) and subjecting the acid halide to esterification, thiol esterification, or amide-forming reaction.

(1) Preparation of compound (I-7) from compound (II-5)

Nitrating agents suitable for use in this reaction include concentrated nitric acid, a mixture of fuming nitric acid with concentrated sulfuric acid, and acetyl nitrate that is formed by mixing concentrated nitric acid with acetic anhydride.

Mineral acids such as sulfuric acid and hydrochloric acid can be used as solvents in this process.

Since this reaction is an equimolar reaction, the use of an equimolar amount of nitrating agent is sufficient though

it may be used in excess.

The reaction temperature is in the range of -10 to 140°C, preferably -10 to 20°C. The reaction period is chosen from the range of several minutes to 48 hours depending upon the amounts of reactants charged, the reaction temperature, etc.

5 After completion of the reaction, the intended 3-(substituted-phenyl)pyrazole derivative of formula (I-7) can be obtained from the reaction product solution by either pouring it into ice-cold water and collecting the precipitated crystals or subjecting the solution to solvent extraction or other separating procedure to isolate the intended product, and if necessary, by further purification procedure such as recrystallization.

10 (2) Preparation compound (I-8) from compound (I-7)

Any inert solvent that does not markedly hinder the reaction from proceeding may be used in this reaction. Such inert solvents include; alcohols, e.g. methanol, ethanol, and propanol; linear ethers, e.g. diethyl ether and methyl Cellosolve; cyclic ethers, e.g. dioxane and tetrahydrofuran; organic acids, e.g. acetic acid; mineral acids, e.g. hydrochloric acid; and water.

Reducing agents suitable for this reaction in an acidic medium include, for example, zinc, iron, tin, and tin chloride. Zinc dust can be used in a neutral or basic medium. For the catalytic hydrogenation, which can be carried out under normal or elevated pressure, suitable catalysts include, for example, Raney nickel, palladium-carbon, palladium oxide, platinum, platinum black, platinum sulfide-carbon, and rhodium-alumina.

20 In an acidic medium, the reducing agent may be used in an equimolar amount though generally in excess. For the catalytic hydrogenation, the amount of Raney nickel or the like when it is used is from 5 to 20% by weight based on the weight of the 3-(substituted)pyrazole derivative of formula (I-7) and the amount of noble metal such as platinum or palladium when it is used is from 0.02 to 5% by weight on the the same basis.

25 The reaction temperature is in the range of 0 to 150°C, preferably 10 to 100°C. The reaction period may be chosen from the range of several minutes to 48 hours depending upon the amounts of reactants charged, the reaction temperature, etc.

30 After completion of the reaction, the intended 3-(substituted phenyl)pyrazole derivative of formula (I-8) can be obtained in the following manner: When the reaction is carried out in an acidic medium, the reaction product solution is poured into ice-cold water, this mixture is turned basic, and the objective product is isolated by a method such as solvent extraction. In case of catalytic hydrogenation, the catalyst is removed from the reaction product fluid by filtration and the objective product is recovered by concentrating the filtrate. If necessary, the objective product is further purified by recrystallization, column chromatography, or some other method.

35 (3) Preparation of compound (I-9) from compound (I-8)

This process consists of the step of diazotization and the step of decomposition. In the diazotization step, a mineral acid such as sulfuric acid or hydrochloric acid can be used as an inert solvent. For the diazotization, sodium nitrite in the form of powder or aqueous solution of suitable concentration is used in an equimolar amount or in excess.

The reaction temperature is in the range of -5 to 5°C.

40 The resulting diazonium salt is subjected to decomposition without isolation.

In the decomposition step, 30 to 70% sulfuric acid may be used as an inert solvent.

The decomposition is carried out in the presence of a catalyst at a temperature of 10 to 180°C. When the diazonium salt in solution form is added dropwise to aqueous sulfuric acid, the reaction is conducted at a temperature of 50 to 180°C. When cuprous oxide is used, as a catalyst, the reaction temperature is in the range of 10 to 50°C.

45 Copper nitrate and cuprous oxide can be used as catalysts. To the reaction system is added copper nitrate and then cuprous oxide is added in limited amounts.

Copper nitrate is used in an amount of 1 to 60 moles, preferably 30 to 60 moles, and cuprous oxide is used in an amount of catalytic to 2 moles, preferably about 1 mole, per mole of the diazonium salt.

50 The total reaction period of diazotization and decomposition may be chosen from the range of several minutes to 48 hours depending upon the amounts of reactants charged and the reaction temperature.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (I-9) can be obtained by pouring the reaction product solution into ice-cold water, isolating the objective product from the mixture by a method such as solvent extraction, and if necessary, purifying the isolated product by recrystallization, column chromatography, or some other method.

55

(4) Preparation of compound (II-6) from compound (II-5)

Any solvent may be used for this reaction that does not markedly hinder the reaction. Such solvents include;

halogenated hydrocarbons, e.g. methylene chloride, chloroform, and carbon tetrachloride, aliphatic nitriles, e.g. acetonitrile; linear ethers, e.g. methyl Cellosolve; cyclic ethers, e.g. dioxane and tetrahydrofuran; and mineral acids, e.g. concentrated sulfuric acid.

For this chlorosulfonyl substitution, chlorosulfonic acid or the like may be used. Another available method comprises sulfonating the starting compound of formula (II-5) with fuming sulfuric acid, converting the resulting sulfonic acid to an alkali metal salt, and chlorinating it with phosphorus pentachloride to give the objective product. A further method comprises reacting the starting compound with fuming sulfuric acid, followed by the reaction with carbon tetrachloride to give the objective product.

Chlorosulfonic acid is used in an equimolar amount or in excess, preferably in excess.

Fuming sulfuric acid is used also in an equimolar amount or in excess, preferably in excess. Carbon tetrachloride is used in an amount chosen properly from the range of an equimolar amount to an excess.

The reaction temperature is in the range of 0 to 180°C, preferably 15 to 100°C. The reaction period may be chosen from the range of several minutes to 48 hours depending upon the amounts of reactants charged and the reaction temperature.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (II-6) can be obtained by pouring the reaction product solution into water, isolating the objective product from the resulting mixture by solvent extraction, and if necessary, purifying the isolated product by recrystallization, column chromatography, or some other procedure.

(5) Preparation of compound (I-10) from compound (II-6)

This reduction can be carried out in an inert solvent such as glacial acetic acid. Reducing agents suitable for this reaction include, for example, zinc, tin, and tin chloride, which may be used in an amount of 1 mole or more, preferably 5 moles or more, per mole of the compound of formula (II-6).

The reaction temperature is in the range of 0 to 180°C, preferably 15 to 120°C. The reaction period is chosen from the range of several minutes to 48 hours depending upon the amounts of reactants charged and the reaction temperature.

After completion of the reaction, the 3-(substituted phenyl)pyrazole derivative of formula (I-10) can be obtained by pouring the reaction product solution into water, isolating the objective product from the mixture by solvent extraction, and if necessary, purifying the isolated product by recrystallization, column chromatography, or some other procedure.

(6) Preparation of compound (I-2) from compound (I-9) or compound (I-10)

This reaction can be carried out according to the procedure of the above reaction process ①-5, whereby the 3-(substituted phenyl)pyrazole derivative of formula (I-2) can be produced.

③ Salts of 3-(substituted phenyl)pyrazole derivatives of formula (I)

These salts are formed from acids including mineral acids, e.g. hydrochloric acid and sulfuric acid, and organic acids, e.g. p-toluenesulfonic acid. The production of these salts can be carried out by treating 3-(substituted phenyl)pyrazole derivatives of formula (I), obtained according to the above stated process, with the above-cited mineral acid or organic acid.

Typical examples of the 3-(substituted phenyl)pyrazole derivatives represented by formula (I) and the salts thereof are shown in Table 1.

Formula (I):

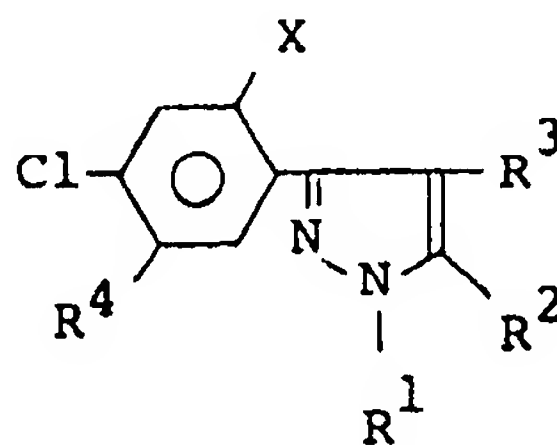

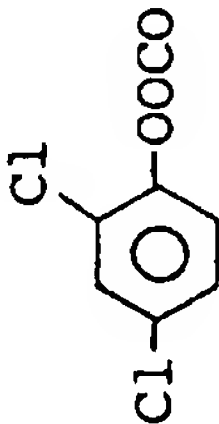
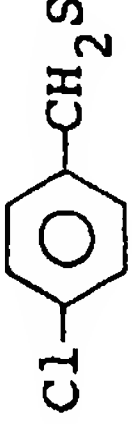


Table 1

Comp'd No.	R ¹	R ²	R ³	R ⁴	X	Property
1	CH ₃	CH ₃ S	H	HO	Cl	m.p. 157.1°C
2	CH ₃	CH ₃ S	Cl	HO	Cl	m.p. 163.2°C
3	CH ₃	CH ₃ S	Cl	i-C ₃ H ₇ O	Cl	nD 1.5882 (26.5°C)
4	CH ₃	CH ₃ S	Cl	CH ₂ =CHCH ₂ O	Cl	nD 1.6131 (25.3°C)
5	CH ₃	CH ₃ S	Cl	CH≡CCH ₂ O	Cl	m.p. 71.5 - 72.5°C
6	CH ₃	CH ₃ S	Cl	C ₂ H ₅ OCH ₂ CH ₂ O	Cl	nD 1.5944 (25,8°C)
7	CH ₃	CH ₃ S	Cl		Cl	nD 1.6075 (25.6°C)
8	CH ₃	CH ₃ S	Cl		Cl	nD 1.5988 (25.9°C)
9	CH ₃	CH ₃ S	Cl	HOOCCH(CH ₃)O	Cl	m.p. 191 - 194°C
10	CH ₃	CH ₃ S	Cl	CH ₃ OOCCCH(CH ₃)O	Cl	m.p. 90 - 93°C

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11	CH ₃	CH ₃ S	Cl	C ₂ H ₅ OOCCH(CH ₃)O	Cl	nD	1.5763 (28.8°C)
12	CH ₃	CH ₃ S	Cl	i-C ₃ H ₇ OOCCH(CH ₃)O	Cl	m.p.	87 - 90°C
13	CH ₃	CH ₃ S	Cl	CH ₂ =CHCH ₂ OOCCH(CH ₃)O	Cl	nD	1.5802 (23.4°C)
14	CH ₃	CH ₃ S	Cl	CH≡CCH ₂ OOCCH(CH ₃)O	Cl	nD	1.5858 (23.7°C)
15	CH ₃	CH ₃ S	Cl	HS	Cl	nD	1.6303 (24.3°C)
16	CH ₃	CH ₃ S	Cl	i-C ₃ H ₇ S	Cl	m.p.	91.8°C
17	CH ₃	CH ₃ S	Cl	CHF ₂ S	Cl	m.p.	87.2°C
18	CH ₃	CH ₃ S	Cl	CHF ₂ CF ₂ S	Cl	nD	1.5820 (24.7°C)
19	CH ₃	CH ₃ S	Cl	CH ₂ =CHCH ₂ S	Cl	Viscous substance	
20	CH ₃	CH ₃ S	Cl	CH ₂ =C(Cl)CH ₂ S	Cl	nD	1.6370 (25.1°C)
21	CH ₃	CH ₃ S	Cl	CH≡CCH ₂ S	Cl	m.p.	94.5°C
22	CH ₃	CH ₃ S	Cl		Cl	nD	1.6528 (24.9°C)
23	CH ₃	CH ₃ S	Cl	C ₂ H ₅ OOCCH ₂ S	Cl	m.p.	112.1°C
24	CH ₃	CH ₃ S	Cl	C ₂ H ₅ OOCCH ₂ CH ₂ S	Cl	m.p.	74.8°C
25	CH ₃	CH ₃ S	Cl	HO ₂ CCH(CH ₃)S	Cl	m.p.	180.0°C

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26	CH ₃	CH ₃ S	Cl	i-C ₃ H ₇ NH ₃ [⊕] ⊖OOCCH(CH ₃)S	Cl	nD	1.5878 (22.1°C)
27	CH ₃	CH ₃ S	Cl	(CH ₃) ₂ NH ₂ [⊕] ⊖OOCCH(CH ₃)S	Cl	m.p.	176.1°C
28	CH ₃	CH ₃ S	Cl	CH ₃ OOCCH(CH ₃)S	Cl	nD	1.6131 (24.9°C)
29	CH ₃	CH ₃ S	Cl	C ₂ H ₅ OOCCH(CH ₃)S	Cl	nD	1.6010 (25.0°C)
30	CH ₃	CH ₃ S	Cl	i-C ₃ H ₇ OOCCH(CH ₃)S	Cl	nD	1.5936 (25.0°C)
31	CH ₃	CH ₃ S	Cl	Cl(CH ₂) ₃ OOCCH(CH ₃)S	Cl	nD	1.6025 (25.2°C)
32	CH ₃	CH ₃ S	Cl	CH ₂ =CHCH ₂ OOCCH(CH ₃)S	Cl	nD	1.6030 (25.0°C)
33	CH ₃	CH ₃ S	Cl	CH≡CCH ₂ OOCCH(CH ₃)S	Cl	nD	1.5954 (25.0°C)
34	CH ₃	CH ₃ S	Cl	C ₂ H ₅ CH(CH ₃)CH ₂ OOCCH(CH ₃)S	Cl	nD	1.5834 (25.0°C)
35	CH ₃	CH ₃ S	Cl	(C ₂ H ₅) ₂ NCOCH(CH ₃)S	Cl	nD	1.5983 (26.4°C)
36	CH ₃	CH ₃ S	Cl	H ₂ NSO ₂	Cl	m.p.	203.5°C
37	CH ₃	CH ₃ S	Cl	CH ₃ NHSO ₂	Cl	m.p.	79.1°C
38	CH ₃	CH ₃ S	Cl	(CH ₃) ₂ NSO ₂	Cl	m.p.	135.7°C
39	CH ₃	CH ₃ S	Cl	n-C ₃ H ₇ NHSO ₂	Cl	m.p.	102-105°C
40	CH ₃	CH ₃ S	Cl	CH ₃ SO ₂ NHSO ₂	Cl	m.p.	184 - 187°C

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41	CH ₃	CH ₃ S	Cl	H ₂ N	Cl	nD	1.6403 (25.0°C)
42	"	"	"	$\text{Cl}^{\ominus}\text{H}_3\text{N}^{\oplus}$	"	m.p.	174.5°C
43	"	"	"	CH ₃ NH	"	nD	1.6368 (21.7°C)
44	"	"	"	(CH ₃) ₂ N	"	nD	1.6241 (22.1°C)
45	"	"	"	CH ₂ =CHCH ₂ NH	"	nD	1.6238 (22.2°C)
46	"	"	"	(CH ₂ =CHCH ₂) ₂ N	"	nD	1.6088 (21.6°C)
47	"	"	"	CH≡CCH ₂ NH	"	m.p.	140.8°C
48	"	"	"	CH ₃ OOCNH	"	nD	1.6301 (22.2°C)
49	"	"	"	HOOC	"	m.p.	213 - 214°C
50	"	"	"	C ₂ H ₅ OOC	"	nD	1.6029 (20.1°C)
51	"	"	"	CH ₃ SO ₂ NH	"	nD	1.6002 (25.9°C)
52	"	"	"	CH ₃ NHSO ₂ NH	"	m.p.	134.5°C
53	"	"	"	CH ₃ NHSO ₂ N(CH ₃)	"	nD	1.5763 (26.4°C)
54	"	"	"	(CH ₃) ₂ NSO ₂ N(CH ₃)	"	nD	1.5839 (26.0°C)
55	"	"	"	i-C ₃ H ₇ NHSO ₂ NH	"	m.p.	148.6°C

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56	CH ₃	CH ₃ S	Cl		Cl	m.p.
57	"	"	"	CH≡CCH ₂ N(CH ₃)SO ₂ N(CH=CCCH ₂)	"	111.5°C
58	"	"	"	CH ₃ OCH ₂ CH ₂ OOCNH	"	112 - 115°C
59	"	"	"	BrCH ₂ CH ₂ OOCNH	"	119.5°C
60	"	"	"	NCCH ₂ CH ₂ OOCNH	"	1.5938 (24.7°C)
61	"	"	"	CH ₂ =C(CH ₃)CH ₂ OOCNH	"	125.8°C
62	"	"	"		"	103 - 105°C
63	"	"	"	CH ₂ =CHCH ₂ OOCCH ₂ NH	"	112.0°C
64	"	"	"	CH≡CCH ₂ OOCCH ₂ NH	"	127.0°C
66	"	"	"	C ₂ H ₅ OOCCH(CH ₃)NH	"	Viscous substance
67	"	"	"	HOOCCH ₂ NH	"	207.1°C
68	"	"	"	CH ₃ OOCCH ₂ NH	"	132.4°C
69	"	"	"	HOOCCH(CH ₃)NH	"	218.7°C
				CH ₃ OOCCH(CH ₃)NH	"	144.2°C

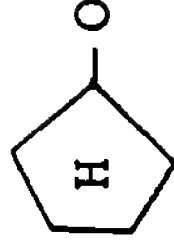
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70	CH ₃	CH ₃ SO	Cl	NO ₂	Cl	m.p.	168.1°C
71	"	"	"	CH ₃ SO ₂ NH	"	m.p.	168.0°C
72	"	"	"	CH ₃ NHSO ₂ NH	"	m.p.	146.3°C
73	"	"	"	i-C ₃ H ₇ NHSO ₂ NH	"	m.p.	184.8°C
74	"	"	"	CH ₃ NHSO ₂ N(CH≡CCH ₂)	"	m.p.	192.5°C
75	CHF ₂	CH ₃ S	"	NH ₂	"	m.p.	87.4°C
76	"	"	"	CH ₃ NHSO ₂ NH	"	m.p.	133.3°C
77	"	"	"	CH ₃ NHSO ₂ N(CH≡CCH ₂)	"	nD	1.5625 (25.6°C)
78	"	"	"	CH ₃ N(CH≡CCH ₂)SO ₂ N(CH≡CCH ₂)	"	nD	1.5675 (25.6°C)
79	"	CH ₃ SO	"	NH ₂	"	m.p.	187.1°C
80	"	"	"	NO ₂	"	m.p.	165.1°C
81	CH ₃	CHF ₂ O	"	CH ₃ O	"	nD	1.5567 (23.4°C)
82	"	"	"	C ₂ H ₅ O	"	nD	1.5434 (23.5°C)
83	"	"	"	n-C ₃ H ₇ O	"	nD	1.5410 (25.1°C)
84	"	"	"	i-C ₃ H ₇ O	"	nD	1.5422 (23.5°C)

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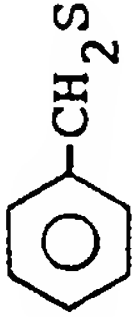
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85	CH ₃	CHF ₂ O	Cl		Cl	nD	1.5482 (25.4°C)
86	"	"	"	CH ₂ =CHCH ₂ O	"	nD	1.5536 (28.4°C)
87	"	"	"	CH≡CCH ₂ O	"	m.p.	84.0°C
88	"	"	"	C ₂ H ₅ OCH ₂ CH ₂ O	"	m.p.	79.5°C
89	"	"	"	NCCH ₂ O	"	m.p.	98.9°C
90	"	"	"	CH ₃ OOO	"	nD	1.5387 (23.2°C)
91	"	"	"	CH ₃ OOOCH ₂ O	"	m.p.	119.8°C
92	"	"	"	n-C ₃ H ₇ OOOCH ₂ O	"	m.p.	89.7°C
93	"	"	"	i-C ₃ H ₇ OOOCH ₂ O	"	m.p.	105.0°C
94	"	"	"	n-C ₄ H ₉ OOOCH ₂ O	"	m.p.	75.3°C
95	"	"	"	CH ₃ CH ₂ CH(CH ₃)OOOCH ₂ O	"	m.p.	107.8°C
96	"	"	"	n-C ₆ H ₁₃ OOOCH ₂ O	"	nD	1.5244 (23.0°C)
97	"	"	"	CH ₂ =CHCH ₂ OOOCH ₂ O	"	m.p.	84.7°C
98	"	"	"	CH≡CCH ₂ OOOCH ₂ O	"	m.p.	119.6°C

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99	CH ₃	CHF ₂ O	Cl	<div><div><div>H</div><div></div></div><div>OOCCCH₂O</div></div>	Cl	m.p. 91.8°C
100	"	"	"	C ₂ H ₅ O(CH ₂) ₂ OOCCCH ₂ O	"	nD 1.5261 (22.9°C)
101	"	"	"	C ₂ H ₅ O(CH ₂) ₂ O(CH ₂) ₂ OOCCCH ₂ O	"	nD 1.5287 (23.0°C)
102	"	"	"	<div><div><div></div><div></div></div><div>CH₂OOCCCH₂O</div></div>	"	m.p. 139.1°C
103	"	"	"	HOOCCH(CH ₃)O	"	m.p. 158.7°C
104	"	"	"	<div><div><div></div><div></div></div><div><div><div></div><div></div></div><div>⊕ ⊖</div><div>i-C₃H₇NH₃OOCCCH(CH₃)O</div></div></div>	"	Viscous substance
105	"	"	"	C ₂ H ₅ OOCCCH(CH ₃)O	"	nD 1.5238 (25.7°C)
106	"	"	"	n-C ₄ H ₉ OOCCCH(CH ₃)O	"	nD 1.5253 (18.5°C)
107	"	"	"	CH ₂ =CHCH ₂ OOCCCH(CH ₃)O	"	m.p. 84.3°C
108	"	"	"	CH≡CCH ₂ OOCCCH(CH ₃)O	"	m.p. 109.7°C
109	"	"	"	CH ₃ NHCOCH(CH ₃)O	"	nD 1.5365 (35.2°C)
110	"	"	"	(CH ₃) ₂ NCOCH(CH ₃)O	"	m.p. 148.9°C
111	"	"	"	HS	"	nD 1.5828 (21.8°C)

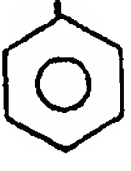
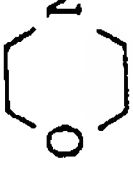
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	CH ₃	CHF ₂ O	Cl		Cl	m.p.	
112	CH ₃			CH ₃ S		95.7°C	
113	"	"	"	C ₂ H ₅ S	"	110.5°C	
114	"	"	"	i-C ₃ H ₇ S	"	83.8°C	
115	"	"	"	n-C ₄ H ₉ S	"	155.4°C	
116	"	"	"	CHF ₂ S	"	123.4°C	
117	"	"	"	CH ₂ =CHCH ₂ S	"	52.0 - 55.0°C	
118	"	"	"	CH ₂ =CHCH ₂ SO	"	91.5°C	
119	"	"	"	CH≡CCH ₂ S	"	127 - 129°C	
120	"	"	"		"	1.5972 (20.2°C)	
121	"	"	"	(C ₂ H ₅ O) ₂ P(O)CH ₂ S	"	1.5245 (25.3°C)	
122	"	"	"	CH ₃ OOC	"	1.5656 (17.2°C)	
123	"	"	"	CH ₃ OOCCH ₂ S	"	102.0°C	
124	"	"	"	C ₂ H ₅ OOCCH ₂ S	"	171.5°C	
125	"	"	"	CH ₂ =CHCH ₂ OOCCH ₂ S	"	102.4°C	

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126	CH ₃	CHF ₂ O	Cl	HOOCCH(CH ₃)S	Cl	m.p. 188.9°C
127	"	"	"	$\oplus \ominus$ NaOOCCH(CH ₃)S	"	m.p. 205.4°C
128	"	"	"	$\oplus \ominus$ i-C ₃ H ₇ NH ₃ OOCCH(CH ₃)S	"	Viscous substance
129	"	"	"	CH ₃ OOCCH(CH ₃)S	"	nD 1.5654 (19.8°C)
130	"	"	"	C ₂ H ₅ OOCCH(CH ₃)S	"	nD 1.5565 (28.0°C)
131	"	"	"	C ₂ H ₅ OOCCH(CH ₃)S(O)	"	nD 1.5582 (20.0°C)
132	"	"	"	C ₂ H ₅ OOCCH(CH ₃)SO ₂	"	m.p. 105.8°C
133	"	"	"	i-C ₃ H ₇ OOCCH(CH ₃)S	"	nD 1.5417 (22.1°C)
134	"	"	"	n-C ₄ H ₉ OOCCH(CH ₃)S	"	nD 1.5458 (20.2°C)
135	"	"	"	(CH ₃) ₂ CH(CH ₂) ₂ OOCCH(CH ₃)S	"	nD 1.5437 (22.2°C)
136	"	"	"	CH ₂ =CHCH ₂ OOCCH(CH ₃)S	"	nD 1.5600 (20.4°C)
137	"	"	"	CH≡CCH ₂ OOCCH(CH ₃)S	"	nD 1.5602 (22.2°C)
138	"	"	"	CH ₃ O(CH ₂) ₂ O(CH ₂) ₂ OOCCH(CH ₃)S	"	nD 1.5431 (23.5°C)
139	"	"	"	(C ₂ H ₅ O) ₂ P(O)CH ₂ OOCCH(CH ₃)S	"	nD 1.5386 (21.2°C)
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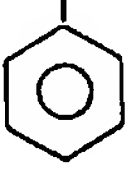
140	CH ₃	CHF ₂ O	Cl	CH ₃ NHCOCH(CH ₃)S	Cl	m.p.	99.9°C
141	"	"	"	(CH ₃) ₂ NCOCH(CH ₃)S	"	nD	1.5672 (19.9°C)
142	"	"	"	(C ₂ H ₅) ₂ NCOCH(CH ₃)S	"	nD	1.5532 (19.8°C)
143	"	"	"	i-C ₃ H ₇ NHCOCH(CH ₃)S	"	m.p.	99.9°C
144	"	"	"	 NHCOCH(CH ₃)S	"	m.p.	132.3°C
145	"	"	"	 N - COCH(CH ₃)S	"	nD	1.5457 (19.9°C)
146	"	"	"	NO ₂	"	m.p.	99.1°C
147	"	"	"	NH ₂	"	nD	1.5743 (20.8°C)
148	"	"	"	CH ₃ SO ₂ NH	"	m.p.	142.9°C
149	"	"	"	CH ₃ NHSO ₂ NH	"	m.p.	133.0°C
150	"	"	"	CH ₃ NH	"	nD	1.4952 (25.2°C)
151	"	"	"	(CH ₃) ₂ N	"	nD	1.5561 (24.9°C)
152	"	"	"	CH ₂ =CHCH ₂ NH	"	nD	1.5719 (25.3°C)
153	"	"	"	CH≡CCH ₂ NH	"	m.p.	76.3°C

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154	CH ₃	CHF ₂ O	Cl	CH ₃ OOCNH $\ominus \oplus$ Cl H ₃ N	Cl	m.p. 162.3°C
155	"	"	"	$\ominus \oplus$	"	m.p. 137.1°C
156	"	"	"	$\text{CH}_3\text{--}\langle\bigcirc\rangle\text{--SO}_3\text{H}_3\text{N}^{\oplus}$	"	m.p. 244.1°C
157	"	"	"	CH ₃ OOCCH ₂ NH	"	m.p. 96.2°C
158	"	"	"	HOOCCH(CH ₃) NH	"	m.p. 192.3°C
159	"	"	"	$\oplus \ominus$ Na OOCCH(CH ₃) NH	"	m.p. 248.0 (decomp.)
160	"	"	"	$\oplus \ominus$ i-C ₃ H ₇ NH ₃ OOCCH(CH ₃) NH	"	Viscous substance
161	"	"	"	C ₂ H ₅ OOCCH(CH ₃) NH	"	nD 1.5371 (23.4°C)
162	"	"	"	CH ₃ NHOCCH(CH ₃) NH	"	nD 1.5500 (25.6°C)
163	"	"	"	(CH ₃) ₂ NOCCH(CH ₃) NH	"	nD 1.5510 (25.2°C)
164	"	"	"	CH ₃ NHSO ₂	"	nD 1.5461 (17.9°C)
165	"	"	"	(CH ₃) ₂ NSO ₂	"	m.p. 135.2°C
166	"	"	"	(C ₂ H ₅) ₂ NCOS	"	nD 1.5623 (17.7°C)

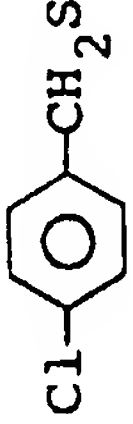
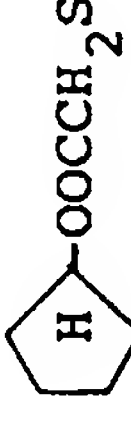
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167	CH ₃	CHF ₂ S	Cl	CH ₃ O	Cl	nD	1.5767 (21.9°C)
168	"	"	"	C ₂ H ₅ O	"	nD	1.5810 (23.8°C)
169	"	"	"	i-C ₃ H ₇ O	"	nD	1.5536 (24.2°C)
170	"	"	"	n-C ₆ H ₁₃ O	"	nD	1.5460 (24.2°C)
171	"	"	"	CH ₂ =CHCH ₂ O	"	nD	1.5791 (19.8°C)
172	"	"	"	CH≡CCH ₂ O	"	nD	1.5720 (20.0°C)
173	"	"	"	 -CH ₂ O	"	nD	1.6226 (24.3°C)
174	"	"	"	CH ₃ OOCCH ₂ O	"	nD	1.5629 (23.1°C)
175	"	"	"	C ₂ H ₅ OOCCH(CH ₃)O	"	nD	1.5569 (23.6°C)
176	"	"	"	HS	"	nD	1.6218 (23.5°C)
177	"	"	"	CH ₃ S	"	nD	1.6069 (21.7°C)
178	"	"	"	C ₂ H ₅ S	"	nD	1.6047 (24.1°C)
179	"	"	"	i-C ₃ H ₇ S	"	nD	1.5893 (23.6°C)
180	"	"	"	n-C ₅ H ₁₁ S	"	nD	1.5881 (25.0°C)

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181	CH ₃	CHF ₂ S	Cl	CHF ₂ S	Cl	Chemical Structure	Cl	nD	1.5869 (24.2°C)
182	"	"	"	"	"	CHF ₂ S	"	nD	1.6173 (21.3°C)
183	"	"	"	"	"	CH ₂ =CHCH ₂ S	"	nD	1.6261 (23.4°C)
184	"	"	"	"	"	CH≡CCH ₂ S	"	nD	1.5816 (22.3°C)
185	"	"	"	"	"	C ₂ H ₅ OCH ₂ CH ₂ S	"	nD	1.6222 (24.2°C)
186	"	"	"	"	"	CH ₃ SCH ₂ S	"	m.p.	95.9°C
						NCCH ₂ S			
187	"	"	"	"	"		"	nD	1.6273 (24.3°C)
188	"	"	"	"	"	C ₂ H ₅ OOC	"	nD	1.5720 (21.8°C)
189	"	"	"	"	"	CH ₃ OOCCH ₂ S	"	m.p.	112.1°C
190	"	"	"	"	"	C ₂ H ₅ OOCCH ₂ S	"	m.p.	97.8°C
191	"	"	"	"	"		"	m.p.	104.4°C
192	"	"	"	"	"	CH ₃ NHOCCH ₂ S	"	m.p.	138.6°C
193	"	"	"	"	"	(CH ₃) ₂ NOCCH ₂ S	"	m.p.	138.3°C

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	CH ₃	CHF ₂ S	Cl	HOOCCH(CH ₃)S ⊕⊖ NaOOCCH(CH ₃)S ⊕⊖ i-C ₃ H ₇ NH ₃ OOCCH(CH ₃)S C ₂ H ₅ OOCCH(CH ₃)S n-C ₃ H ₇ OOCCH(CH ₃)S i-C ₃ H ₇ OOCCH ₂ S (CH ₃) ₂ CHCH ₂ OOCCH ₂ S CH ₂ =CHCH ₂ OOCCH(CH ₃)S C ₂ H ₅ OOC(CH ₂) ₃ S CH ₃ NHOCCH(CH ₃)S (CH ₃) ₂ NOCCH(CH ₃)S NO ₂ NH ₂ (CH ₃) ₂ N	Cl	m.p. 149.5°C m.p. 220.8°C Viscous substance nD 1.5788 (23.4°C) nD 1.5643 (19.4°C) nD 1.5695 (19.4°C) nD 1.5812 (21.2°C) nD 1.5800 (19.4°C) m.p. 110.7°C m.p. 140.8°C nD 1.5791 (21.0°C) m.p. 109.6°C nD 1.5837 (20.3°C) nD 1.5838 (21.6°C)
194						
195	"	"	"		"	
196	"	"	"		"	
197	"	"	"		"	
198	"	"	"		"	
199	"	"	"		"	
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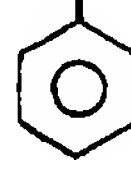
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208	CH ₃	CHF ₂ S	Cl	CH ₂ =CHCH ₂ NH	Cl	nD	1.6060 (21.2°C)
209	"	"	"	(CH ₂ =CHCH ₂) ₂ N	"	nD	1.5884 (24.0°C)
210	"	"	"	CH≡CCH ₂ NH	"	nD	1.6052 (23.1°C)
211	"	"	"	CH ₃ SO ₂ NH	"	m.p.	165.7°C
212	"	"	"	CH ₃ NHSO ₂ NH	"	nD	1.5812 (20.4°C)
213	"	"	"	CH ₃ OOCNH	"	m.p.	144.5°C
214	"	"	"	(C ₂ H ₅) ₂ NCOS	"	nD	1.5753 (21.5°C)
215	"	CH ₃ S	"	C ₂ H ₅ OOCCH(CH ₃)O	"	nD	1.5584 (25.7°C) Dimethyl sulfate quaternary ammonium salt
216	"	CHF ₂ O	"	(CH ₃) ₃ SiCH ₂ S	"	nD	1.5568 (22.8°C)
217	"	CH ₃ SO	"	5-NO ₂	F	m.p.	151.0°C
218	"	CH ₃ S	"	5-NH ₂	"	nD	1.6260 (27.3°C)
219	"	"	"	5-SH	"	Viscous substance	
220	"	"	"	5-CH≡CCH ₂ O	"	nD	1.5980 (25.2°C)
221	"	"	"	5-CH≡CCH ₂ S	"	m.p.	93 - 96°C

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222	CH ₃	CH ₃ S	Cl	5-C ₂ H ₅ OOCCH(CH ₃)S	F	nD	1.5880 (26.4°C)
223	"	SCH ₃	"	CH ₃ O-COCH ₂ O	Cl	mp.	126.2°C
224	"	"	"	C ₂ H ₅ O-COCH ₂ O	"	mp.	106.5°C
225	"	OCHF ₂	H	CH ₂ =CHCH ₂ O	"	mp.	56.2°C
226	"	"	"	CH≡CCH ₂ O	"	mp.	100.0°C
227	"	"	Cl	OHC	"	mp.	138.2°C
228	"	"	"	CH ₂ =C(CH ₃)CH ₂ O	"	nD	1.5387 (24.3°C)
229	"	"	"	(CH ₃) ₃ SiCH ₂ O	"	nD	1.5271 (23.9°C)
230	"	"	"	HO-COCH ₂ O	"	mp.	162.0°C
231	"	"	"	$\begin{matrix} \oplus & \ominus \\ \text{Na} & \text{O-COCH}_2\text{O} \end{matrix}$	"	mp.	238.0°C
232	"	"	"	$\begin{matrix} \oplus & \ominus \\ \text{K} & \text{O-COCH}_2\text{O} \end{matrix}$	"	mp.	194.0°C
233	"	"	"	$\begin{matrix} \oplus & \ominus \\ \text{i-C}_3\text{H}_7\text{NH}_3 & \text{O-COCH}_2\text{O} \end{matrix}$	"	mp.	131.0°C
234	"	"	"	C ₂ H ₅ O-COCH ₂ O	"	mp.	102.3°C
235	"	"	"	t-C ₄ H ₉ O-COCH ₂ O	"	mp.	92.2°C

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236	CH ₃	OCHF ₂	Cl	CH ₂ =CHC(CH ₃) ₂ O-COCH ₂ O	Cl	mp.	92.8°C
237	"	"	"	CH ₃ C≡CCH ₂ O-COCH ₂ O	"	mp.	104.3°C
238	"	"	"	CH ₃ S-COCH ₂ O	"	mp.	117.5°C
239	"	"	"	C ₂ H ₅ S-COCH ₂ O	"	mp.	86.0°C
240	"	"	"	i-C ₃ H ₇ S-COCH ₂ O	"	mp.	98.0°C
241	"	"	"	CH ₃ NHCOCH ₂ O	"	mp.	149.7°C
242	"	"	"	(CH ₃) ₂ NCOCH ₂ O	"	mp.	153.3°C
243	"	"	"	n-C ₄ H ₉ NHCOCH ₂ O	"	mp.	106.9°C
244	"	"	"	H ₂ C=CHCH ₂ NHCOCH ₂ O	"	mp.	127.7°C
245	"	"	"	(H ₂ C=CHCH ₂) ₂ NCOCH ₂ O	"	mp.	80.0°C
246	"	"	"	 -NHCOCH ₂ O	"	mp.	205.6°C
247	"	"	"	 -N(CH ₃)COCH ₂ O	"	nD	1.5570 (18.7°C)
248	"	"	"	CH ₃ SO ₂ NHCOCH ₂ O	"	mp.	124.5°C
249	"	"	"	(CH ₃) ₃ SiCH ₂ O-COCH ₂ O	"	nD	1.5223 (29.5°C)

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250	CH ₃	OCHF ₂	Cl	CH ₃ OC ₂ H ₄ OC ₂ H ₄ O-COCH(CH ₃)O	Cl	nD	1.5142 (23.9°C)
251	"	"	"	C ₂ H ₅ O-COCH(C ₂ H ₅)O	"	nD	1.5286 (26.5°C)
252	"	"	"	C ₂ H ₅ O-CO(CH ₂) ₃ O	"	nD	1.5238 (17.3°C)
253	"	"	"	CH ₃ O-COCH(i-C ₃ H ₇)O	"	nD	1.5273 (24.3°C)
254	"	"	"	CH ₃ SO ₂ NHCOCH(CH ₃)O	"	mp.	51.6°C
255	"	"	"	OHC	F	mp.	107.3°C
256	"	"	"	CH ₂ =CHCH ₂ O	"	mp.	63.7 - 64.1°C
257	"	"	"	CH ₂ =C(CH ₃)CH ₂ O	"	nD	1.5376 (17.3°C)
258	"	"	"	CH≡CCH ₂ O	"	mp.	98.0 - 98.1°C
259	"	"	"	CH ₃ C≡CCH ₂ O	"	mp.	108.5°C
260	"	"	"	HO-COCH ₂ O	"	mp.	143.7°C
261	"	"	"	CH ₃ O-COCH ₂ O	"	mp.	122.8 - 123.1°C
262	"	"	"	C ₂ H ₅ O-COCH ₂ O	"	mp.	127.6°C
263	"	"	"	n-C ₃ H ₇ O-COCH ₂ O	"	mp.	97.6 - 97.8°C
264	"	"	"	i-C ₃ H ₇ O-COCH ₂ O	"	mp.	120.3 - 120.5°C

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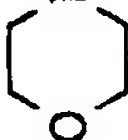
265	CH ₃	OCHF ₂	Cl	n-C ₄ H ₉ O-COCH ₂ O	F	mp.	91.4 - 91.6°C
266	"	"	"	CH ₂ =CHCH ₂ O-COCH ₂ O	"	mp.	89.2 - 89.4°C
267	"	"	"	CH≡CCH ₂ O-COCH ₂ O	"	mp.	99.0°C
268	"	"	"	C ₂ H ₅ S-COCH ₂ O	"	mp.	69.3°C
269	"	"	"	i-C ₃ H ₇ S-COCH ₂ O	"	mp.	83.8°C
270	"	"	"	HO-COCH(CH ₃)O	"	mp.	94.5°C
271	"	"	"	CH ₃ O-COCH(CH ₃)O	"	mp.	95.6°C
272	"	"	"	C ₂ H ₅ O-COCH(CH ₃)O	"	mp.	67.0 - 67.2°C
273	"	"	"	CH ₂ =CHCH ₂ O-COCH(CH ₃)O	"	Viscous substance	
274	"	"	"	CH≡CCH ₂ O-COCH(CH ₃)O	"	nD	1.5266 (25.8°C)
275	"	"	"	C ₂ H ₅ O-COCH(C ₂ H ₅)O	"	nD	1.5105 (22.6°C)
276	"	"	"	CH ₂ =C(CH ₃)CH ₂ S	Cl	nD	1.5775 (21.7°C)
277	"	"	"	CH ₂ =C(Cl)CH ₂ S	"	nD	1.5865 (21.5°C)
278	"	"	"	ClCH=CHCH ₂ S	"	nD	1.5867 (21.2°C)
279	"	"	"	ClCH=C(Cl)CH ₂ S	"	nD	1.5926 (22.8°C)
280	"	"	"	NCCH ₂ S	"	mp.	122.3°C

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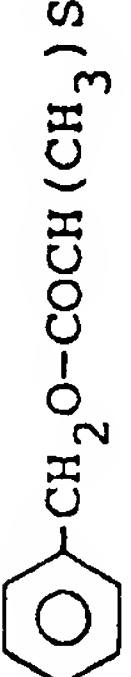

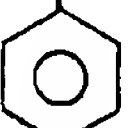
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281	"	"	"	$\begin{matrix} \oplus & \ominus \\ (\text{i-C}_3\text{H}_7)\text{NH}_3 & \text{O-COCH}_2\text{S} \end{matrix}$	"	232.0°C	
282	"	"	"	HO-COCH ₂ S	"	113.0°C	
283	"	"	"	i-C ₃ H ₇ O-COCH ₂ S	"	164.2°C	
284	"	"	"	n-C ₄ H ₉ O-COCH ₂ S	"	99.1°C	
285	"	"	"	C ₁₂ H ₂₅ O-COCH ₂ S	"	62.0°C	
286	"	"	"	CH ₃ S-COCH ₂ S	"	1.5010 (27.2°C)	
287	"	"	"	C ₂ H ₅ S-COCH ₂ S	"	98.3°C	
288	"	"	"	i-C ₃ H ₇ S-COCH ₂ S	"	73.1°C	
289	"	"	"	CH≡CCH ₂ O-COCH ₂ S	"	97.0°C	
290	"	"	"	C ₂ H ₅ OC ₂ H ₄ OC ₂ H ₄ O-COCH ₂ S	"	85.9°C	
291	"	"	"	 -O-COCH ₂ S	"	76.5°C	
292	"	"	"	 -CH ₂ O-COCH ₂ S	"	87.8°C	
293	"	"	"		"	82.8°C	

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	CH ₃	OCHF ₂	Cl		Cl	nD	1.5387 (29.1°C)
294				(CH ₃) ₃ SiCH ₂ O-COCH ₂ S			
295	"	"	"	CH ₃ NHCOCH ₂ S	"	mp.	177.8°C
296	"	"	"	(CH ₃) ₂ NCOCH ₂ S	"	mp.	121.6°C
297	"	"	"	H ₂ C=CHCH ₂ NHCOCH ₂ S	"	mp.	146.6°C
298	"	"	"	(H ₂ C=CHCH ₂) ₂ NCOCH ₂ S	"	mp.	71.8°C
299	"	"	"	HC≡CCH ₂ NHCOCH ₂ S	"	mp.	172.0°C
300	"	"	"		"	mp.	178.1°C
301	"	"	"		"	nD	1.5661 (21.5°C)
302	"	"	"		"	mp.	120.4°C
303	"	"	"		"	mp.	126.5°C
304	"	"	"	i-C ₄ H ₉ O-COCH(CH ₃)S	"	Viscous substance	
305	"	"	"	t-C ₄ H ₉ CH ₂ O-COCH(CH ₃)S	"	nD	1.5462 (13.9°C)

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306	CH ₃	CH ₃	Cl	n-C ₆ H ₁₃ O-COCH(CH ₃)S	Cl	nD	1.5210 (23.2°C)
307	"	"	"		"		Viscous substance
308	"	"	"	C ₂ H ₅ S-COCH(CH ₃)S	"	nD	1.5790 (24.0°C)
309	"	"	"	i-C ₃ H ₇ S-COCH(CH ₃)S	"	nD	1.5669 (24.2°C)
310	"	"	"	NCCH(CH ₃)O-COCH(CH ₃)S	"	nD	1.5501 (23.1°C)
311	"	"	"	CH ₃ SC ₂ H ₄ O-COCH(CH ₃)S	"	nD	1.5609 (23.1°C)
312	"	"	"		"		Viscous substance
313	"	"	"		"	nD	1.5956 (14.3°C)
314	"	"	"	(CH ₃) ₃ SiCH ₂ O-COCH(CH ₃)S	"	nD	1.5429 (26.8°C)
315	"	"	"	CH ₂ =CHCH ₂ NHCOCH(CH ₃)S	"	nD	1.5619 (22.6°C)
316	"	"	"	(H ₂ C=CHCH ₂) ₂ NCOCH(CH ₃)S	"	nD	1.5603 (22.9°C)
317	"	"	"	HC≡CCH ₂ NHCOCH(CH ₃)S	"	nD	1.5496 (23.1°C)
318	"	"	"	CH ₃ CO-O-(CH ₂) ₂ S	"	nD	1.5603 (26.9°C)
319	"	"	"	C ₂ H ₅ O-COC(CH ₃) ₂ S	"	mp.	96.4°C

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320	CH ₃	OCHF ₂	Cl	CH ₃ O-COCH(i-C ₃ H ₇)S	Cl	nD	1.5572 (23.4°C)
321	"	"	"	C ₂ H ₅ O-COCH(CH ₃)N(CH ₃)	"	nD	1.5365 (22.8°C)
322	"	"	"	HO-CO	"	mp.	216.7°C
323	"	"	"	CH ₃ O-CO	"	mp.	63.9°C
324	"	"	"	C ₂ H ₅ O-CO	"	nD	1.5446 (26.8°C)
325	"	"	"	CH ₂ =CHCH ₂ O-CO	"	nD	1.5317 (26.8°C)
326	"	"	"	CH ₃ O-COCH(CH ₃)O-CO	"	nD	1.5370 (25.7°C)
327	"	"	"	C ₂ H ₅ O-COCH(CH ₃)O-CO	"	nD	1.5672 (26.0°C)
328	"	"	"	CH≡CCH ₂ O-CO	"	mp.	78.5°C
329	"	"	Br	BrCH ₂ CH(Br)CH ₂ O	"	nD	1.5862 (22.3°C)
330	"	"	"	BrC≡CCH ₂ O	"	nD	1.5603 (22.9°C)
331	"	"	"	CH ₃ O-COCH ₂ O	"	mp.	133.8°C
332	"	"	"	C ₂ H ₅ O-COCH(CH ₃)O	"	nD	1.5396 (20.8°C)
333	"	SCHF ₂	Cl	NCCH ₂ O	"	nD	1.5571 (26.1°C)
334	"	"	"	CH ₂ =C(CH ₃)CH ₂ O	"	nD	1.5767 (26.3°C)



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335	CH ₃	SCHF ₂	Cl	ClCH=CHCH ₂ O	Cl	nD	1.5930 (25.8°C)
336	"	"	"	CH ₃ OCH ₂ O	"	nD	1.5765 (23.8°C)
337	"	"	"	C ₂ H ₅ OC ₂ H ₄ O	"	nD	1.5713 (23.4°C)
338	"	"	"	CH ₃ OCH ₂ S	"	mp.	82.6°C
339	"	"	"	ClCH=CHCH ₂ S	"	nD	1.6194 (26.3°C)
340	"	"	"	CH ₂ =C(Cl)CH ₂ S	"	nD	1.6137 (26.2°C)
341	"	OCHF ₂	"	HS	F	mp.	72.6°C
342	"	"	"	CH ₃ S	"	mp.	84.5°C
343	"	"	"	F ₂ HCS	"	nD	1.5377 (24.0°C)
344	"	"	"	CH ₂ =CHCH ₂ S	"	nD	1.5670 (17.9°C)
345	"	"	"	CH ₂ =C(CH ₃)CH ₂ S	"	nD	1.5637 (19.1°C)
346	"	"	"	CH≡CCH ₂ S	"	mp.	82.8°C
347	"	"	"	C ₂ H ₅ OC ₂ H ₄ S	"	nD	1.5523 (24.0°C)
348	"	"	"	HO-COCH ₂ S	"	mp.	132.2°C
349	"	"	"	CH ₃ O-COCH ₂ S	"	nD	1.5579 (24.0°C)

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350	CH ₃	OCHF ₂	Cl	C ₂ H ₅ O-COCH ₂ S	F	nD	1.5342 (17.9°C)
351	"	"	"	C ₂ H ₅ O-COCH ₂ S(O)	"	nD	1.5420 (25.0°C)
352	"	"	"	n-C ₄ H ₉ O-COCH ₂ S	"	nD	1.5391 (24.1°C)
353	"	"	"	CH ₂ =CHCH ₂ O-COCH ₂ S	"	nD	1.5562 (24.1°C)
354	"	"	"	CH ₃ SCH ₂ O-COCH ₂ S	"	mp.	57.9°C
355	"	"	"		"	mp.	58.9°C
356	"	"	"		"	nD	1.5794 (17.9°C)
357	"	"	"	CH ₃ NHCOCH ₂ S	"	mp.	147.0°C
358	"	"	"	(CH ₃) ₂ NCOCH ₂ S	"	mp.	141.2°C
359	"	"	"	HO-COCH(CH ₃)S	"	nD	1.5463 (19.1°C)
360	"	"	"	CH ₃ O-COCH(CH ₃)S	"	nD	1.5494 (25.0°C)
361	"	"	"	C ₂ H ₅ O-COCH(CH ₃)S	"	nD	1.5328 (18.0°C)
362	"	"	"	n-C ₄ H ₉ O-COCH(CH ₃)S	"	nD	1.5343 (25.1°C)

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363	CH ₃	OCHF ₂	Cl	CH ₂ =CHCH ₂ O-COCH(CH ₃)S	F	nD	1.5471(25.0°C)
364	"	"	"	CH≡CCH ₂ O-COCH(CH ₃)S	"	Viscous substance	
365	"	"	"	CH ₃ OC ₂ H ₄ OC ₂ H ₄ O-COCH(CH ₃)S	"	nD	1.5355(18.4°C)
366	"	"	"	CH ₃ SO ₂ NH	"	mp.	146.9°C
367	"	"	"	CH ₃ NH ₂ SO ₂ NH	"	mp.	131.2°C
368	"	"	"	CH ₂ =CHCH ₂ NH	"	nD	1.5424(23.6°C)
369	"	"	"	CH≡CCH ₂ NH	"	nD	1.5510(27.1°C)
370	"	"	"	C ₂ H ₅ O-COCH ₂ NH	"	mp.	111.5°C
371	"	"	"	C ₂ H ₅ O-COCH(CH ₃)NH	"	nD	1.5264(26.6°C)
372	"	"	"	CH ₃ O-CO	"	nD	1.5430(17.0°C)
373	"	"	"	C ₂ H ₅ O-CO	"	nD	1.5320(21.0°C)
374	"	"	"	CH ₂ =CHCH ₂ O-CO	"	nD	1.5218(20.5°C)
375	"	"	"	CH≡CCH ₂ O-CO	"	nD	1.5820(24.5°C)
376	"	"	"	CH ₃ O-COCH(CH ₃)O-CO	"	nD	1.5314(23.0°C)
377	"	"	"	C ₂ H ₅ O-COCH(CH ₃)O-CO	"	nD	1.5212(14.1°C)

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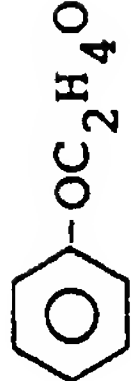


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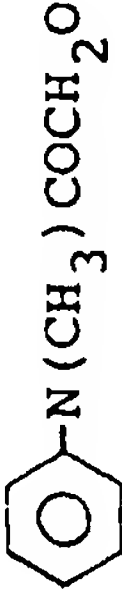


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378	CH ₃	CHF ₂	Cl		Cl	mp.	102.9°C
379	"	"	"		"	mp.	104.4°C
380	"	"	"	HO-COCH ₂ O	"	mp.	157.7°C
381	"	"	"	$\oplus \ominus$ Na O-COCH ₂ O	"	mp.	135.2°C
382	"	"	"	C ₂ H ₅ O-COCH ₂ O	"	nD	1.5736 (24.0°C)
383	"	"	"	i-C ₃ H ₇ O-COCH ₂ O	"	mp.	113.3°C
384	"	"	"	CF ₃ CH ₂ O-COCH ₂ O	"	mp.	117.5°C
385	"	"	"	CH ₂ =CHCH ₂ O-COCH ₂ O	"	mp.	77.2°C
386	"	"	"		"	nD	1.5619 (28.3°C)
387	"	"	"	CH ₃ NHCOCH ₂ O	"	mp.	116.8°C
388	"	"	"	CH ₃ O-COCH(CH ₃)O	"	nD	1.5620 (23.4°C)
389	"	"	"	i-C ₃ H ₇ O-COCH(CH ₃)O	"	nD	1.5496 (24.5°C)

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390	CH ₃	SCHF ₂	Cl	CH ₂ =CHCH ₂ O-COCH(CH ₃)O	Cl	nD	1.5643 (23.5°C)
391	"	"	"	CH ₃ NHCOCH(CH ₃)O	"	nD	1.5641 (23.5°C)
392	"	"	"	CH ₃ O-COCH(n-C ₄ H ₉)O	"	nD	1.5488 (28.1°C)
393	"	"	"	NCCH ₂ O-COCH ₂ O	"	nD	1.5694 (28.0°C)
394	"	"	"	CH ₃ O-COCH(CH ₃)S	"	nD	1.5891 (26.4°C)
395	"	"	"	NCCH ₂ O-COCH(CH ₃)S	"	nD	1.5801 (26.5°C)
396	"	"	"	C ₂ H ₅ O-CO(CH ₂) ₃ S	"	mp.	107.0°C
397	"	"	"		"	mp.	136.9°C
398	"	"	"	HO-COCH ₂ NH	"	mp.	204.0°C (decomp.)
399	"	"	"	C ₂ H ₅ O-COCH ₂ NH	"	mp.	110.0°C
400	"	"	"	C ₂ H ₅ O-COCH(CH ₃)NH	"	nD	1.5730 (23.9°C)
401	"	OCHF ₂	"		"	mp.	91.8°C
402	"	"	"		"	mp.	139.1°C

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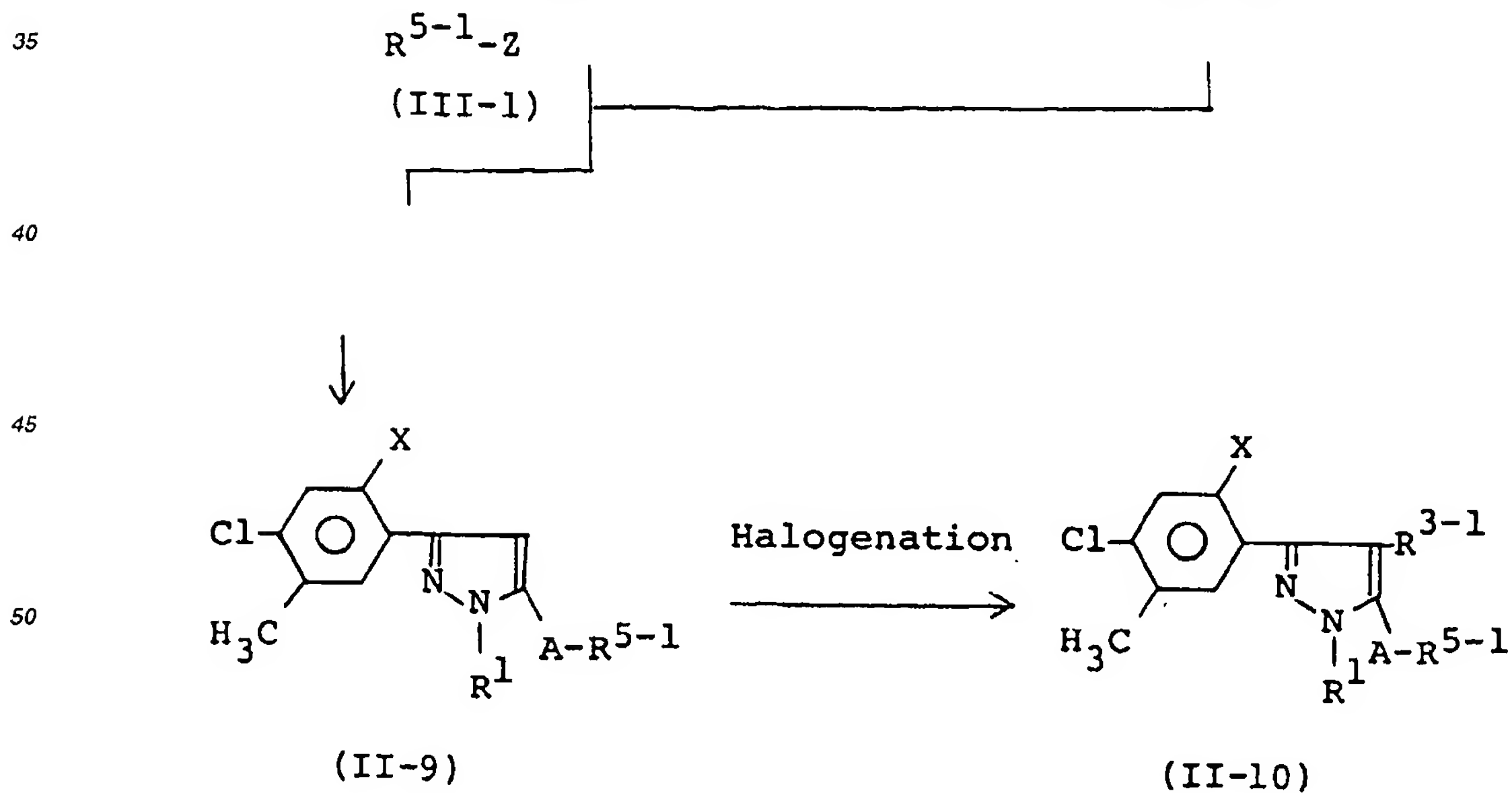
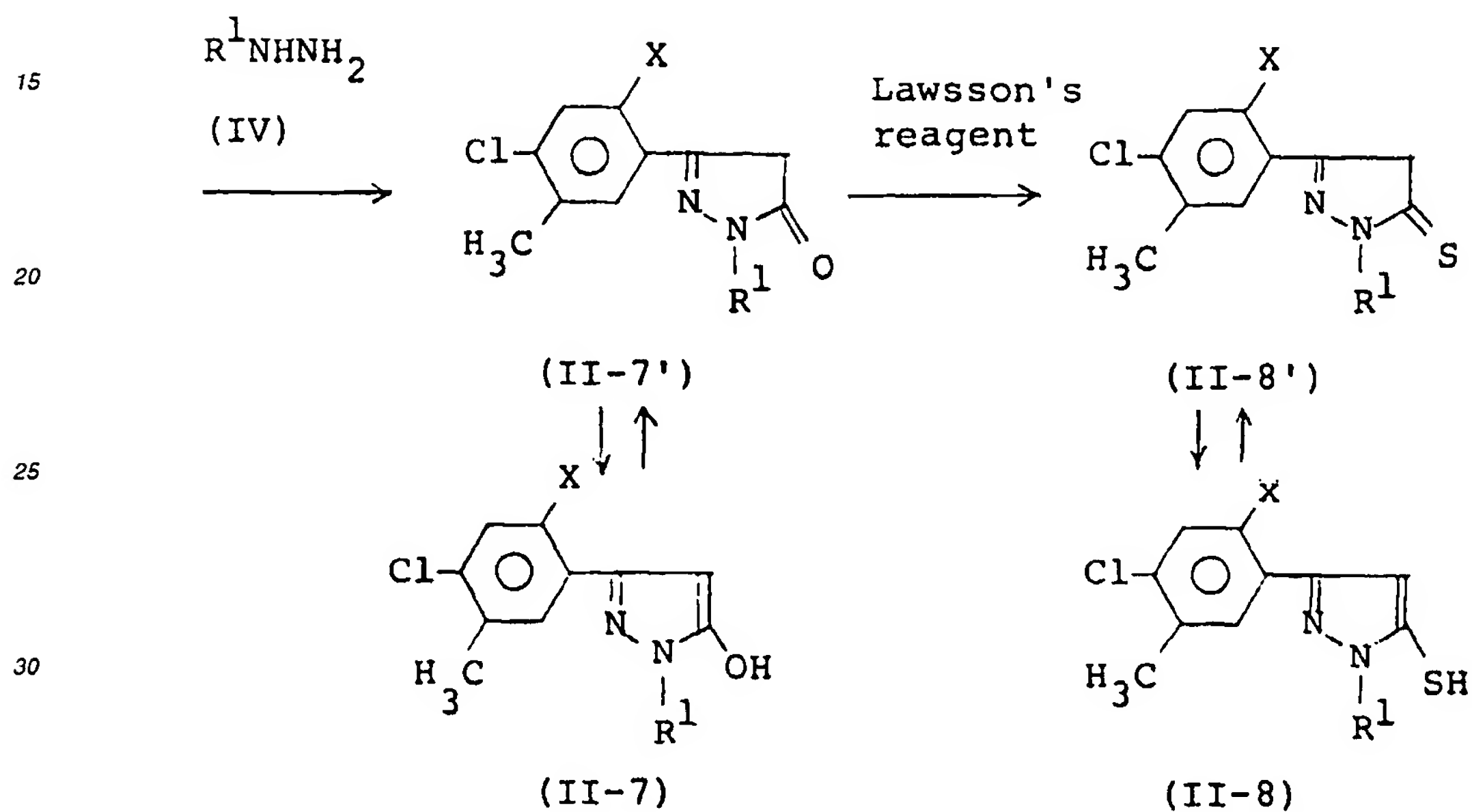
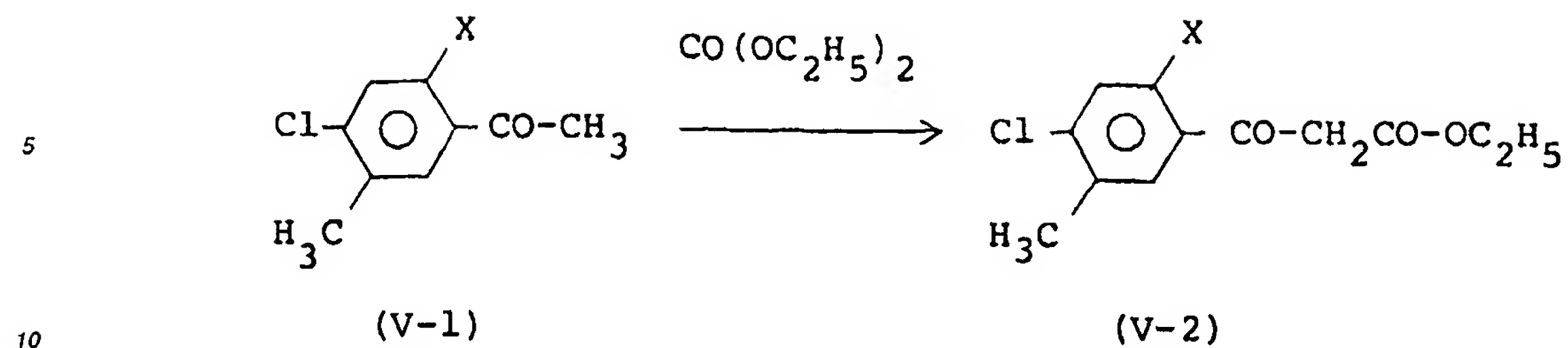
403	CH ₃	OCHF ₂	Cl	HO-CO	F	Solid substance
404	"	"	"	NO ₂	"	"
405	"	"	"	NH ₂	"	Viscous substance
406	"	"	"	HO-	"	Solid substance
407	"	"	"	$\oplus \ominus$ Na O-CO	Cl	mp. 246.4°C
408	"	"	"	CH ₃ O-CO-OCH ₂ S-CO	"	nD 1.5727 (24.1°C)
409	"	"	"	(CH ₃) ₂ NCO	F	Viscous substance
410	"	"	"	i-C ₃ H ₇ \oplus NH ₄ \ominus CO	"	nD 1.5269 (23.5°C)

Table 2 shows NMR data on viscous or crystalline substances shown in Table 1.

Table 2

Comp'd No.	NMR (δ value, CDCl ₃ /TMS, ppm)
19	2.38 (3H, s), 3.50 (2H, br), 3.96 (3H, s), 4.93 - 5.33 (2H, m), 5.46 - 6.23 (1H, m), 7.25 (1H, s), 7.43 (1H, s).
64	1.23 (3H, t), 1.51 (3H, d), 2.40 (3H, s), 3.97 (3H, s), 4.18 (2H, q), 3.80 - 4.16 (1H, m), 4.90 (1H, d), 6.64 (1H, s), 7.40 (1H, s).
104	1.13 (6H, d), 1.58 (3H, d), 2.83 - 3.45 (1H, m), 3.80 (3H, s), 4.59 (1H, q), 6.75 (1H, t), 6.92 (1H, s), 7.51 (1H, s), 7.75 - 8.35 (3H, br).
128	1.18 (6H, d), 1.49 (3H, d), 2.92 - 3.58 (1H, m), 3.81 (3H, s), 4.06 (1H, q), 6.68 (1H, t), 7.43 (2H, s), 7.62 - 8.32 (3H, br).
160	1.12 (6H, d), 1.48 (3H, d), 2.60 - 3.30 (1H, m), 3.73 (3H, s), 3.91 (1H, q), 6.55 (1H, s), 6.68 (1H, s), 7.33 (1H, s), 7.98 (3H, s).
196	1.20 (6H, d), 1.48 (3H, d), 3.28 (1H, m), 3.90 (1H, m), 4.00 (3H, s), 6.74 (1H, t, J=56 Hz), 7.45 (2H, s), 8.00 (3H, s).
219	2.28 (3H, t), 3.91 (1H), 3.99 (3H, s), 7.21 (1H, d), 7.50 (1H, d).
273	1.70 (3H, d), 3.80 (3H, s), 4.5 - 4.9 (2H), 5.0 - 5.5 (2H), 5.5 - 5.2 (1H), 6.70 (1H, t, J=73 Hz), 7.0 - 7.4 (2H).
304	0.82 (6H, d), 1.56 (3H, d), 3.80 (2H, d), 3.73 - 4.23 (2H, m), 6.70 (1H, t), 7.20 (1H, s), 7.23 (1H, s).
307	1.53 (3H, d), 3.75 (3H, s), 4.06 (1H, q), 5.10 (2H, s), 6.61 (1H, t), 7.18 (5H, s), 7.41 (1H, s).
312	1.53 (3H, d), 0.68 - 2.11 (10H, m), 3.80 (3H, s), 4.10 (1H, q), 4.33 - 4.90 (1H, br), 6.70 (1H, t), 7.20 (1H, s), 7.23 (1H, s).
364	1.54 (3H, d), 2.42 (1H, t), 3.80 (3H, s), 3.98 (1H, q), 4.61 (2H, d), 6.71 (1H, t, J=72 Hz), 7.31 (1H, d, J=9 Hz), 7.80 (1H, d, J=7.6 Hz).
404 (CDCl ₃ +d ₆ · DMSO)	3.80 (3H, s), 6.70 (1H, t, J=72 Hz), 7.22 (1H, d, J=10 Hz), 8.12 (1H, d, J=8 Hz).
405	3.83 (3H, s), 6.67 (1H, t, J=72 Hz), 7.36 (1H, d, J=10 Hz), 8.20 (1H, d, J=8 Hz).
406	3.73 (3H, s), 3.85 (2H, s, br), 6.64 (1H, t, J=72 Hz), 6.82 (1H, d, J=8 Hz), 7.04 (1H, d, J=10 Hz).
407	3.80 (3H, s), 6.00 (1H, Br), 6.65 (1H, t, 72 Hz), 7.01 (1H, d, J=10 Hz), 7.01 (1H, d, J=8 Hz)
410	3.10 (6H, s), 3.80 (3H, s), 6.70 (1H, t, J=72 Hz), 7.23 (1H, d, J=10 Hz), 7.53 (1H, d, J=8 Hz)

Pyrazole derivatives of formula (II-2) that are starting materials for 3-(substituted phenyl)-pyrazole derivatives of formula (I) or salts thereof can be produced, for instance, according to the following reaction schemes. Pyrazole derivatives of formula (II-2), in connection with their raw materials, are classified into groups of formulae (II-7), (II-8), (II-9), and (II-10).



In the above equations, R^1 , R^2 , R^5 , X , and Z are as defined above, R^{3-1} denotes halogen, and R^{5-1} denotes alkyl or haloalkyl.

As shown above, a pyrazole derivative of formula (II-8') can be prepared by reacting a compound of formula (V-1) with diethyl carbonate, reacting the resulting compound of formula (V-2) with a hydrazine of formula (IV), and reacting the resulting pyrazole derivative of formula (II-7') with Lawesson's reagent.

The pyrazole derivative of formula (II-10) can be produced from a pyrazole derivative of formula (II-7) that is the tautomer of pyrazole derivative of formula (II-7') or from a pyrazole derivative of formula (II-8) that is the tautomer of pyrazole derivative of formula (II-8') by reacting a halide of formula (III-1), followed by halogenating the resulting pyrazole derivative of formula (II-9).

Thus, the pyrazole derivatives of formula (II-2) can be produced that are raw materials used in the above production process^①.

Pyrazole derivatives of formula (II-5) that are raw materials in the process^②A can be produced from compounds of the following formula (V-3) in the same manner as stated above.



Table III shows typical examples of the compounds of formula (II-2) which have been prepared according to the above process

Formula (II-2):

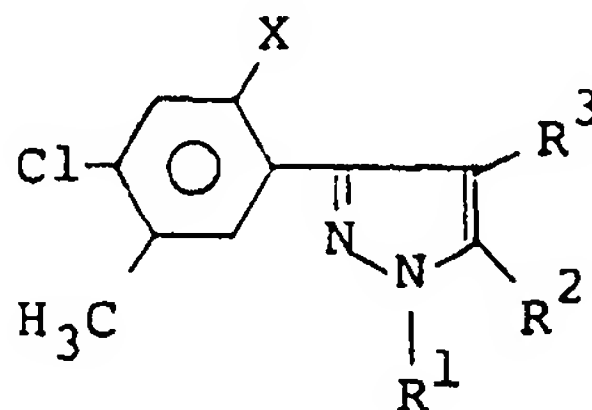


Table 3

Comp'd No.	R ¹	R ²	R ³	X	Property
II.1	CH ₃	OCH ₃	Cl	F	nD 1.5633(26.5°C)
II.2	CH ₃	OCHF ₂	H	Cl	nD 1.5467(24.4°C)
II.3	CH ₃	OCHF ₂	H	F	Oily matter
II.4	CH ₃	OCHF ₂	Cl	Cl	nD 1.5511(24.5°C)
II.5	CH ₃	OCHF ₂	Cl	F	mp. 93.7°C
II.6	CH ₃	OCHF ₂	Br	Cl	nD 1.5627(24.8°C)
II.7	CH ₃	OCHF ₂	Br	F	mp. 126.4°C
II.8	i-C ₃ H ₇	SCH ₃	H	F	Oily matter
II.9	i-C ₃ H ₇	SCH ₃	Cl	F	nD 1.5830(15.1°C)
II.10	t-C ₄ H ₉	OCHF ₂	Cl	F	nD 1.5402(24.8°C)
II.11	t-C ₄ H ₉	OCHF ₂	Br	F	nD 1.4848(24.2°C)

Table 4 shows typical examples of the pyrazole derivatives of formula (II-5)

Formula (II-5) :

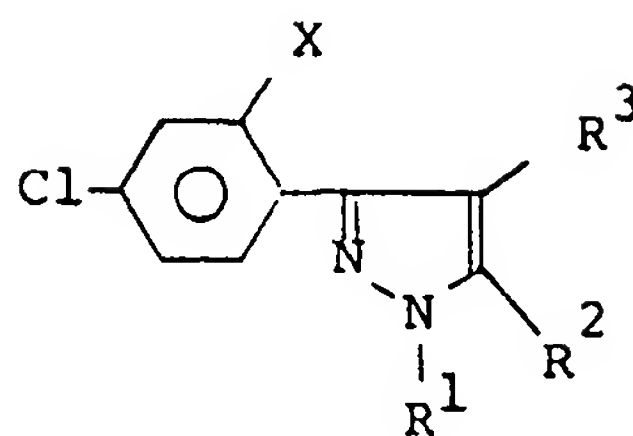


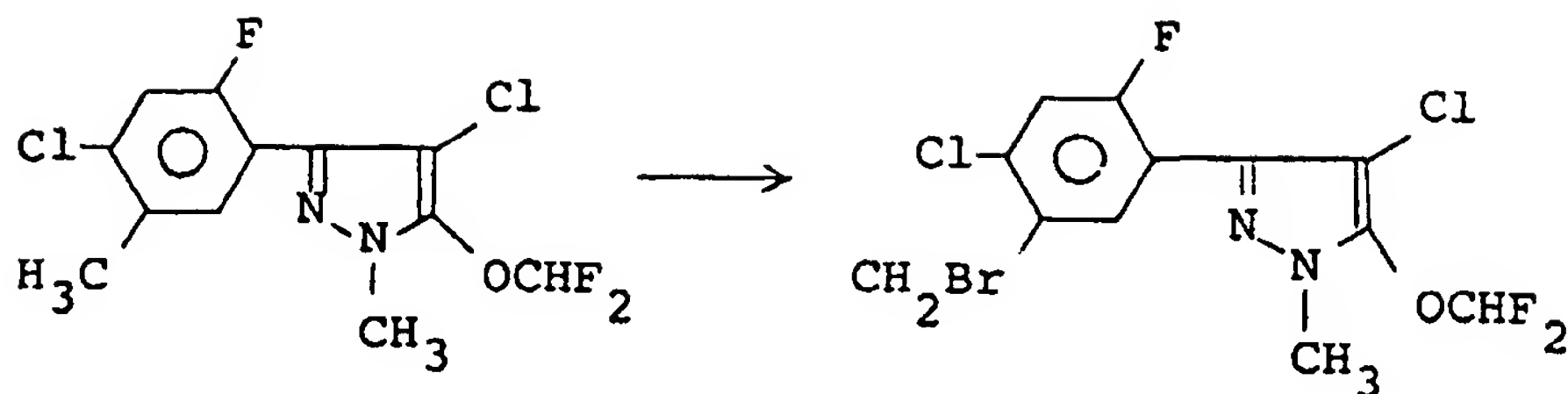
Table 4

Comp'd No.	R ¹	R ²	R ³	X	Property
II.12	CH ₃	OCH ₃	H	Cl	mp. 79.4°C
II.13	CH ₃	SCH ₃	H	Cl	mp. 59.5 - 60.0°C
II.14	CH ₃	SCH ₃	H	F	mp. 40.8°C
II.15	CH ₃	OCH ₃	Cl	Cl	mp. 71 - 73°C
II.16	CH ₃	SCH ₃	Cl	Cl	mp. 62.5 - 63.5°C
II.17	CH ₃	SCH ₃	Cl	F	nD 1.6066(21.5°C)
II.18	CH ₃	OCH ₃	Br	Cl	mp. 83 - 85°C
II.19	CH ₃	SCH ₃	Br	Cl	mp. 70.0 - 75.5°C
II.20	CH ₃	OCHF ₂	H	Cl	mp. 39.2°C
II.21	CH ₃	SCHF ₂	H	Cl	nD 1.6001 (20.2°C)
II.22	CH ₃	OCHF ₂	H	F	mp. 41.2 - 41.5°C
II.23	CH ₃	OCHF ₂	Cl	Cl	mp. 27 - 28°C
II.24	CH ₃	SCHF ₂	Cl	Cl	nD 1.5900(20.2°C)
II.25	CH ₃	OCHF ₂	Cl	F	mp. 35.9 - 36.2°C)
II.26	CH ₃	OCHF ₂	Br	Cl	nD 1.5664(21.0°C)
II.27	CH ₃	SCHF ₂	Br	Cl	nD 1.5993(18.3°C)
II.28	CH ₃	OCHF ₂	Br	F	mp. 63.9°C
II.29	CH ₃	OCHF ₂	F	Cl	nD 1.5441(20.8°C)
II.30	CH ₃	OCF ₂ CHF ₂	H	Cl	nD 1.5237(20.5°C)
II.31	CH ₃	OCF ₂ CHF ₂	Cl	Cl	nD 1.5283(10.6°C)
II.32	CH ₃	OCF ₂ CF ₂ Cl	Cl	Cl	mp. 68 - 70°C
II.33	CH ₃	OCF ₂ CHFCF ₃	H	Cl	mp. 68 - 70°C
II.34	C ₂ H ₅	OCHF ₂	Cl	Cl	nD 1.5422(27.3°C)

The present invention is illustrated with reference to typical examples thereof.

Example 1 Preparation of 2-chloro-5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-benzoic acid (compound No. 404)

1-1 Preparation of 3-(5-bromomethyl-4-chloro-2-fluorophenyl)-4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazole

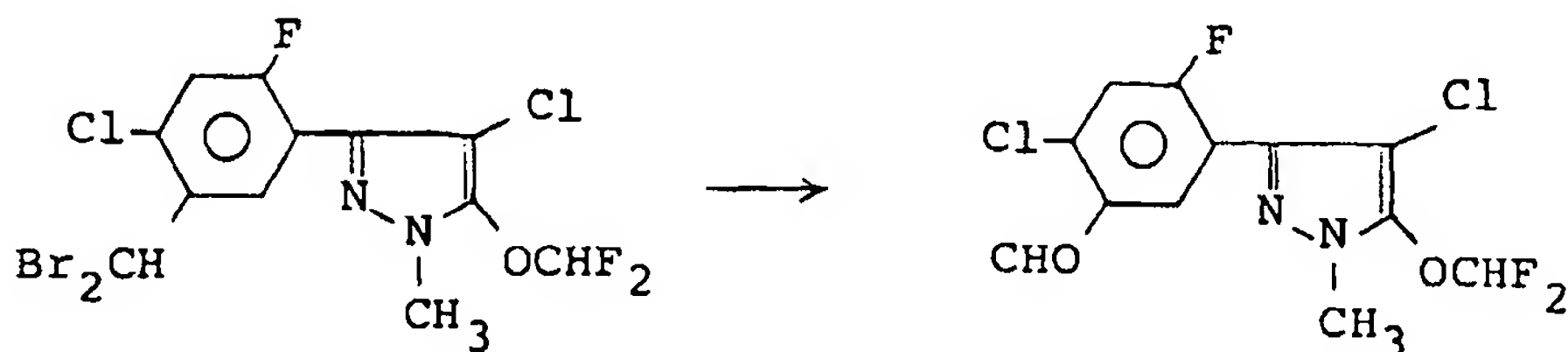


4-Chloro-3-(4-chloro-2-fluoro-5-methylphenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (6.9 g, 20 mmoles) was suspended in 100 ml carbon tetrachloride, and N-bromosuccinimide (3.97 g, 22 mmoles) together with a catalytic amount of benzoyl peroxide was added and reacted with the pyrazole derivative under reflux for 5 hours. Then the removal of insoluble matter from the product solution by filtration, concentration of the filtrate, and purification of the residue by column chromatography gave the title compound (5.64 g) as a viscous substance, yield 70%.

NMR (CDCl₃/TMS), δ (ppm):

3.83 (3H, s), 4.53 (2H, s), 6.67 (1H, t, J=72 Hz), 7.22 (1H, d, J=10 Hz), 7.60 (1H, d, J=8 Hz).

1-2 Preparation of 2-chloro-5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-fluorobenzaldehyde (compound No. 255)

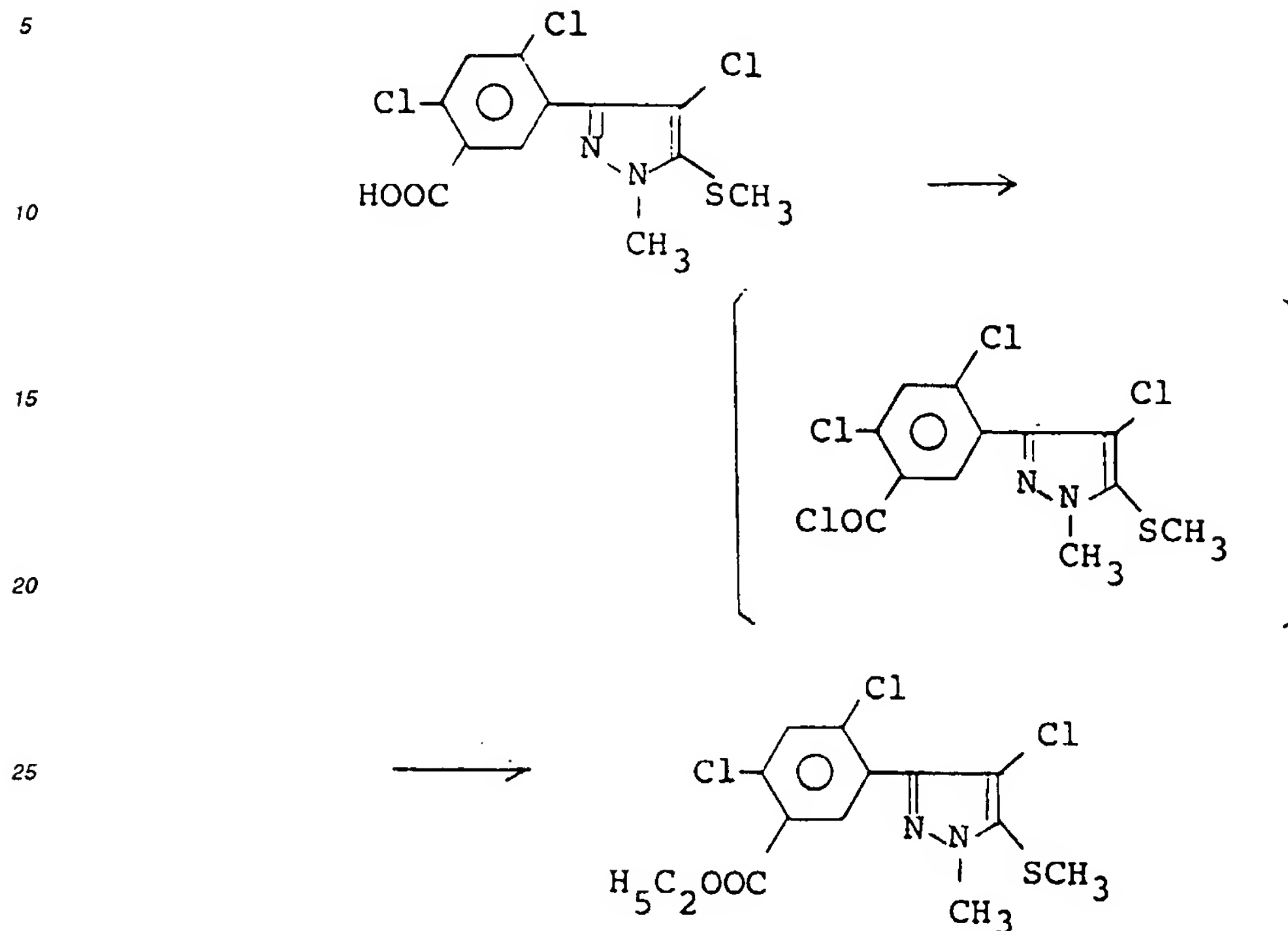


4-Chloro-3-(4-chloro-5-dibromomethyl-2-fluorophenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (14.4 g, 30 mmoles) was suspended in 50 ml conc. hydrochloric acid and reacted under reflux for 5 hours. Then the reaction product solution, poured into ice-cold water, was extracted with ethyl acetate. The extract solution was washed successively with 5% aqueous NaHCO₃ and water, and dehydrated and concentrated. The residue was purified by column chromatography, giving the title compound (6.0 g), m.p. 107.3°C, yield 59%.

1-3 Preparation of 2-chloro-5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-fluorobenzoic acid (compound No. 403)

A solution of potassium permanganate (1.1 g, 7 mmoles) in 22 ml water was dropped slowly into a suspension of 2-chloro-5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-fluorobenzaldehyde (1.7 g, 5 mmoles) in 18 ml water at 70 - 80°C to react. After dropping was finished, the reaction was further continued for 1 hour at 70 - 80°C. Then the reaction product solution was cooled to room temperature, and made alkaline with 10% aqueous NaOH. The by-product manganese dioxide was filtered off, and washed with hot water. The filtrate and the washing were mixed together, acidified with hydrochloric acid, and extracted with ether. The ether layer was washed with water, dehydrated, and evaporated to dryness, giving the title compound (1.0 g) in block form, yield 56%.

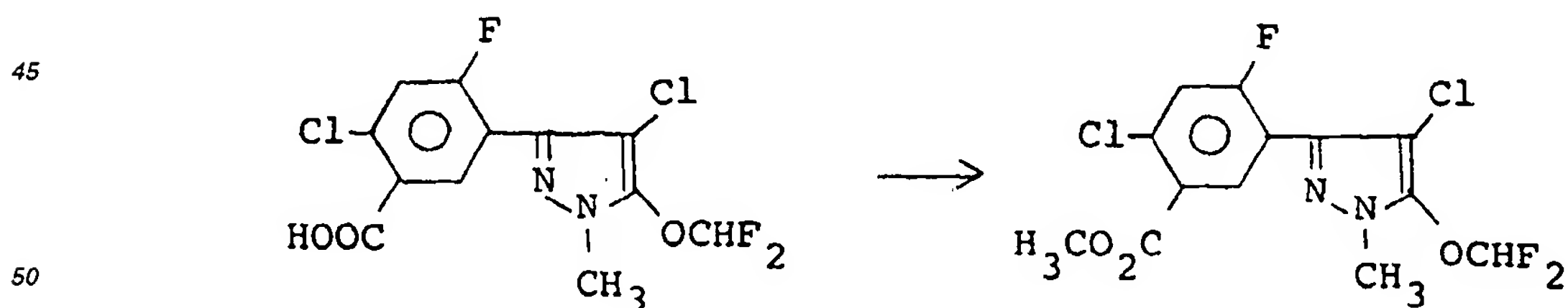
Example 2 Preparation of ethyl 5-(4-chloro-1-methyl-5-methylthio-1H-pyrazol-3-yl)-2,4-dichlorobenzoate (compound No. 50)



Ethyl 5-(4-chloro-1-methyl-5-methylthio-1H pyrazol-3-yl)-2,4-dichlorobenzoate (0.30 g, 0.85 mmole) and thionyl chloride (0.10 g, 0.85 mmole), mixed with 20 ml methylene chloride, were reacted together under reflux for 2 hours. Then the solvent was evaporated away under reduced pressure to give an acid chloride, which in turn was reacted with a large excess of ethanol under reflux for 1.5 hours. Then, removal of the solvent by evaporation under reduced pressure and purification of the resulting residue by silica gel column chromatography gave the title compound (0.11 g) in paste form, yield 33.9%.

$n_D(20.1^\circ\text{C})$: 1.6029

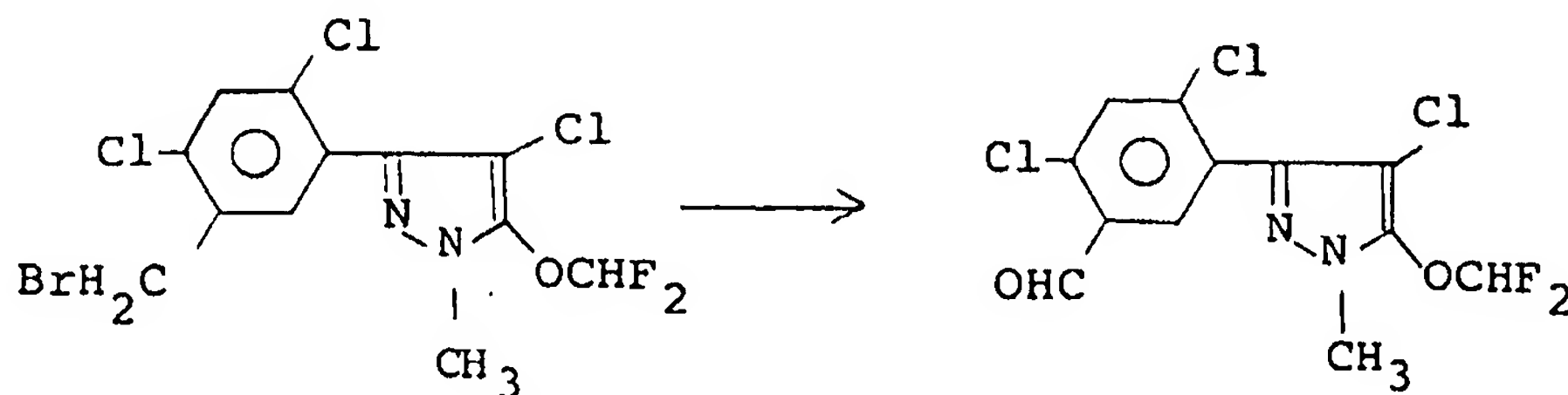
Example 3 Preparation of methyl 2-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-fluorobenzoate (compound No. 372)



A KOH powder (0.1 g, 1.8 mmols) and methyl iodide (0.26 g, 1.8 mmols) were added to a solution of 2-chloro-5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-4-fluorobenzoic acid in 20 ml acetone and reacted with this pyrazole derivative under reflux 3 hours. Then the extraction with ethylacetate, washing with water, dehydration, and concentration of the extract solution, and purification of the residue by column chromatography gave the title compound (0.15 g), yield 48%.

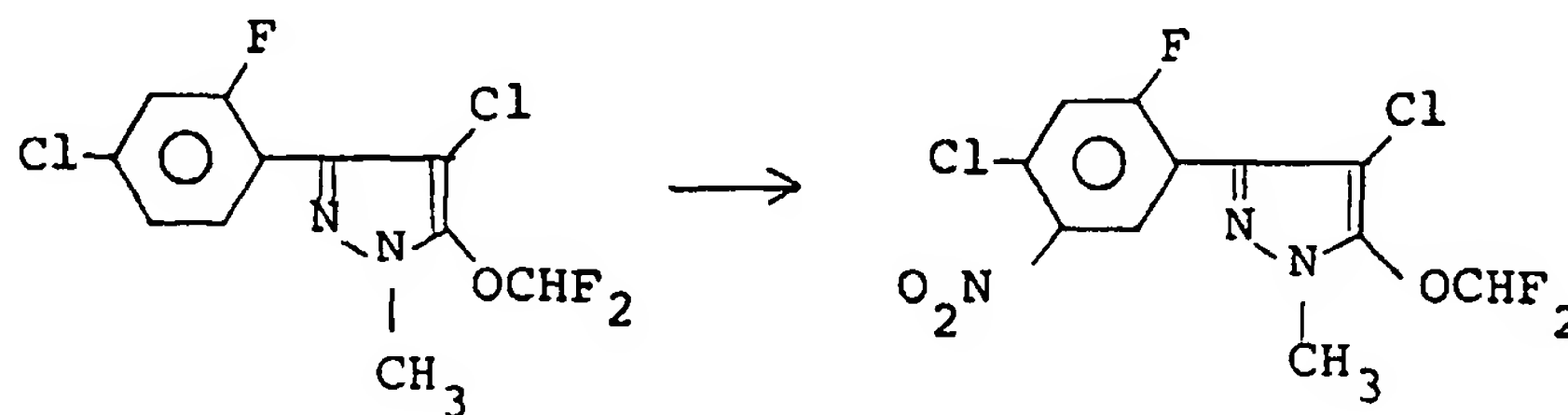
$n_D(17.0^\circ\text{C})$: 1.5430

Example 4 Preparation of 5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-2,4-dichlorobenzaldehyde (compound No. 227)



After addition of 100 ml water to a solution of 3-(5-bromomethyl-2,4-dichlorophenyl)-4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazole (26.60 g, 63.3 mmol) in 100 ml glacial acetic acid, hexamethylenetetramine (8.87 g, 63.3 mmol) was added and reacted under reflux for 2 hours. Further, 40 ml conc. sulfuric acid was added and reacted under reflux for 13 hours. Then the reaction product mixture, poured into ice-cold water, was extracted with ethyl acetate. The extract solution was washed with 5% aqueous NaHCO_3 and then with water, dehydrated, and concentrated. Purification of the residue by column chromatography gave the title compound (12.66 g), m.p. 138.2°C, yield 56.3%.

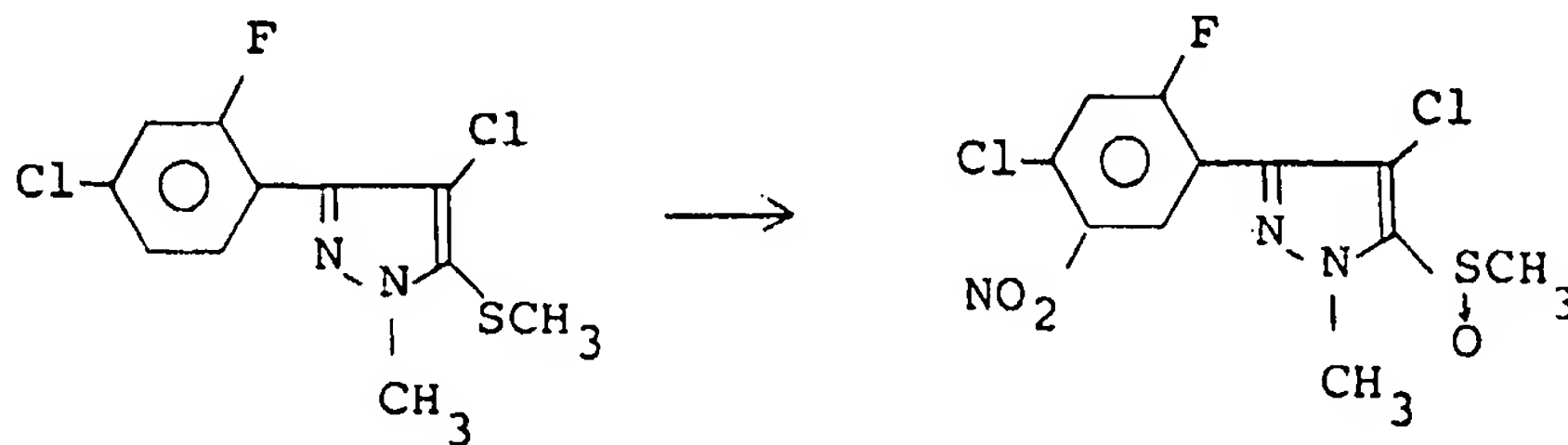
Example 5 Preparation of 4-chloro-3-(4-chloro-2-fluoro-5-nitrophenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (compound No. 404)



A mixture of 60% nitric acid (1.58 g, 0.015 mole) and conc. sulfuric acid (4.90 g, 0.05 mole) was added dropwise to a solution of 4-chloro-3-(4-chloro-2-fluorophenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (3.11 g, 0.01 mole) in 6 ml conc. sulfuric acid at temperatures of -10 to 10°C. Thereafter the reaction mixture was further stirred for 3 hours at room temperature. The reaction product mixture, poured into ice-cold water, was extracted with ethyl acetate. The ethyl acetate layer was washed successively with 5% aqueous NaHCO_3 and water, dehydrated, and evaporated to dryness, giving the title compound (3.0 g), yield 84%.

NMR, δ (ppm): 3.83 (3H, s), 6.67 (1H, t, $J=72$ Hz), 7.36 (1H, d, $J=10$ Hz), 8.20 (1H, d, $J=8$ Hz).

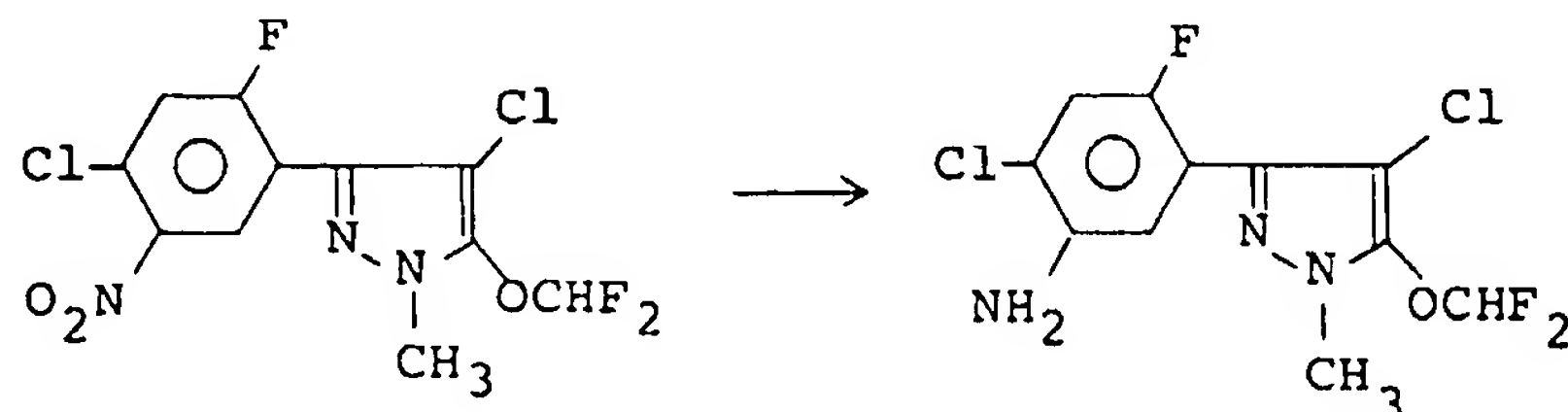
Example 6 Preparation of 4-chloro-3-(4-chloro-2-fluoro-5-nitrophenyl)-1-methyl-5-methylsulfinyl-1H-pyrazole (compound No. 217)



A mixture of conc. sulfuric acid (1.65 g) and conc. nitric acid (sp. gr. 1.38, 8.4 mmol) was added dropwise to a

solution of 4-chloro-3-(4-chloro-2-fluorophenyl)-1-methyl-5-methylthio-1H-pyrazole (0.98 g, 3.4 mmol) in 2 ml conc. sulfuric acid under cooling with ice. After the addition was finished, the reaction mixture was further stirred for 2 hours at room temperature, and allowed to stand overnight. Then the resulting mixture was poured into ice-cold water, and the formed crystals were filtered off, washed successively with water and ether, and dried, giving crystals (1.04 g) of the title compound, m.p. 151.0°C, yield 87.5%.

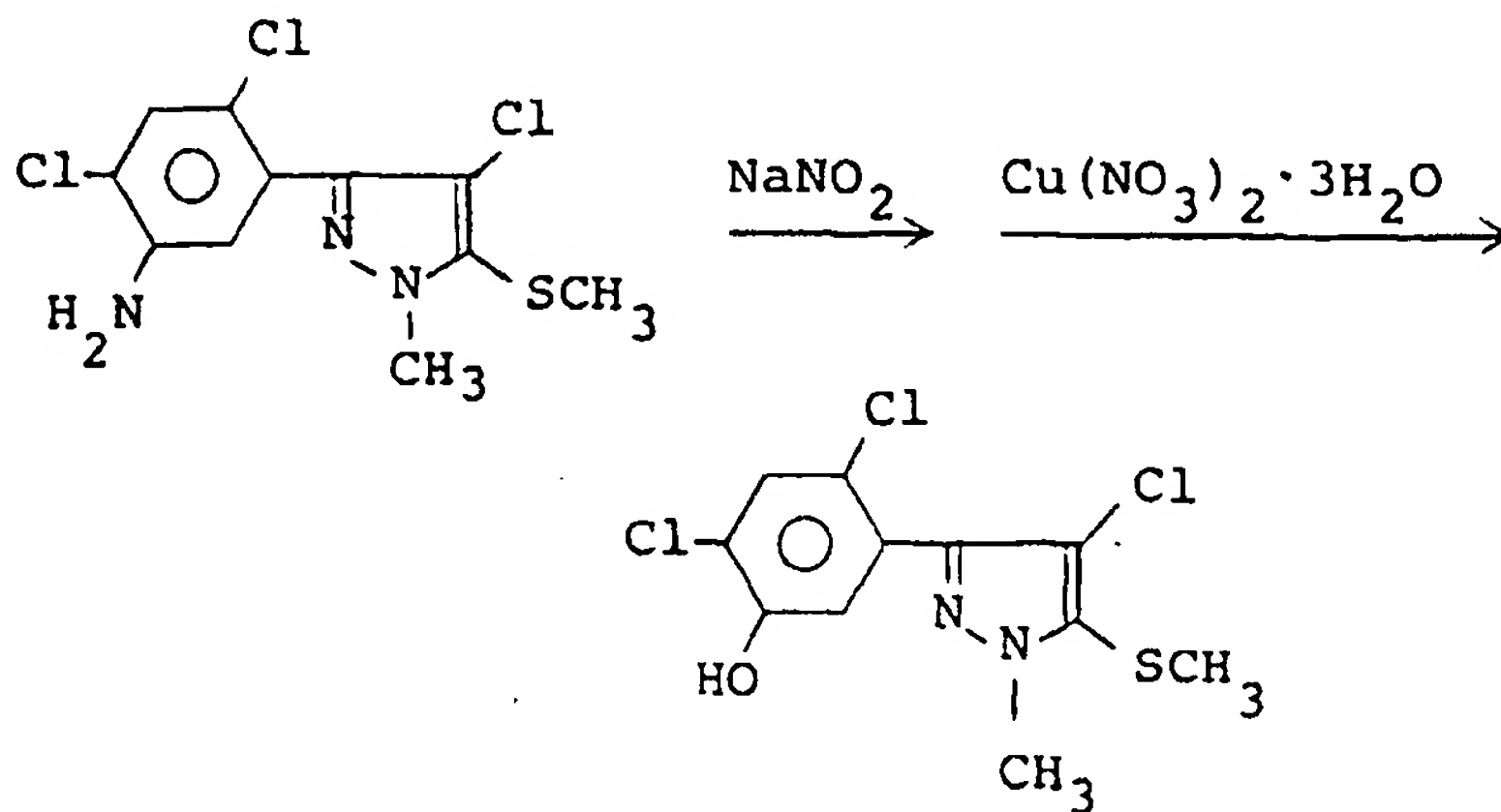
Example 7 Preparation of 3-(5-amino-4-chloro-2-fluorophenyl)-4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazole (compound No. 405)



To 4-chloro-3-(4-chloro-2-fluoro-5-nitrophenyl)-1-methyl-5-methylthio-1H-pyrazole (3.0 g, 8.5 mmol) in a mixture of 15 ml ethanol and 15 ml conc. hydrochloric acid was added stannous chloride (7.70 g, 34 mmol) and the reduction was conducted under reflux for 8 hours. Then the product solution was poured into ice-cold water, the resulting mixture was made alkaline with 20% aqueous NaOH, and the objective product was extracted with ethyl acetate. Dehydration and concentration of the extract solution and purification of the residue by column chromatography gave the title compound (2.15 g), yield 78%.

NMR, δ (ppm): 3.73 (3H, s), 3.85 (2H, s, br), 6.64 (1H, t, $J=72$ Hz), 6.82 (1H, d, $J=8$ Hz), 7.04 (1H, d, $J=10$ Hz).

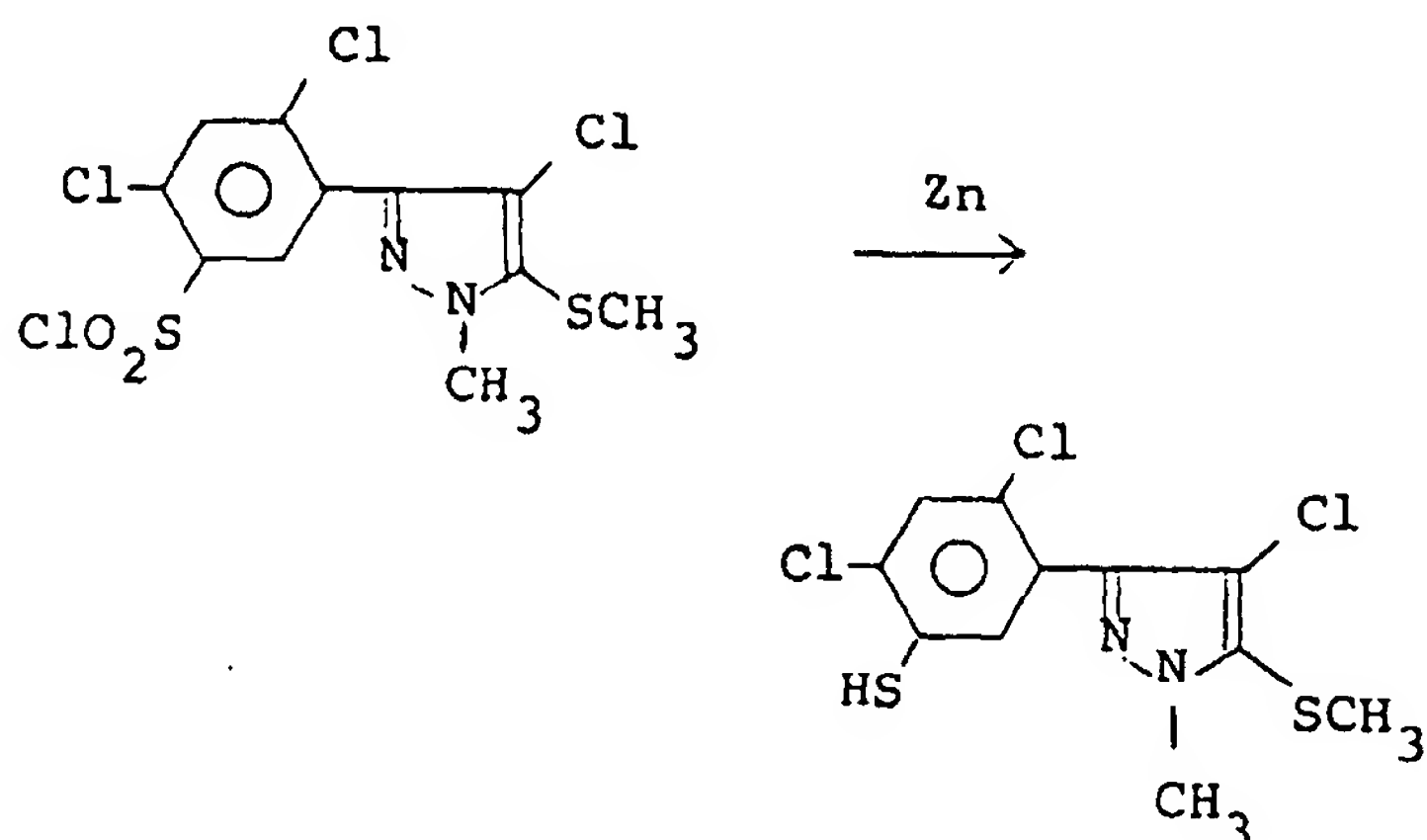
Example 8 Preparation of 4-chloro-3-(2,4-dichloro-5-hydroxyphenyl)-1-methyl-5-methylthio-1H-pyrazole (compound No. 2)



A solution of sodium nitrile (1.33 g) in 4.5 ml water was added dropwise to a 5°C or lower temperature solution of 4-chloro-3-(5-amino-2,4-dichlorophenyl)-1-methyl-5-methylthio-1H-pyrazole (5.67 g, 17 mmol) in 20 ml 50% sulfuric acid while maintaining the reaction temperature between -5 and +5°C. After the addition was finished, the reaction mixture was further stirred for 20 minutes at 0 - 5°C. Then the resulting solution was removed into a 1-ℓ Erlenmeyer flask, and a 40% aqueous copper nitrate solution (about 100 g of a total of 318.8 g) was first added dropwise to the former solution kept at temperatures of 0 - 10°C, and secondly the remainder copper nitrate solution was added dropwise. After stirring of the whole mixture for 30 minutes at room temperature, cuprous oxide (2.51 g, 17 mmol) was added in limited amounts, conducting the reaction for 1 hour at room temperature. Then the resulting solution was poured into ice-cold water, and the objective product was extracted with ethyl acetate. The extract solution was washed with water and concentrated. The residue was purified by recrystallization from methylene chloride, giving crystals

(2.61 g) of the title compound, m.p. 163.2°C, yield 46.4%.

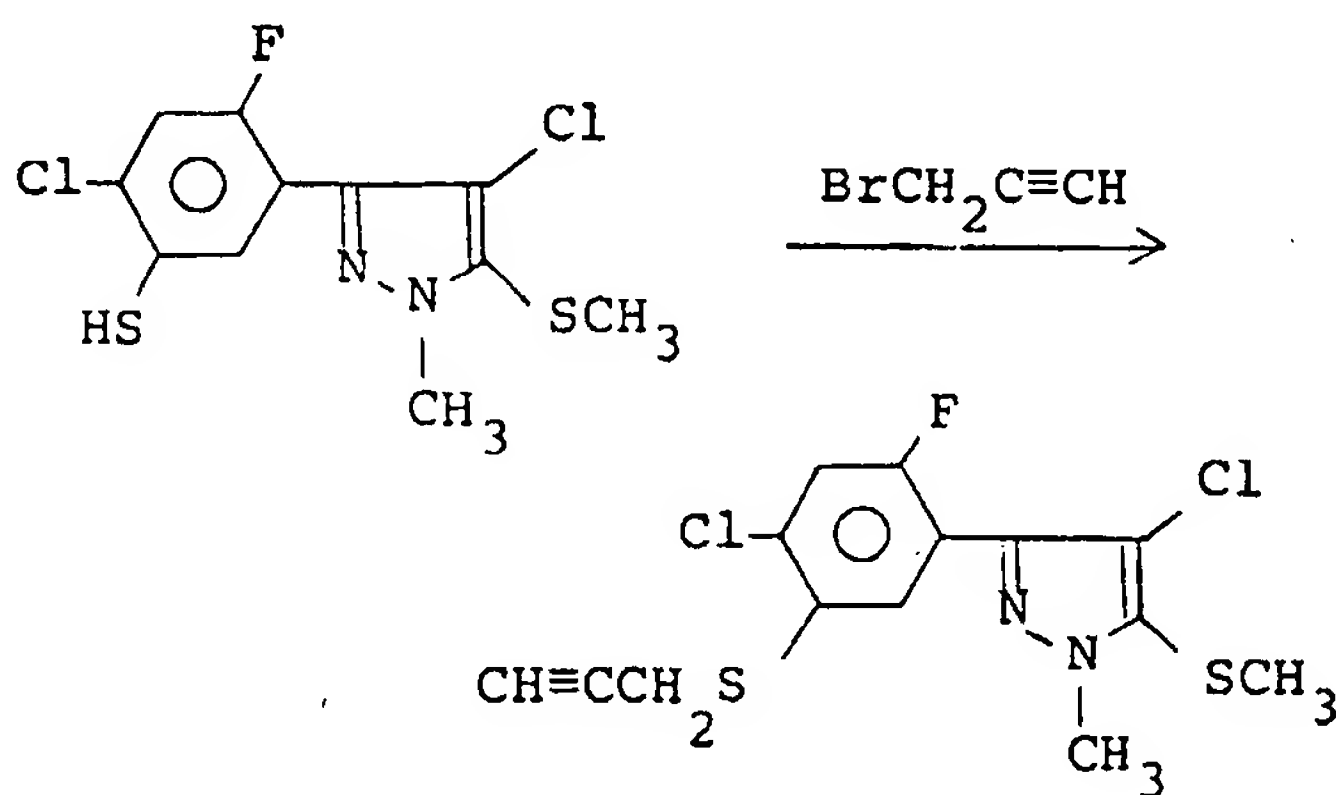
Example 9 Preparation of 4-chloro-3-(2,4-dichloro-5-mercaptophenyl)-1-methyl-5-methylthio-1H-pyrazole (compound No. 15)



A mixture of 4-chloro-3-(5-chlorosulfonyl-2,4-dichlorophenyl)-1-methyl-5-methylthio-1H-pyrazole (7.71 g, 19 mmol), 80 ml glacial acetic acid, and zinc dust (24.8 g, 380 mmol) was subjected to reaction under reflux for 3.5 hours. The resulting solution was poured into ice-cold water, and the objective product was extracted with ethyl acetate. The extract solution was washed with water, dehydrated, and concentrated, giving the title compound (5.75 g) as oily matter, yield 89.1 %.

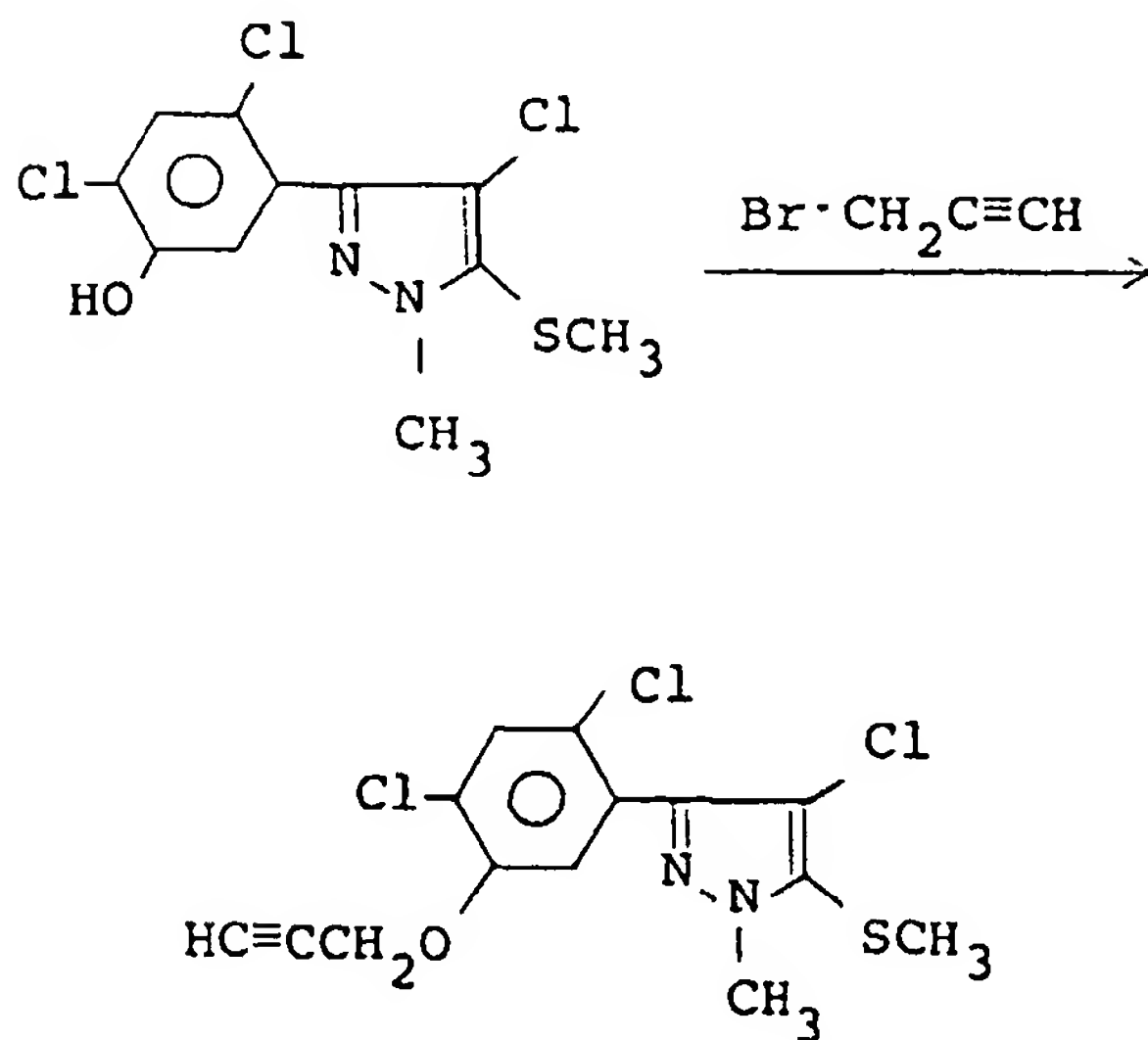
n_D (24.3°C): 1.6303

Example 10 Preparation of 4-chloro-3-[4-chloro-2-fluoro-5-(2-propynylthio)phenyl]-1-methyl-5-methylthio-1H-pyrazole (compound No. 221)



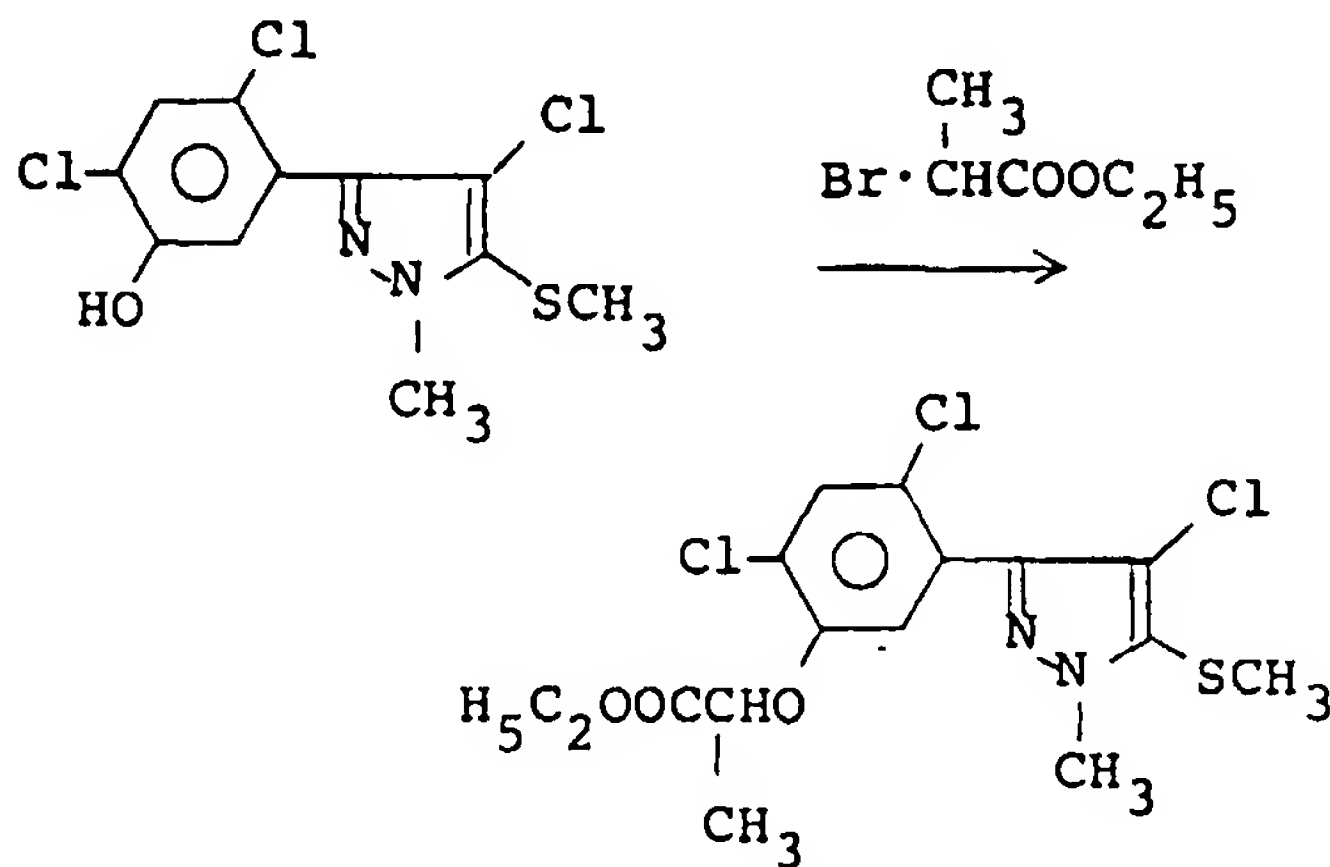
A mixture of 4-chloro-3-(4-chloro-2-fluoro-5-mercaptophenyl)-1-methyl-5-methylthio-1H-pyrazole (0.60 g, 1.86 mmol), 30 ml acetone, a K_2CO_3 powder (2.04 mmol), and propargyl bromide (2.23 mmol), was subjected to reaction under reflux for 2 hours. Then the resulting solution was filtered to remove acetone-insoluble matter and the filtrate was concentrated and purified by silica gel column chromatography, giving crystals (0.54 g) of the title compound, m.p. 93 - 96°C, yield 80.5%.

Example 11 Preparation of 4-chloro-3-[2,4-dichloro-5-(2-propynyloxy)phenyl]-1-methyl-5-methylthio-1H-pyrazole (compound No. 5)



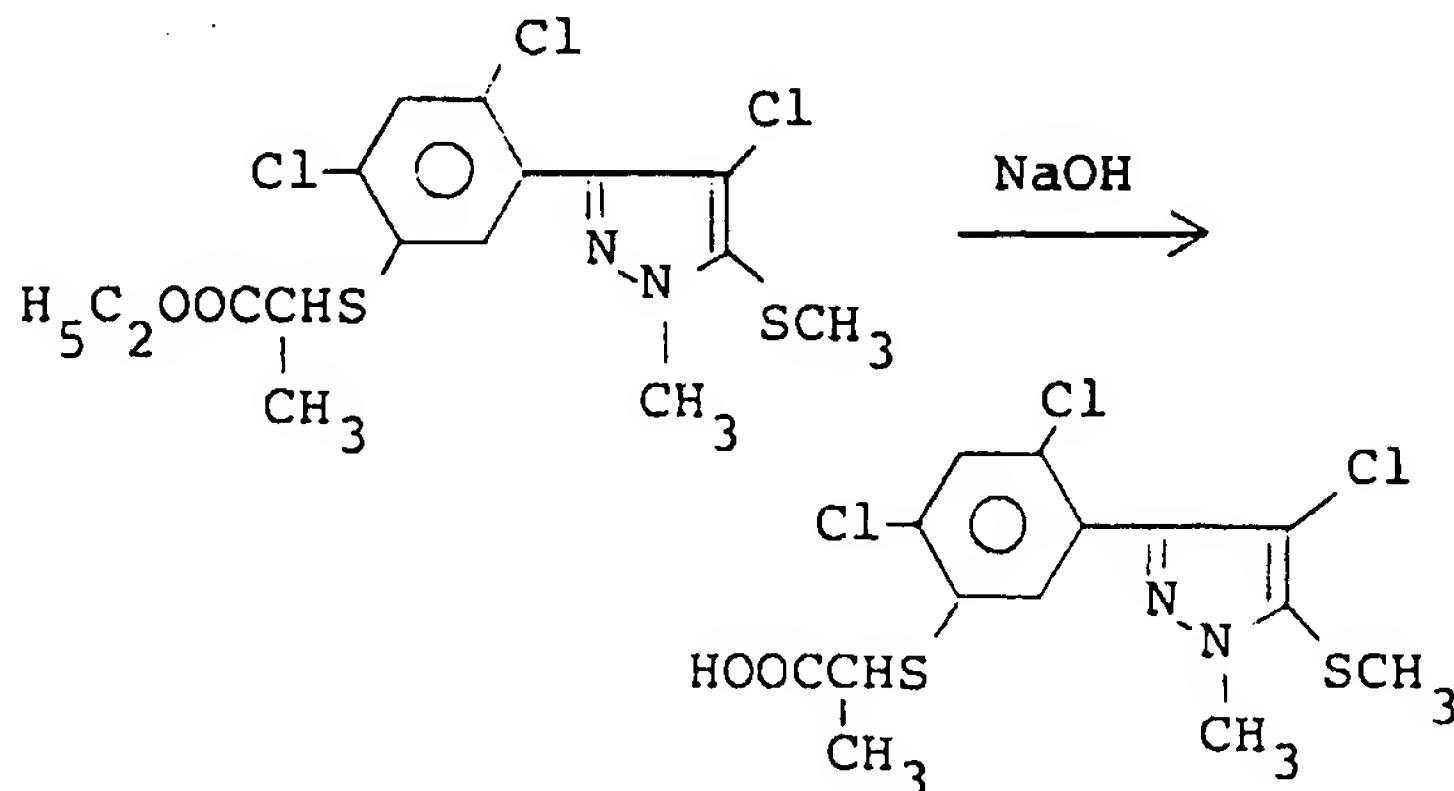
A mixture of 4-chloro-3-(2,4-dichloro-5-hydroxyphenyl)-1-methyl-5-methylthio-1H-pyrazole (2.00 g, 6.2 mmoles), 20 ml acetone, anhydrous K_2CO_3 (1.28 g, 9.3 mmoles), and propargyl bromide (1.10 g, 9.3 mmoles) was subjected to reaction under reflux for 2 hours. Then the resulting solution was filtered to remove acetone-insoluble matter. The filtrate was concentrated and purified by silica gel column chromatography, giving crystals (1.89 g) of the title compound, m.p. 71.5 - 72.5°C, yield 84.3%.

Example 12 Preparation of ethyl 2-[5-(4-chloro-1-methyl-5-methylthio-1H-pyrazol-3-yl)-2,4-dichloro-phenoxy] propionate (compound No. 11)



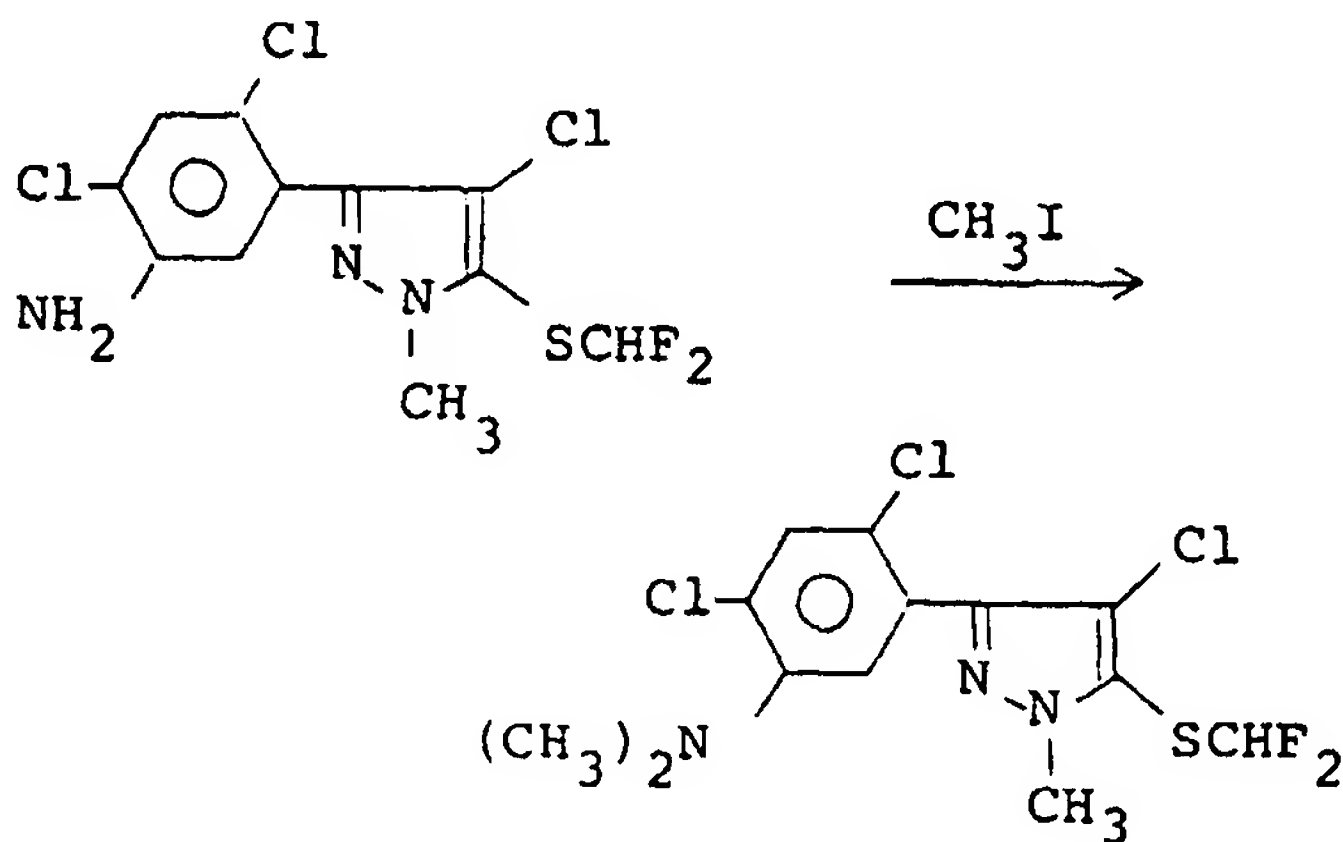
A mixture of 4-chloro-3-(2,4-dichloro-5-hydroxyphenyl)-1-methyl-5-methylthio-1H-pyrazole (3.50 g, 10.8 mmoles), 50 ml acetone, anhydrous K_2CO_3 , and ethyl 2-bromopropionate (2.06 g, 11.4 mmoles) was subjected to reaction under reflux for 2 hours. Then the resulting solution was filtered to remove acetone-insoluble matter, and the filtrate was concentrated and purified by silica gel column chromatography, giving the title compounds as oily matter, yield 70.0%. $n_D(28.8^\circ C): 1.5763$

Example 13 Preparation of 2-[5-(4-chloro-1-methyl-5-methylthio-1H-pyrazol-3-yl)-2,4-dichlorophenylthio]propionic acid (compound No. 25)



After addition of NaOH (0.31 g, in pellet form, 95% content) to a solution of ethyl 2-[5-(4-chloro-1-methyl-5-methylthio-1H-pyrazol-3-yl)-2,4-dichlorophenylthio]propionate (2.28 g, 5.18 mmol) in 50 ml ethanol, several drops of water added, thereby conducting the reaction for 3 hours at room temperature. Then the solvent was removed by evaporation of the resulting solution under reduced pressure. Water and ethyl acetate were added to the residue and the formed aqueous layer was separated. The aqueous layer was acidified with hydrochloric acid. The objective product was extracted from the aqueous layer with ethyl acetate. The extract solution was washed with water, dehydrated, and evaporated to dryness, giving crystals (1.74 g) of the title compound, m.p. 180.0°C, yield 81.6%.

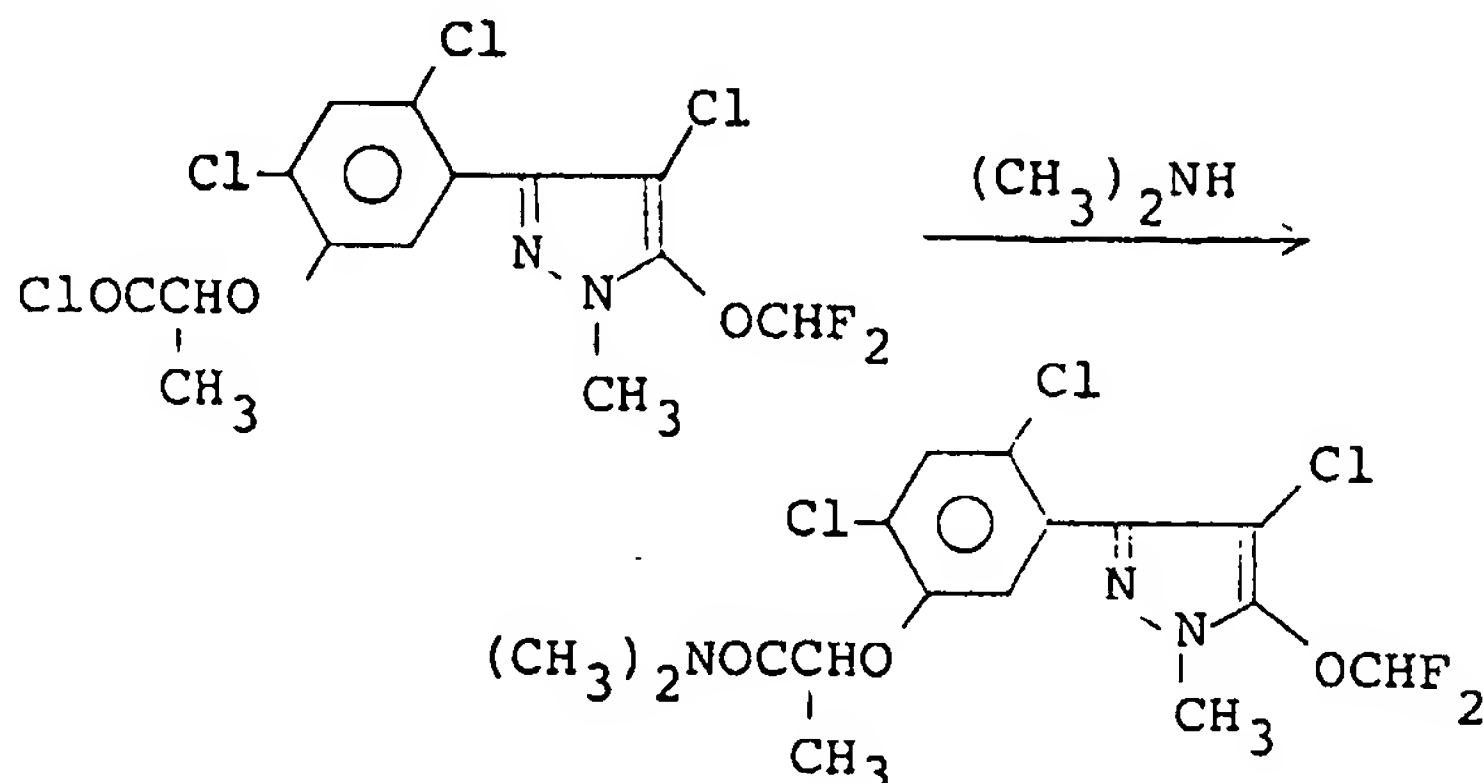
Example 14 Preparation of 4-chloro-3-(2,4-dichloro-5-dimethylaminophenyl)-5-difluoromethylthio-1-methyl-1H-pyrazole (compound No. 207)



A mixture of 3-(5-amino-2,4-dichlorophenyl)-4-chloro-5-difluoromethylthio-1-methyl-1H-pyrazole (0.72 g, 2 mmol), 15 ml sulforane, NaHCO₃ (0.18 g, 2.2 mmol), and methyl iodide (0.34 g, 2.4 mmol) was subjected to reaction at 80°C for 16 hours. Then the resulting mixture was poured into water and the objective product was extracted with ether. The extract solution was washed with water, dehydrated, and evaporated to dryness, giving the title compound (0.62 g) in paste form, yield 80.2%.

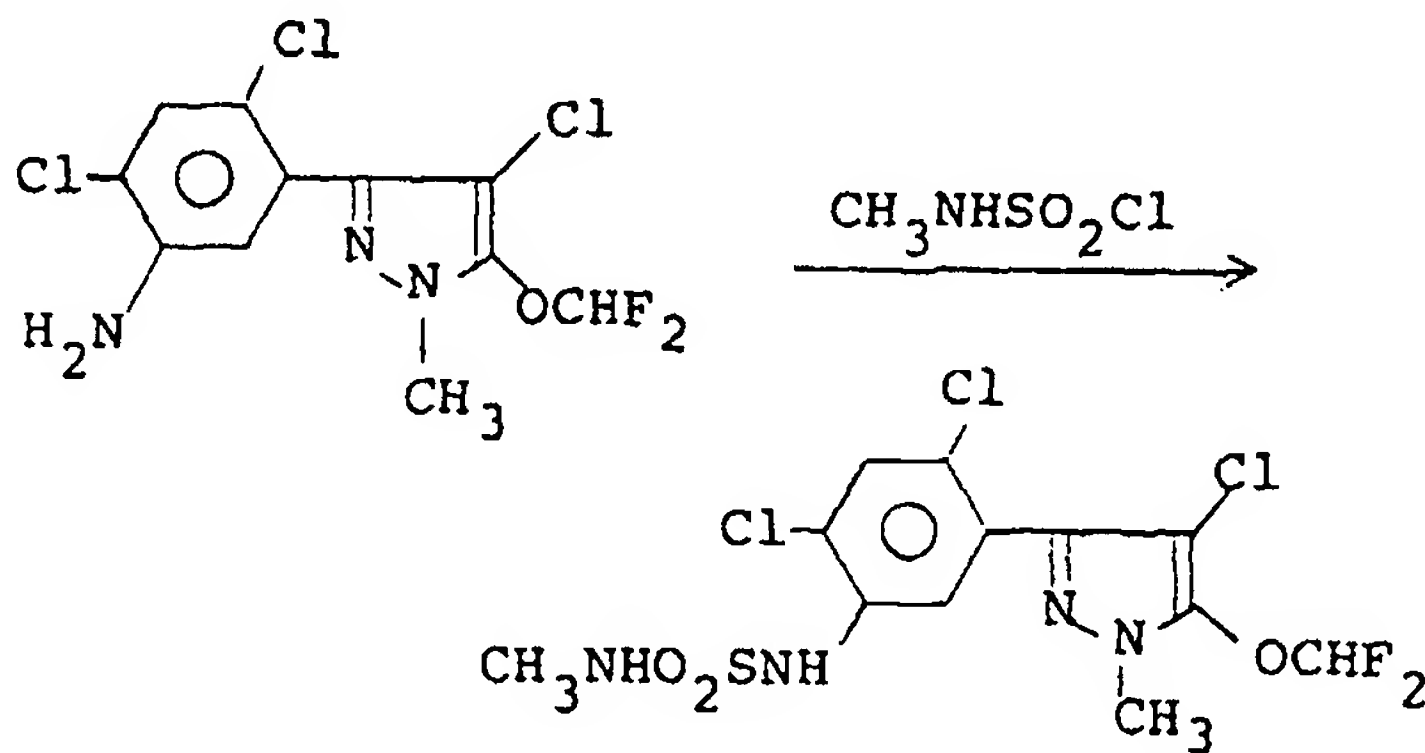
n_D (21.6°C): 1.5838

Example 15 Preparation of N,N-dimethyl 2-[5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-2,4-dichlorophenoxy]propionamide (compound No. 110)



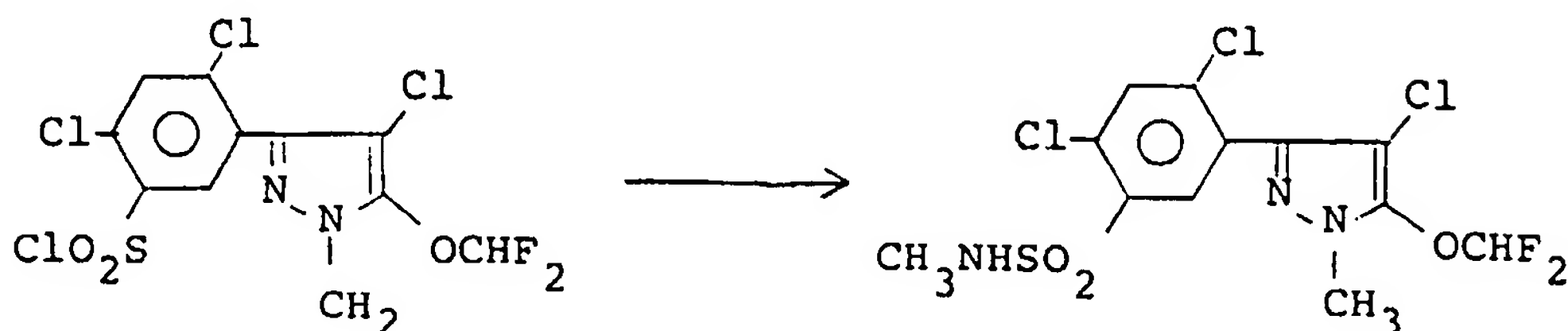
A solution of 2-[5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-2,4-dichlorophenoxy]propionyl chloride (0.80 g, 1.78 mmol) in 20 ml anhydrous tetrahydrofuran was added dropwise to a solution of 50% dimethylamine (0.32 g, 3.56 mmol) in 20 ml tetrahydrofuran at room temperature to react the pyrazole derivative with dimethylamine. After the addition was finished, the reaction was continued further for 1 hour. Then ethyl acetate was added to the reaction product solution, and the mixture was washed with water. The separated organic layer was dehydrated, concentrated under reduced pressure, and purified by silica gel chromatography, giving crystals (0.24 g) of the title compound, m.p. 148.9°C, yield 27%.

Example 16 Preparation of 4-chloro-3-(2,4-dichloro-5-methylaminosulfonylaminophenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (compound No. 149)



N-Methylaminosulfonyl chloride (0.22 g, 1.67 mmol) was added dropwise to a solution of 3-(5-amino-2,4-dichlorophenyl)-4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazole (0.44 g, 1.28 mmol) and triethylamine (0.17 g, 1.67 mmol) in 20 ml anhydrous tetrahydrofuran under cooling with ice to react with the pyrazole derivative. After the addition was finished, the reaction was continued further for 2 hours. Then ethyl acetate was added to the reaction product solution, and the mixture was washed with water. The separated organic layer was dehydrated, concentrated, and purified by recrystallization from a n-hexane-ether mixture, giving the title compound (0.40 g), m.p. 133.0°C, yield 71.7%.

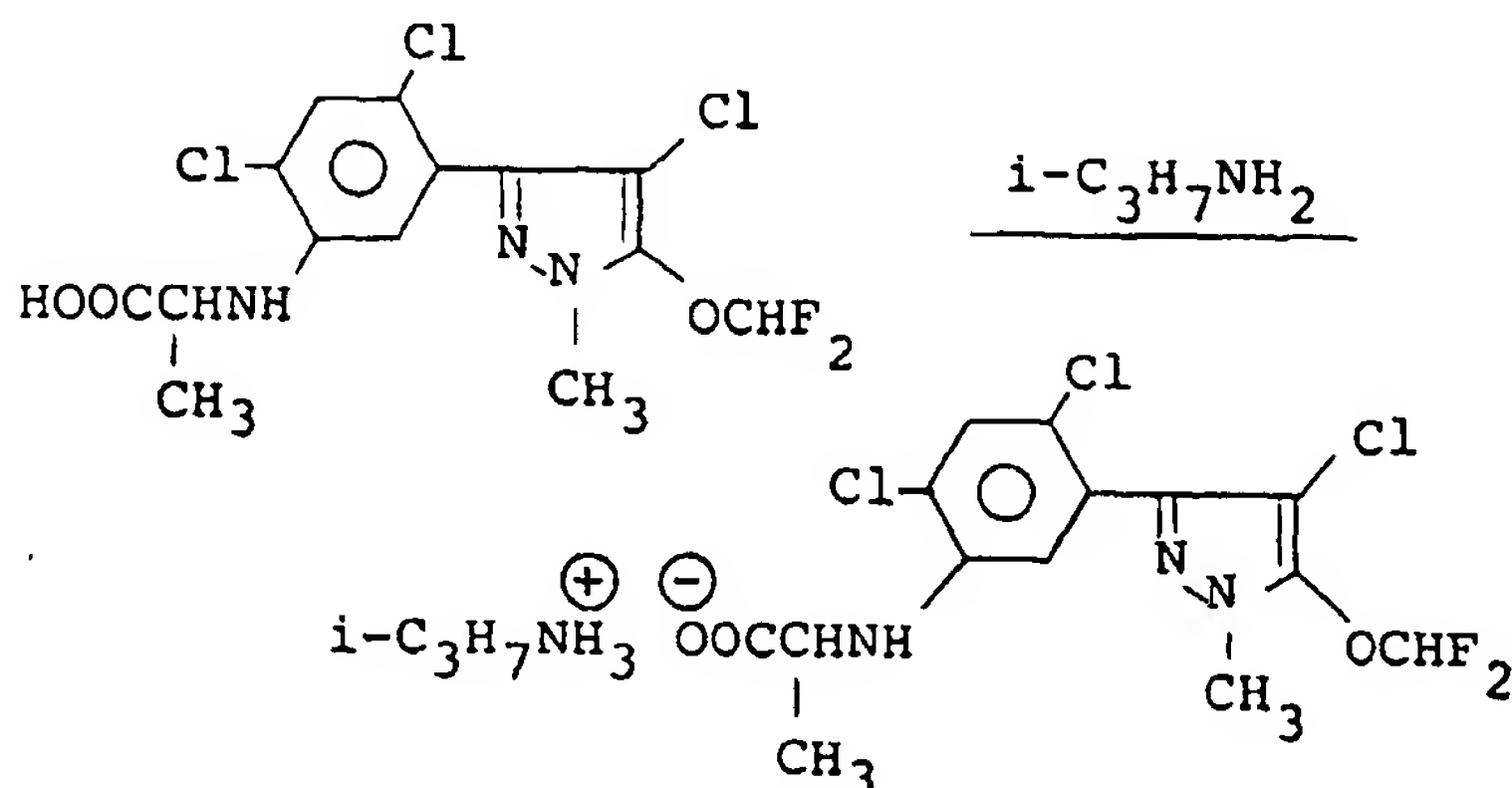
Example 17 Preparation of 5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-N-methyl-2,4-dichlorobenzenesulfonamide (compound No. 164)



A solution of 4-chloro-3-(5-chlorosulfonyl-2,4-dichlorophenyl)-5-difluoromethoxy-1-methyl-1H-pyrazole (0.43 g, 1 mmole) in 2 ml tetrahydrofuran was added dropwise to a 40% aqueous methylamine solution dissolved in 20 ml tetrahydrofuran to react the pyrazole derivative with methylamine. After the addition was finished, the reaction was continued further for 30 minutes at room temperature. Then ethyl acetate was added to the reaction product solution to extract the objective product. The extract solution was washed with water, dehydrated, and evaporated to dryness, giving the title compound (0.42 g) in paste form, yield 100%.

$n_D(17.9^\circ\text{C}): 1.5461$

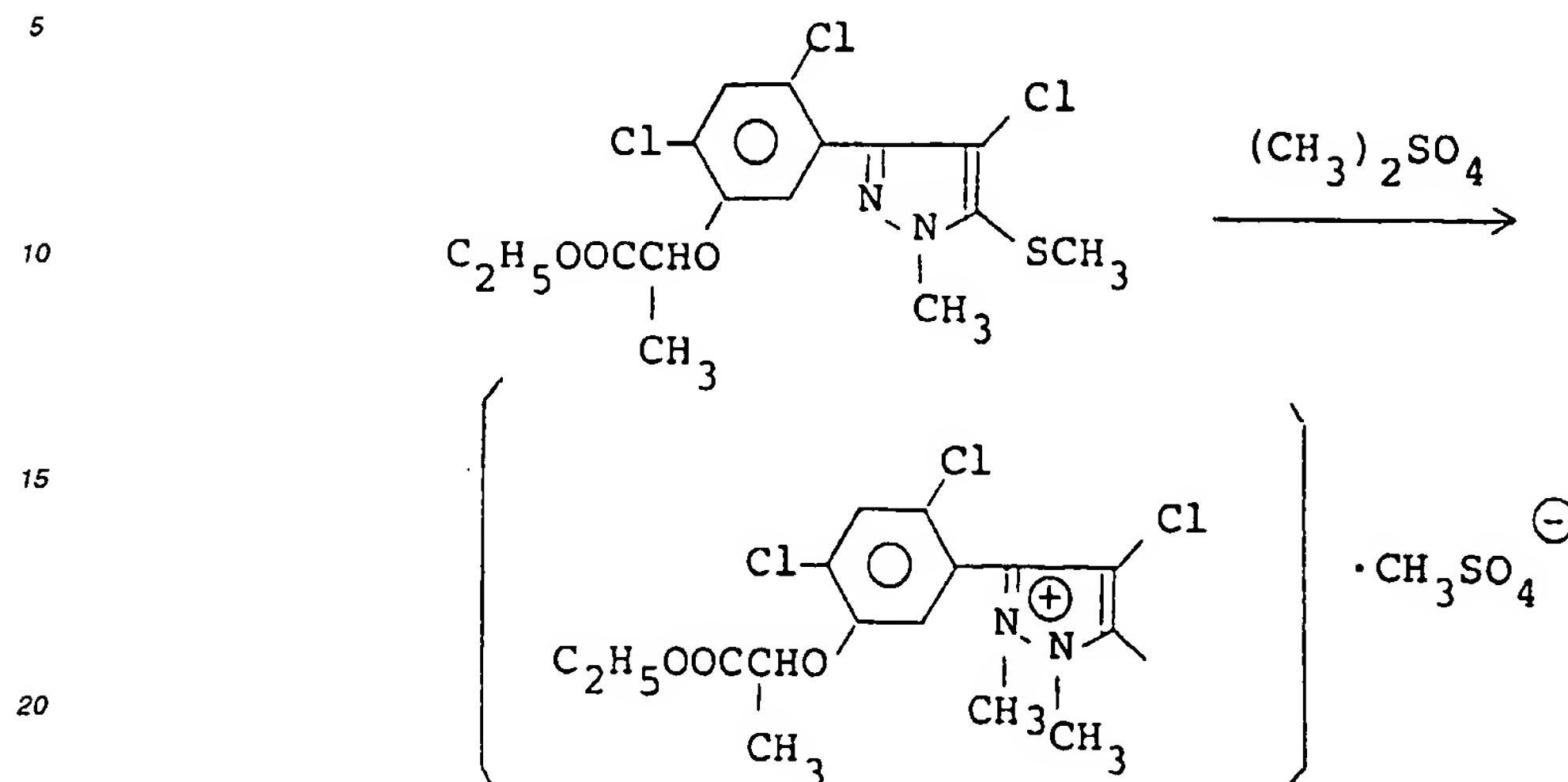
Example 18 Preparation of 2-[5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-2,4-dichloroanilino] isopropylammonium propionate (compound No. 160)



Isopropylamine (0.05 g, 0.88 mmole) was added to a solution of 2-[5-(4-chloro-5-difluoromethoxy-1-methyl-1H-pyrazol-3-yl)-2,4-dichloroanilino]propionic acid (0.33 g, 0.8 mmole) in 20 ml tetrahydrofuran, thereby being reacted with the acid for 30 minutes at room temperature. Thereafter the solvent was removed from the reaction product solution by evaporation under reduced pressure, giving the title compound quantitatively (100% yield) as oily matter.

NMR δ (ppm); 1.12 (6H, d), 1.48 (3H, d), 2.60 - 3.30 (1H, m), 3.73 (3H, s), 3.91 (1H, q), 6.55 (1H, s), 6.68 (1H, s), 7.73 (1H, s), 7.98 (3H, s).

Example 19 Preparation of ethyl 2-[5-(4-chloro-1,2-dimethyl-5-methylthio-1H-pyrazolium-3-yl)-2,4-dichlorophenoxy] propionatemethyl sulfate (compound No. 215)



Dimethyl sulfate (0.13 g, 1.04 mmoles) was added to a solution of ethyl 2-[5-(4-chloro-1-methyl-5-methylthio-1H-pyrazol-3-yl)-2,4-dichlorophenoxy]propionate (0.22 g, 0.52 mmole) in 20 ml benzene, thereby being reacted with the pyrazole derivative under reflux for 5 days. Thereafter the solvent was removed from the reaction product solution by evaporation under reduced pressure. The residue was dissolved in ethyl acetate, the solution was washed with a small amount of water, and the organic layer was dehydrated and evaporated to dryness, giving the title quaternary salt in paste form, yield 52.6%.

nD (25.7°C): 1.5548

The present inventive 3-(substituted phenyl)-pyrazole derivatives represented by formula (I) and salts thereof are capable of controlling annual and perennial weeds grown in paddy fields, upland fields, orchards, and swamps, such as barnyardgrass (*Echinochloa Crus-galli* Beauv., an annual gramineous grass which is a typical weed strongly injurious, grown in paddy fields), Mizugayatsuri (*Cyperus serotinus* Rottb., a perennial weed of Cyperaceae family, grown in swamps, ditches and paddy fields), Urikawa (*Sagittaria pygmaea* Miq., an injurious perennial weed of Alismataceae family, grown in swamps, ditches, and paddy fields), Hotarui (*Scirpus juncoides* Roxb. subsp. *juncoides*, a perennial cyperaceous weed grown in swamps, water ways, and paddy fields), wild oats (*Avena fatua* L., an annual gramineous grass grown in plains, highlands, and upland fields), large crabgrass (*Digitaria adscendens* Henr., an annual gramineous grass which is a typical strongly injurious weed grown in upland fields and orchards), Curlydock (*Rumex japonicus* Houtt., a perennial polygonaceous weed grown in upland fields and on roadsides), umbrella sedge (*Cyperus Iria* L., an annual cyperaceous weed grown in upland fields and on roadsides), Redroot pigweed (*Amaranthus retroflexus* L., an annual weed of Amaranthaceae family grown in upland fields, roadsides, and vacant lands), Cleavers (*Galium aparine* L., a strongly injurious annual weed of Rubiaceae family grown in upland fields), Bridseye Speedwell (*Veronica persica* L., a strongly injurious weed of Scrophulariaceae family grown in upland fields and orchards), Scented mayweed (*Matricaria chamomilla* L., injurious composite weed grown in upland fields), velvetleaf (*Abutilon theophrasti* L., a strongly injurious weed of Malvaceae family grown in upland fields), cocklebur (*Xanthium strumarium* L., a strongly injurious annual composite weed grown in upland fields), and tall morningglory (*Ipomoea purpurea* Voigt, a strongly injurious weed of Convolvulaceae family grown in upland fields).

Since the 3-(substituted phenyl)pyrazole derivatives of formula (I) and salts thereof exhibit excellent controlling effect on weeds before or immediately after germination, characteristic physiological activities of these compounds can be manifested by treating fields with the derivative or the salt before or after planting of useful plants therein (including fields where useful plants are already planted) in the stage prior to weed emergence or in the period from the initial stage to the middle stage of weed growth. However, the applications of the present inventive herbicides are not limited such forms as stated above. The present herbicides can be applied to control not only weeds in paddy fields or upland fields but also general weeds grown in other places, for example, reaped fields, temporarily non-cultivated paddy fields and upland fields, ridges between paddy fields, agricultural pathways, waterways, lands constructed for pasture, graveyards, parks, roads, playgrounds, unoccupied areas around buildings, reclaimed lands, railways, and forests. It is most desirable in economy as well to treat these areas with the present herbicides, before the end of the

initial stage of weed growth, but the treatment is not limited to this but can be carried out in the middle stage of weed growth.

The 3-(substituted phenyl)pyrazole derivative of formula (I) or salt thereof, when applied as a herbicide, is generally made up, according to the customary procedure for preparing agricultural chemicals, into a form convenient to use. That is, the pyrazole derivative or salt thereof is blended with a suitable inert carrier and if necessary, further with an adjuvant, in proper ratios, and the mixture is made up into a suitable form of preparation, e.g. a suspension, emulsifiable concentrate, solution, wettable powder, granules, dust, or tablets, through dissolution, dispersion, suspension, mixing, impregnation, adsorption, or sticking. In the present invention, either solid or liquid inert carriers may be used. Suitable materials as solid carriers include soybean flour, cereal flour, wood flour, bark flour, saw dust, powdered tobacco stalks, powdered walnut shells, bran, powdered cellulose, extraction residues of vegetables, powdered synthetic polymers or resins; clays (e.g. kaolin, bentonite, and acid clay), talcs (e.g. talc and pyrophyllite), silica powders or flakes (e.g. diatomaceous earth, silica sand, mica and white carbon, i.e. highly dispersed silicic acid, also called finely divided hydrated silica or hydrated silicic acid), activated carbon, powdered sulfur, powdered pumice, calcined diatomaceous earth, ground brick, fly ash, sand, calcium carbonate powder, calcium phosphate powder, and other inorganic or mineral powders, chemical fertilizers (e.g. ammonium sulfate, ammonium nitrate, urea, and ammonium chloride) and compost. These materials may be used alone or in combination.

Suitable materials as liquid carriers include liquids which themselves have solvent power as well as liquids which do not have solvent power but can disperse active ingredients with the aid of adjuvants. Typical examples of such liquids, which may be used alone or in combination, are water, alcohols (including methanol, ethanol, isopropanol, butanol, and ethylene glycol), ketones (including acetone, methyl ethyl ketone, methyl isobutyl ketone, di-isobutyl ketone, and cyclohexanone), ethers (including ethyl ether, dioxane, Cellosolve, dipropyl ether, and tetrahydrofuran), aliphatic hydrocarbons (including gasoline, and mineral oils), aromatic hydrocarbons (including benzene, toluene, xylene, solvent naphtha, and alkylnaphthalenes), halogenated hydrocarbons (including dichloroethane, chloroform and carbon tetrachloride), esters (including ethyl acetate, diisopropyl phthalate, dibutyl phthalate, and dioctyl phthalate), amides (including dimethylformamide, diethylformamide, and dimethylacetamide), nitriles (including acetonitrile), and dimethylsulfoxide.

The following materials are cited as typical examples of the adjuvants which may be used alone or in combination or may not be used at all.

For the purpose of emulsifying, dispersing, solubilizing, and/or wetting active ingredients, there may be used surface active agents such as polyoxyethylene alkyl ethers, polyoxyethylene alkylaryl ethers, polyoxyethylene higher fatty acid esters, polyoxyethylene resins, polyoxyethylene sorbitan monoaurate, polyoxyethylene sorbitan monooleate, alkylarylsulfonates, naphthalenesulfonic acid condensation products, ligninsulfonates, and higher alcohol sulfate esters.

For the purpose of stabilizing the dispersion of active ingredients, tackifying and/or binding them, there may be used adjuvants, for example, casein, gelatin, starch, methylcellulose, carboxymethylcellulose, gum arabic, polyvinyl alcohol, turpentine, bran oil, bentonite, and ligninsulfonates.

For the purpose of improving flow properties of solid herbicidal products there may be used adjuvants, for example, waxes, stearates, and alkyl phosphates.

Adjuvants, e.g. naphthalenesulfonic acid condensation products and polyphates, can be used as peptizers in dispersible herbicidal products.

Adjuvants, e.g. silicone oils, can also be used as defoaming agents.

The content of active ingredients may be varied as occasion demands; the suitable contents are from 0.01 to 50% by weight for preparing, for example, dusts or granules as well as emulsifiable concentrates or wettable powders.

For destroying various weeds or inhibiting their growth, the herbicidal composition containing the 3-(substituted phenyl)pyrazole derivative of formula (I) or salt thereof as an active ingredient is applied as such or after being properly diluted with or suspended in water or other media, in amounts effective for destroying weeds or inhibiting their growth, to the foliage and stalks of the weeds or to soil in the area where the emergence or growth of weeds is undesirable.

The amount of the present herbicidal composition to be used, depending upon various factors, e.g. the purpose of application, object weeds, emergence or growth states of weeds and crops, emergence tendency of weeds, weather, environmental conditions, form of the herbicidal composition, mode of the application, type or state of the application locus, and time of the application, is chosen properly according to the purpose from the range of 1.0 g to 10 Kg, in terms of the amount of active ingredient, per hectare.

When the present herbicidal composition is applied to paddy fields or upland fields, it is desirable to choose such low doses as not to injure crops but to destroy weeds or control their growth. When the composition is applied to non-farming areas, suitable doses of active ingredient for destroying the weeds are chosen from amounts of 100 g/hectare and more.

The present herbicidal composition can be applied jointly with other herbicides for the purpose of expanding both the range of controllable weed species and the period of time when effective applications are possible or for the purpose

of reducing the dosage.

The following examples illustrate herbicidal effects and formulations of the present inventive herbicidal composition without limiting the scope of the invention.

5 Test Example 1 Herbicidal effect on paddy field weeds of pre-emergence stage

Pots (1/10,000-are) were filled with soil to simulate a paddy field, then plated separately with seeds of barn-yardgrass, konagi (*Monochoria vaginalis* (Burm.f.) Presl var *plantaginae*), and hotarui which are all injurious weeds grown in paddy fields, and were conditioned so that the seeds might become in pre-emergence stage.

10 Soil in each pot was sprayed with each of solutions containing compounds (listed in Table 1) of the present invention as active ingredients at a predetermined concentration. 21 Days later, the herbicidal effect was examined, the percentages of killed weeds were calculated in comparison with those on the untreated soil, and the herbicidal activity was judged and the chemical injury of a rice plant was examined at the same time and judged according to the following criterion:

15 ° Herbicidal activity

Rating	Percentage of killed weed
5	95% or more
4	70 - 95% (exclusive)
3	50 - 70% (exclusive)
2	30 - 50% (exclusive)
1	10 - 30% (exclusive)
0	less than 10%

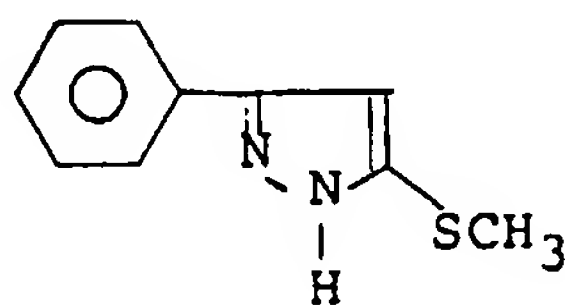
° Phytotoxicity

Rating	Degree of phytotoxicity
0	No phytotoxicity
1	Browning occurs but disappears in the initial growth stage, growth inhibition is not observed.
2	Browning and distinct growth inhibition are observed but normal conditions are soon restored.
3	Browning and growth inhibition are remarkable and the restoration is slow.
4	Browning and growth inhibition are remarkable and some of the rice plants are killed.
5	All the rice plants were killed.

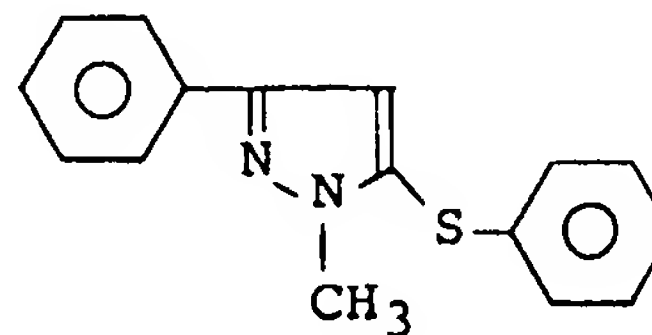
For comparison the following compounds were also tested.

40 Compound A: 3-phenyl-5-methylthiopyrazole, described on page 3 in Japanese Patent Application Kokai No. Sho. 51-91861; compound B: described in Example 1 on page 4 in the same patent application; compound C: compound No. 8 described in Japanese Patent Application Kokai No. Sho. 54-70270; compound D: compound No. 159 described on page 9 in Japanese Patent Application Kokai No. Sho. 55-9062.

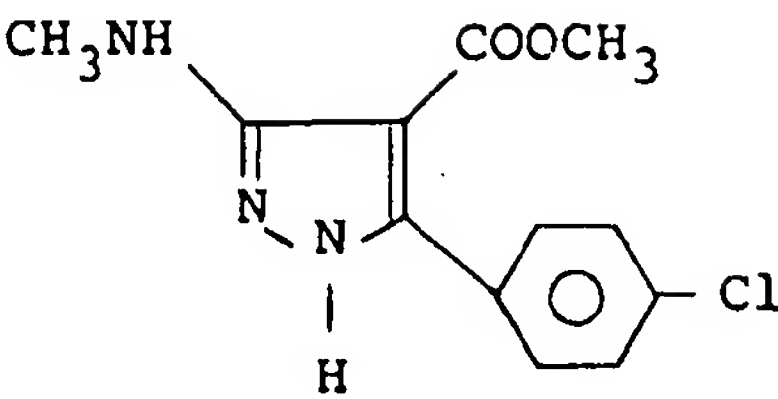
45 Compound A.



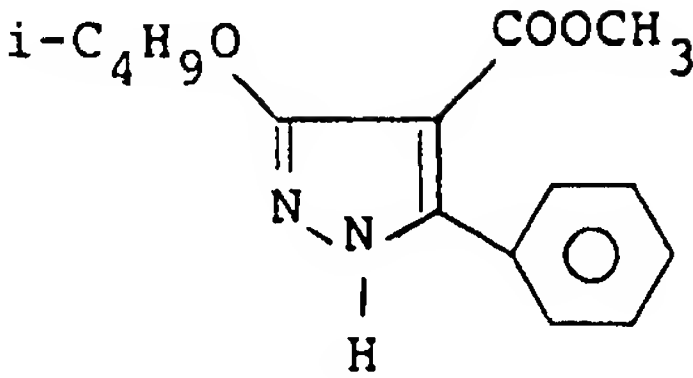
Compound B.



Compound C.



Compound D.



Results of the test are shown in Table 5.

Table 5

Comp'd No.	Amount of active ingredient Kg/ha	Phytotoxicity	Pre-emergence treatment		
		rice	Barnyardgrass	Hatarui	Konagi
86	0.8	5	5	5	5
	0.1	4	5	5	5
87	0.8	5	5	5	5
	0.1	4	5	5	5
117	0.8	4	5	5	5
	0.1	3	5	5	5
119	0.8	4	5	5	5
	0.1	2	5	5	5
152	0.8	3	5	5	5
	0.1	2	5	5	5
153	0.8	5	5	5	5
	0.1	3	5	5	5
253	0.8	5	5	5	5
	0.1	1	3	5	5
256	0.8	5	5	5	5
	0.1	4	5	5	5
257	0.8	5	5	5	5
	0.1	3	5	5	5
258	0.8	5	5	5	5
	0.1	4	5	5	5
259	0.8	5	4	5	5
	0.1	3	2	4	4
260	0.8	5	5	5	5
	0.1	3	2	5	5
268	0.8	5	5	5	5
	0.1	3	5	5	5
269	0.8	5	5	5	5
	0.1	3	2	5	5
275	0.8	5	5	5	5
	0.1	2	3	5	5
344	0.8	5	5	5	5
	0.1	4	5	5	5
346	0.8	5	5	5	5
	0.1	4	5	5	5
359	0.8	5	5	5	5
	0.1	3	5	5	5

Table 5 (continued)

Comp'd No.	Amount of active ingredient Kg/ha	Phytotoxicity	Pre-emergence treatment		
		rice	Barnyardgrass	Hatarui	Konagi
368	0.8	5	5	5	5
	0.1	2	5	5	5
369	0.8	5	5	5	5
	0.1	3	5	5	5
370	0.8	5	5	5	5
	0.1	2	2	5	5
371	0.8	5	5	5	5
	0.1	2	1	5	5
A	5	0	0	0	0
B	5	0	0	2	0
C	5	0	0	0	0
D	5	0	0	0	0

As shown in table 5, the present 3-(substituted phenyl) pyrazole derivatives or salt thereof, represented by the general formula (I), exhibit more excellent controlling effects of weeds than comparative compound A, B, C or D at pre-emergence treatment in paddy fields. Even when the derivatives exhibit both herbicidal activity on weeds and phytotoxicity on crops at the same dosage, however, by the selection of a proper lower dosage, sufficient herbicidal activity is retained but phytotoxicity on crops is reduced remarkably.

Test Example 2 Herbicidal effect on paddy field weeds of post-emergence stage

Pots (1/10,000-are) were filled with soil to simulate a paddy field, then planted separately with seeds of barnyardgrass and hotarui and with tubers of mizugayatsuri and urikawa, which are all injurious weeds grown in paddy fields, and were conditioned so that these weeds might grow to 1-leaf stage.

Soil in each pot was sprayed with each of solutions containing compounds (listed in Table 1) of the present invention as active ingredient at a predetermined concentration. 21 Days later, the herbicidal effect was examined, the percentages of killed weeds were calculated in the same manner as in Test Example 1, and the phytotoxicity of a rice plant also examined and judged according the criterion shown in Test Example 1.

Results of the test are shown in Table 6.

Table 6

Comp'd No.	Amount of active ingredient Kg/ha	Phyto- toxicity rice	Effect of past-emergence treatment			
			Barnyardgrass	Hotarui	Mizugaytsuri	Urikawa
2	5	1	4	1	0	0
3	5	2	5	4	2	3
4	5	5	5	5	3	4
5	0.8	3	5	3	1	2
	5	5	5	5	5	5
6	0.8	4	5	4	4	3
	5	3	5	4	5	3
7	5	4	5	4	3	1
8	0.8	2	4	1	0	1
	5	3	5	4	1	2
9	5	5	5	5	5	5
10	0.8	4	5	4	5	5
	5	5	5	5	5	5
	0.8	5	5	4	5	5
	0.1	2	5	2	3	4

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5	5	4	2	5	5	3	5	5	2	5	5	3	3	3	2	3	4	5	4	5	4
10	5	5	5	5	5	0	5	5	0	5	5	0	1	0	0	5	2	3	3	4	3
15																					
20	5	4	3	5	4	2	5	4	1	5	4	2	2	2	1	2	3	5	2	5	3
25																					
30	5	5	5	5	5	3	5	5	3	5	5	3	4	5	5	5	5	5	4	5	3
35	5	4	2	5	4	0	5	4	1	5	5	1	1	2	1	2	2	5	3	5	2
40																					
45	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	5	5	5	5	5	5	5	0.8	5	0.8
50	11			12			13			14		16	18	19	20	21	23	25		26	
55																					

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5 3 5 4 5 4 5 4 5 4 5 4 5 4 5 4 5 4 5 5

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5 2 5 2 5 3 5 2 5 2 5 2 5 2 5 2 5 2 4 2

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5 3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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4 3 5 2 5 3 5 2 5 3 5 2 5 2 5 2 5 2 5 2

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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27 28 29 30 31 32 33 34 35 37

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38	5	2	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
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2 1 2 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 1 5 2

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2 0 4 2 0 2 0 2 0 1 2 0 1 0 3 0 5 5 3 5 5

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2 2 3 3 0 3 0 5 3 2 3 0 2 0 5 3 5 5 3 5 5

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5 2 5 5 2 5 2 5 5 3 5 2 3 0 5 5 5 5 5 5 3

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2 0 2 3 1 3 1 5 2 3 3 1 5 2 5 2 5 5 2 5 3

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5 5 5 5 0.8 5 0.8 5 0.8 5 5 0.8 5 0.8 5 0.8 5 0.8 0.1 5 0.8

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55 57 59 62 63 64 66 67 68 69 71 72

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2 4 1 0 2 2 0 5 5 5 5 5 5 5 5 5 1 5 4 5 5

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4 4 0 4 5 1 2 5 5 5 5 4 5 5 5 5 5 5 5 5 5

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4 4 1 2 4 4 2 5 5 5 4 4 5 4 3 5 0 5 3 5 4

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4 5 3 0 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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2 5 3 2 2 2 1 5 3 5 3 3 5 3 2 5 3 5 3 5 3

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5 5 0.8 5 5 5 5 5 0.8 5 0.8 5 5 0.8 5 5 0.1 5 0.1 5 0.8

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73 74 75 76 77 78 81 82 83 84 85 86 87 88

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5 5 5 3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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5 5 5 3 5 5 5 5 5 0 5 5 5 5 5 5 5 5 5 5

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5 5 5 5 5 4 5 4 5 2 5 5 5 5 5 5 5 4 5 4

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5 4 5 3 5 3 5 3 5 2 5 3 5 4 5 4 5 2 5 2

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5 0.8 5 0.8 5 0.1 0.8 0.1 0.8 0.1 0.8 0.1 5 0.8 5 0.8 0.8 0.1 0.8 0.1

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89 90 91 92 93 94 95 96 97 98

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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5 5 5 5 5 1 5 5 5 0 5 0 5 4 5 0 5 4 5 5

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5 5 5 5 5 3 5 5 5 0 5 0 5 4 5 4 5 4 5 5

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5 3 5 4 5 2 5 4 5 2 5 2 5 4 5 3 5 3 5 3

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5 0.8 5 0.8 0.8 0.1 5 0.8 0.8 0.1 0.8 0.1 0.8 0.1 0.8 0.1 5 0.1 5 0.1

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99 100 101 102 103 104 105 106 108 109

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5 5 5 3 5 1 5 5 5 5 5 4 5 0 5 5 5 5 3

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5 3 4 2 5 0 5 2 5 5 5 2 5 0 5 3 3 0 5 3

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5 4 5 2 5 2 5 3 5 3 5 2 5 3 5 2 5 3 5 2

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5 0.1 5 0.8 0.8 0.1 5 0.8 5 0.8 5 0.8 5 0.1 5 0.8 5 0.8 5 0.8

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110 111 112 113 114 115 116 117 118 119

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120	5	5	5	5	4	5	5	5	5	4	5	5	4	5	5	4	5	5	4	5	5
121	5	5	5	5	3	5	0	5	5	3	5	5	1	5	5	2	5	4	0	5	5
122	3	5	3	5	3	5	3	5	5	3	5	5	2	5	4	3	5	5	2	5	5
123	5	5	5	5	5	5	5	5	5	4	5	5	5	5	5	3	5	5	1	5	5
125	3	5	3	4	3	5	2	5	4	3	5	5	3	5	5	3	5	5	3	5	5
126	5	5	0.8	5	0.8	5	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8
127	5	5	0.8	5	0.8	5	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8
128	5	5	0.8	5	0.8	5	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8
129	5	5	0.8	5	0.8	5	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8

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5	5	5	5	5	3	5	3	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
10	2	5	2	5	0	5	1	5	5	5	5	5	5	0	5	5	5	3	5	5	2	5	5
15																							
20	4	5	5	5	2	5	3	5	5	0	5	5	5	5	5	5	5	4	4	3	2	5	5
25																							
30	5	5	5	5	5	5	4	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
35	5	5	2	5	3	5	3	5	5	3	5	5	3	5	5	3	5	5	3	5	3	5	5
40																							
45	0.1	0.8	0.1	5	0.8	5	0.8	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.1	5	0.8
50	129	130	131		132		133			134		135			136					137			
55																							

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- to be cont'd -

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5	5	5	4	5	5	5	5	5	5	5	5	5	5	5	3	5	5	5	5	3	5
10	5	5	0	5	5	5	5	5	5	5	4	2	5	5	0	5	3	5	5	3	-
15																					
20	5	5	2	4	5	5	5	5	5	5	3	2	5	5	1	5	5	5	5	2	4
25	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
30																					
35	5	5	3	2	2	5	3	5	3	5	4	2	5	5	3	5	3	5	5	3	3
40																					
45	5	0.8	0.1	5	5	5	0.8	5	0.8	5	0.8	0.1	5	0.8	0.1	5	0.8	5	0.8	0.1	5
50	145			146	147	148		149		150			151			152		153			154
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5 5 5 5 5 5 5 5 5 5 4 5 5 5 5 5 5 5 5 5

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5 5 5 5 5 5 3 1 2 0 2 0 5 4 2 5 5 5 5 3

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5	5	5	5	4	4	5	5	5	4	5	3	5	5	5	5	5	4	5	5	5	5
10	5	5	0	4	5	2	0	4	2	5	5	2	0	4	5	5	4	5	5	1	5
15																					
20	5	5	3	5	4	5	3	5	2	4	2	2	0	0	4	2	0	5	5	2	3
25	5	5	5	5	5	5	5	5	5	5	5	5	4	5	5	5	4	5	5	3	5
30																					
35	5	5	2	3	3	5	3	5	3	5	2	5	2	3	3	5	2	2	3	3	2
40	5	0.8	0.1	5	5	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	0.8	5	5	5	5
45																					
50	164			165	166	167		168		169		170		171	172	173		174	175	176	177
55																					

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178	4	2	0	5	2	5	5	5	5	3	5	5	0	2	0	5	1	5	5	5	5
179	5	4	2	2	5	5	5	5	0	5	5	5	4	4	2	3	1	5	5	3	-
180	3	3	0	2	4	5	4	4	4	3	4	4	2	2	0	5	0	5	3	5	4
181	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	5	5	4
182	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
183	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
184	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
185	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
186	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
187	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
188	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
189	0.8	0.8	0.1	5	0.8	5	0.8	5	0.8	0.1	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5
190	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
191	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5

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192 193 194 195 196 197 198 199 200 201 202

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203	5	5	5	5	4	5	5	2	5	5	5	5	5	5	5	5	3	3	5	5	5
204	4	2	4	2	0	4	1	5	5	5	5	5	5	5	5	4	5	5	3	3	2
205	4	2	5	2	4	5	3	2	4	4	1	5	5	5	4	5	5	4	3	5	5
206	5	4	5	5	5	5	5	5	5	5	5	5	5	5	4	5	5	5	5	5	5
207	5	0.8	5	0.8	5	5	0.8	5	5	5	0.8	5	5	5	0.8	5	0.8	5	5	5	0.8
208	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
209	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
210	5	0.8	5	0.8	5	5	0.8	5	5	5	0.8	5	5	5	0.8	5	0.8	5	5	5	0.8
211	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
212	5	0.8	5	0.8	5	5	0.8	5	5	5	0.8	5	5	5	0.8	5	0.8	5	5	5	0.8
213	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
214	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
216	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
218	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5

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220	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	3
221	0.8	3	5	3	5	3	5	3	5	3	2	2	3	5	4	3	5	3	5	2	5	0	5	2
222	5	5	5	5	5	5	5	5	5	5	5	5	4	5	5	5	5	5	5	5	5	5	5	2
223	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
224	5	3	5	3	5	3	5	3	5	3	2	2	3	5	4	3	5	3	5	2	5	0	5	2
225	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
226	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
227	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
228	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
	0.1	5	0.1	5	0.1	5	0.1	5	0.1	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
229	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
230	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
	0.1	5	0.1	5	0.1	5	0.1	5	0.1	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
231	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2
	0.1	5	0.1	5	0.1	5	0.1	5	0.1	5	5	5	5	5	5	5	5	5	5	5	5	5	5	2

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5 0.8 0.8 0.1 0.8 0.1 0.8 0.1 5 0.8 0.8 0.1 5 0.8 0.1 0.8 0.1 5 0.8 0.1 0.8

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5	3	5	5	5	5	5	5	5	2	5	0	5	5	0	5	5	2	5	1	5	0	
5	4	5	4	5	0	5	0	4	5	2	5	5	5	5	5	5	5	5	3	5	5	
5	5	5	5	5	5	5	5	2	5	3	5	5	5	5	5	5	5	5	2	5	3	
5	3	3	1	5	3	5	1	3	5	2	5	4	3	5	4	2	5	3	5	2		
0.8	0.1	5	0.8	5	0.8	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	0.8	0.1	0.8	0.1		
242	243		244		245		246	247		248		249			250		251					

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5 3 5 4 5 4 5 5 5 3 5 3 5 4 5 4 3 5 3 5 3

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5 0.8 0.8 0.1 0.8 0.1 0.2 0.1 0.2 0.1 0.2 0.1 0.2 0.1 0.8 0.2 0.1 0.8 0.2 0.2 0.1

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5 5 5 4 5 4 5 5 5 5 5 5 3 5 5 5 5 5 5 5 0

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5 5 5 4 5 5 5 5 5 5 5 5 0 5 3 5 5 3 5 5

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5 5 5 4 5 5 5 5 5 5 5 5 3 5 4 5 5 5 5 5

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 0.1 5 0.8 5 0.8 0.1 0.8 0.1

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
10	5	0	5	4	5	5	5	4	0	5	5	0	5	5	0	5	5	5	5
15																			
20	5	4	5	5	5	5	5	5	2	5	5	2	5	5	3	5	5	5	0
25																			
30	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4
35	5	3	5	3	5	3	5	4	3	5	4	3	5	4	2	5	3	5	4
40																			
45	0.8	0.1	5	0.8	5	0.8	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	5	0.8
50	289	290	291	292		293	294		295	296									
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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 0 5 5 5 5

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5 4 4 2 5 5 2 0 5 4 5 5 0 5 3 5 0 5 0 5 4

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5 3 5 3 5 5 5 4 5 4 5 5 3 5 5 5 4 5 3 5 5

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5 5 5 5 5 5 5 4 5 5 5 5 5 5 5 5 5 5 5 5 5

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5 3 5 3 5 3 5 3 5 3 5 4 2 5 3 5 3 5 3 5 3

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 0.1 5 0.8 0.8 0.1 0.8 0.1 5 0.8

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297 298 299 300 301 302 303 304 305 306

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5	5	0	5	5	5	5	5	5	5	5	5	5	5	5	5	4	2	0	5	5	5
10	0	0	5	5	3	5	5	0	5	4	0	5	5	0	2	0	0	5	5	0	
15																					
20	5	2	5	5	5	5	5	5	5	5	0	5	5	3	5	5	3	5	5	3	
25																					
30	5	3	5	5	5	5	5	5	5	5	5	5	5	4	5	5	3	5	5	5	
35	5	3	5	4	3	5	4	3	5	4	2	5	4	2	5	4	2	5	4	3	
40																					
45	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	5	0.8	0.1	
50	307	308				309			310			311			312			313			

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5 4 4 5 5 0 5 5 0 5 5 0 5 4 0 5 3 3 5 5 2

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5 5 0 5 5 5 5 5 5 4 5 5 4 5 5 2 5 4 2 5 4 5

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5 5 3 5 5 5 5 5 5 5 5 5 5 5 5 3 5 5 4 5 5 5

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5 4 2 5 4 3 5 4 3 5 4 2 5 4 2 5 4 2 5 5 3 3

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5 0.8 0.1 5 0.8 0.1 5 0.8 0.1 5 0.8 0.1 5 0.8 0.1 5 0.8 0.1 5 0.8 5

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314 315 316 317 318 319 320 321

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 5 5

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0 5 0 5 5 5 5 5 5 5 5 5 5 5 5 4 2 2 0 5 0

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2 5 3 5 5 5 5 5 5 5 5 5 5 5 5 5 3 4 2 5 5

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5 4 1 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 3 5 3

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1 5 3 5 5 5 5 5 5 5 5 5 5 5 5 5 2 3 1 5 2

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0.8 0.8 0.1 0.8 0.1 0.8 0.1 0.8 0.1 0.8 0.1 0.8 0.1 0.8 0.1 5 0.8 5 0.1 0.8 0.1

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321 322 323 324 325 326 327 328 329 330 331

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	
3	5	3	5	5	5	5	5	3	5	5	5	0	5	3	5	5	5	0	5	5
5	5	5	5	5	5	5	5	5	5	5	5	0	5	3	5	5	5	4	5	4
5	5	5	5	5	5	5	5	5	5	5	5	1	5	5	5	5	5	5	5	5
3	5	5	5	5	5	5	4	3	5	4	5	1	5	5	5	5	5	3	3	1
0.1	0.2	0.1	0.2	0.1	0.8	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.1	0.8	0.2
342	343	344	345			346	347	348	349	350	351									

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
5	5	5	5	5	5	5	5	0	5	0	5	4	5	0	5	4	5	5	5	5
5	5	5	5	5	4	5	4	5	2	5	4	5	0	4	-	5	5	5	5	3
5	5	5	5	5	5	5	5	5	5	4	5	5	5	4	5	5	5	5	5	5
0.2	0.1	0.2	0.1	0.8	0.2	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.8	0.2	0.2	0.1	0.2	0.2	0.1
352	353	354	355	356	357	358	359	360	361											

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	4
10	5	3	5	0	5	0	5	0	5	3	5	5	5	0	5	5	2	0	2	0	
15																					
20	5	3	5	5	5	3	5	0	5	5	5	5	5	2	5	2	5	3	5	5	
25																					
30	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	3	5	4	
35																					
40	5	3	5	3	5	3	5	3	5	3	5	3	5	3	5	3	5	1	5	2	
45																					
50	362	363	364	365	366	367	368	369	370	371											
55	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	0.8	0.2	0.8	0.1	0.2	0.1	0.2	0.1	0.2	0.1	

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5	5	4	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
10	5	5	5	5	5	5	5	5	5	5	5	0	5	4	3	-	5	4	5
15																			
20	5	5	5	5	5	5	5	5	5	5	5	5	5	2	5	4	5	5	5
25																			
30	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	5	5	4
35	5	5	5	5	5	5	5	5	5	5	5	5	5	2	5	4	5	2	5
40																			
45	0.2	0.1	0.8	0.1	0.8	0.1	0.8	0.1	0.8	0.1	0.2	0.1	5	0.8	5	0.8	5	0.8	5
50	372	373	374	375	376	377	378	379	380	381									

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382	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
383	0.8	5	0.8	0.1	5	0.8	5	0.8	0.1	5	0.8	5	0.8	0.1	5	0.8	5	0.8	0.1	5
384	5	3	5	4	1	5	4	5	5	0	5	3	5	3	5	2	5	4	2	5
385	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
386	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
387	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
388	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
389	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
390	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
10	5	5	2	0	5	5	2	0	3	0	0	2	0	5	2	2	0	2	0	5	5	5
15																						
20	5	4	5	3	5	5	5	5	5	5	2	5	2	5	4	4	2	5	4	5	5	5
25	5	5	5	5	5	4	5	5	5	5	4	5	3	5	5	2	0	5	5	5	5	5
30																						
35	5	3	4	2	5	4	5	3	5	4	1	3	1	5	3	3	1	5	2	5	2	2
40																						
45	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	0.1	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	0.8
50	391	392	393	394	395		396	397	398	399	400											

- to be cont'd -

5	5	5	5	5	5	5	5	5	5	5	4	5	3	0	0	0	0
10	5	5	5	5	5	0	5	0	5	0	4	0	0	0	0	0	0
15																	
20	5	5	5	5	4	2	5	5	4	3	3	0	0	0	0	0	0
25	5	5	5	5	5	3	5	5	5	3	4	3	0	2	0	0	0
30																	
35	5	4	5	4	5	4	5	5	5	3	4	3	0	0	0	0	0
40																	
45	5	0.8	5	0.8	0.2	0.1	0.2	0.1	0.2	0.1	0.2	0.1	5	5	5	5	5
50	401	402	403	404	405	406							A	B	C	D	

As shown in table 6, the present 3-(substituted phenyl)pyrazole derivatives or salt thereof, represented by the general formula (I), exhibit more excellent controlling effects of weeds than comparative compound A, B, C or D at post-emergence treatment in paddy fields. Even when the derivatives exhibit both herbicidal activity on weeds and phytotoxicity on crops at the same dosage, however, by the selection of a proper lower dosage, sufficient herbicidal activity is retained but phytotoxicity on crops is reduced remarkably.

Test Example 3 Herbicidal effect on upland field weeds of pre-emergence stage.

Polyethylene vats of 10 cm W x 20 cm L x 5 cm depth were filled with soil, and planted separately with seeds of barnyardgrass, velvetleaf, cocklebur, jimsonweed, birdseye speedwell, cleavers, which are upland field weeds, and separately with seeds of soybean and wheat as upland field crops.

Soil in each pot was sprayed with each of solutions containing compounds (listed in Table 1) of the present invention as active ingredient at a predetermined concentration. 14 Days later, the herbicidal effect was examined, the percentages of killed weeds were calculated in the same manner as in Test Example 1, and the phytotoxicity of soybean and wheat plants were also examined and judged according the criterion shown in Test Example 1.

Results of the test are shown in Table 7.

Table 7

Comp'd No.	Amount of active ingredient Kg/ha	Phytotoxicity		Pre-emergence treatment					
		wheat	soybean	banyard-grass	velvet-leaf	cocklebur	jimsonweed	birdseye speedwell	cleavers
3	5	1	0	5	5	5	5	-	5
4	5	2	1	5	5	0	4	-	5
5	5	3	0	5	4	2	5	-	5
6	5	1	2	5	3	0	2	-	3
7	5	0	1	4	4	0	2	-	2
8	5	0	0	3	3	0	1	-	1
9	5	4	4	5	5	-	5	-	5
10	0.8	2	0	5	4	3	5	-	-
	5	5	4	5	5	5	5	-	5
11	0.8	3	2	3	5	5	5	-	-
	5	5	5	5	4	4	5	-	5
12	5	4	4	5	4	-	5	-	5
13	0.8	3	1	5	2	-	5	-	-
	5	4	4	5	5	5	5	-	5
	0.8	3	1	5	5	5	5	-	-

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5 5 3 0 4 - 0 2 0 0 2 5 4 5 4 5 4 - 5 4

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5 5 4 1 4 5 3 4 4 0 3 5 4 5 4 4 3 5 4 3

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3 0 - 0 2 1 0 0 0 1 0 3 1 4 2 2 0 3 3 1

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5 3 1 0 1 1 1 0 0 0 1 4 2 4 2 4 2 2 3 1

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5 0.8 5 5 5 5 5 5 5 5 5 0.8 5 0.8 5 0.8 5 5 0.8

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14 16 17 18 19 20 21 23 24 25 26 27 28 29 30

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31	5	3	2	5	4	5	4	5	3	2	5	5	0	4	0	5	5	4	4	5	4	5
32	5	3	2	5	4	5	4	5	3	2	5	5	0	4	0	5	5	4	4	5	4	5
33	0.8	1	0	0.8	3	0	4	3	0	0	0	0	0	0	0	3	5	2	4	5	2	5
34	5	4	2	5	3	2	3	2	3	0	4	5	0	0	0	0	0	0	0	1	0	0
35	0.8	1	0	0.8	3	2	3	2	3	0	4	5	0	0	0	0	0	0	0	1	0	0
37	5	3	3	5	4	5	5	0	2	3	4	1	2	3	2	3	2	3	5	2	4	5
38	5	2	1	5	5	5	5	0	2	3	4	1	2	3	2	3	2	3	5	2	4	5
40	5	0	0	5	0	2	3	4	2	3	4	1	2	3	2	3	2	3	5	2	4	5
41	5	2	0	5	2	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
42	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
43	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
44	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
45	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
46	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
47	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
48	5	0	0	5	0	2	0	2	2	3	4	1	2	3	2	3	2	3	5	2	4	5
49	5	3	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5

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50 51 52 54 55 57 59 71 72 73 74 76 77 78 81 82 83

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5	-	5	4	5	5	5	5	5	5	-	5	5	0	0	5	5	5	5	5	5	5
-	-	-	-	-	-	-	-	-	-	-	5	5	-	-	-	-	5	5	5	5	5
5	5	5	5	5	5	5	5	5	5	5	-	-	5	4	5	5	-	-	-	-	-
5	4	5	4	5	5	5	5	5	5	3	3	1	-	3	5	5	5	5	4	5	4
5	5	5	5	5	5	5	5	5	5	4	5	3	5	4	5	5	5	5	4	5	3
5	5	5	4	5	5	5	5	5	5	5	5	5	5	4	5	4	5	5	5	5	4
4	2	4	1	5	3	4	3	3	0	3	1	2	0	4	4	2	3	1	2	1	1
5	3	5	2	5	3	5	3	4	3	4	1	4	2	4	4	2	4	1	4	1	1
5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
84	85	86	87	88	89	90	91	92	93												

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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	-	-
-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	5	-	
5	5	3	1	5	5	5	3	5	4	5	4	3	1	3	1	5	4	2	0		
5	5	4	2	5	5	5	5	5	5	5	3	5	4	5	5	5	5	5	5	4	
5	5	5	5	5	5	5	4	5	4	5	4	5	3	5	4	5	4	5	5	5	
2	1	2	1	4	1	3	1	3	1	3	1	3	1	3	1	3	1	3	1		
4	2	3	1	4	1	3	1	4	1	3	1	4	1	4	1	4	1	4	3		
5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8		
94	95	96	97	98	99	100	101	102	103												

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104	5	5	5	5	5	4	5	5	5	5	5	5	5	5	5	3	5	-	5	-
105	-	-	-	-	5	5	5	5	5	5	-	-	-	-	-	-	-	-	-	-
106	5	-	5	5	-	-	-	-	-	-	5	5	5	5	5	4	2	5	5	5
107	-	0	-	-	2	-	3	-	3	-	2	-	2	-	1	-	5	3	5	4
108	5	5	5	5	5	5	5	4	5	5	5	5	5	5	5	5	4	5	5	5
109	5	4	5	5	5	5	5	5	5	5	5	5	5	5	5	2	1	5	5	5
110	4	2	3	1	3	1	3	1	3	1	3	1	3	1	2	0	4	2	4	2
111	4	3	4	3	4	3	5	3	4	3	4	3	5	3	4	1	4	3	5	3
112	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
113	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8

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5 3 5 4 5 5 5 5 5 5 4 5 5 5 4 5 4 5 4 5 4

3 1 3 0 4 2 3 0 2 0 2 0 3 1 3 1 3 0 3 1

3 0 3 0 4 3 4 2 4 2 4 2 3 1 4 2 4 2 4 2

5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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125	5	5	5	5	5	5	5	5	4	5	5	5	5	5	5	4	2	0	5	4	5	5
126	5	5	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
127	-	-	5	5	5	5	4	3	5	5	5	5	5	5	4	2	0	5	5	5	5	4
128	2	1	-	-	-	-	0	0	-	-	0	0	-	-	-	-	-	-	-	-	-	-
129	5	4	5	5	5	5	5	4	5	5	5	4	4	3	3	1	5	5	5	5	5	5
130	5	5	5	5	5	4	5	4	5	5	5	5	5	4	4	3	5	4	4	4	3	3
131	3	1	4	2	4	1	4	1	4	1	4	0	4	0	3	0	4	2	3	1	1	1
132	4	2	4	2	4	2	4	2	4	2	4	1	4	0	3	0	4	2	4	4	2	2
133	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
134	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8

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5	5	4	5	4	5	5	-	-	5	5	5	5	5	5	5	5	5	5	5	4
10	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
15	5	4	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
20	-	-	-	-	-	-	5	1	5	3	-	-	-	-	-	-	-	-	-	-
25	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4
30	5	4	5	4	5	4	5	4	5	5	5	5	5	5	5	5	5	5	5	4
35	4	2	4	2	4	2	4	2	4	2	4	2	3	1	4	3	4	2	3	1
40	4	2	4	2	4	2	4	2	4	2	4	3	4	2	4	3	5	3	3	1
45	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
50	135	136	137	138	139	140	141	142	143	144										
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5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	4	5	5	5	4
10	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
15	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	4	5	4
20	2	0	4	3	2	5	4	2	3	1	5	4	5	3	3	2	4	3	2	0	
25	5	5	5	5	5	5	5	5	5	5	5	5	5	5	4	2	5	4	4	3	
30	5	4	5	5	5	5	5	5	5	5	5	5	5	5	5	4	5	4	5	4	
35	5	3	4	2	0	5	3	1	3	1	4	2	4	2	2	0	3	0	3	0	
40	5	3	3	2	2	3	4	2	5	3	4	2	4	3	3	1	3	1	3	1	
45	5	0.8	5	5	5	5	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	
50	145	146	147	148	149	150		151	152	153	154	155	156								
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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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5	1	0	5	3	4	2	2	0	5	5	5	5	2	1	1	1	1	5	4	5	5
10	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	5	-	-	5	5
15	2	0	5	5	3	1	5	3	5	5	5	3	2	2	2	2	2	5	4	-	0
20	0	0	0	0	-	0	-	0	-	0	-	-	-	0	0	0	-	-	0	-	-
25	4	3	4	3	4	2	5	4	4	3	3	1	2	2	2	5	3	2	4	2	4
30	5	4	5	4	2	1	4	3	5	4	4	4	4	4	5	5	2	3	5	4	4
35	3	1	3	0	2	0	3	0	3	0	2	0	2	2	2	2	2	2	3	0	3
40	3	1	3	1	3	1	3	0	3	0	3	1	3	3	3	3	2	2	3	1	3
45	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	5	5	5	0.8	5
50	178	179	180	181	182	183	184	185	186	187	188	189	190								
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191	5	5	5	5	1	5	5	1	0	1	0	5	3	5	5	2	1	5	5	5	5
192	5	5	5	5	-	-	-	-	-	-	-	-	-	5	5	5	5	5	5	5	5
193	-	-	-	-	2	5	5	4	3	4	3	5	5	-	-	-	-	-	-	-	-
194	2	1	-	0	-	5	2	-	0	-	0	-	-	-	-	2	1	-	0	-	0
195	5	3	2	1	3	5	5	5	5	5	5	4	3	5	3	4	2	3	1	4	2
196	4	3	5	2	2	5	4	5	4	5	4	5	4	4	3	4	3	3	2	4	3
197	3	1	3	0	1	3	1	3	1	3	1	3	0	3	1	3	1	3	0	3	1
198	3	1	3	0	2	4	2	3	1	3	0	3	1	3	1	3	1	3	1	3	0
199	5	0.8	5	0.8	5	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
200	5	0.8	5	0.8	5	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
201	5	0.8	5	0.8	5	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8

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202	5	2	2	0	2	3	1	3	2	0	2	0	2	0	1	1	2	0	4	2
203	0.8	0	0	0	0	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
204	5	2	2	2	2	3	2	3	4	3	4	3	4	3	4	4	4	2	5	5
206	0.8	0	0	0	0	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
207	5	2	2	2	2	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
208	0.8	0	0	0	0	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
209	5	2	2	2	2	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
210	0.8	0	0	0	0	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
212	5	2	2	2	2	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
216	5	2	2	2	2	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5
220	0.8	0	0	0	0	3	1	1	4	3	4	3	4	3	4	4	4	2	5	5

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0 0 1 0 1 0 2 - 2 - 2 - 1 0 1 0 1 0 5 3

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5 4 5 4 5 3 5 5 5 5 5 5 5 4 5 5 5 4 5 4

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3 2 4 2 3 2 5 4 5 4 5 4 5 3 5 4 4 2 5 4

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2 0 1 0 2 0 4 2 3 1 3 1 2 0 2 0 1 0 3 1

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3 1 2 0 3 1 4 2 3 1 3 1 3 1 3 1 2 0 3 1

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5 0.8 0.8 0.2 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 0.2 5 0.8

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235 236 237 238 239 240 241 242 243 244

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245	5	5	2	1	5	4	5	5	5	5	5	5	5	4	2	5	4	5	5	5	5
247	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
248	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
249	1	0	0	0	2	0	2	0	1	0	4	2	1	0	3	2	2	0	4	2	
250	5	4	4	1	5	3	5	4	5	5	5	5	4	2	5	5	5	4	5	5	
251	5	3	3	2	5	3	5	3	5	5	5	5	4	3	5	4	5	4	5	4	
252	3	1	3	1	2	0	3	1	3	1	3	1	2	0	3	1	3	1	5	3	
253	3	1	3	1	3	1	3	1	4	2	4	2	3	1	4	2	3	1	5	3	
254	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	0.8	0.2	
255																					

- to be cont'd -

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5	5	5	5	4	5	5	4	2	3	1	5	5	4	2	3	2	3	0	3	0
10	5	5	5	5	5	5	5	4	5	5	5	5	5	5	5	5	5	5	5	5
15	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
20	2	1	2	0	4	2	1	0	1	0	4	3	1	0	1	0	1	1	1	0
25	5	5	5	5	5	5	3	1	5	3	5	5	5	3	4	2	4	1	5	4
30	5	4	5	5	5	4	4	2	4	3	5	4	4	2	3	0	3	0	4	2
35	3	2	2	0	4	3	3	0	2	0	3	1	1	0	1	0	1	0	1	0
40	5	3	3	1	5	3	3	0	3	1	4	2	2	0	2	0	2	0	3	1
45	0.2	0.1	0.8	0.2	0.2	0.1	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2
50	256	257	258	259	260	261	262	263	264	265										
55																				

- to be cont'd -

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5 5 5 3 4 3 4 2 5 3 5 5 5 1 5 4 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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1 0 1 0 0 0 1 0 3 3 2 0 1 0 1 1 1 0

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3 0 4 2 4 2 5 3 5 5 5 5 4 2 5 5 5 4

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4 2 3 0 5 2 5 5 5 4 5 5 5 4 5 3 5 3

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1 0 2 1 1 0 1 0 1 0 2 1 1 0 1 0 1 0

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2 0 2 0 3 1 3 1 3 1 3 2 3 1 3 1 3 1

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0.8
0.2
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- to be cont'd -

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276	5	4	3	4	2	4	3	4	3	5	5	5	5	5	5	5	5	5	5	5	5	5
277	0.8	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
278	0.8	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
279	0.8	1	0	1	0	1	0	1	0	1	0	1	0	2	0	2	0	1	0	3	2	
280	5	5	4	5	5	5	4	5	3	5	3	5	5	5	5	5	5	5	3	5	5	
281	0.8	5	3	4	3	4	3	4	2	5	4	5	4	5	4	5	4	5	4	5	4	
282	0.8	3	1	3	1	3	1	3	0	2	0	3	1	3	1	3	1	3	1	3	1	
283	0.8	4	2	3	1	3	1	3	0	4	2	3	1	3	1	3	1	3	1	3	1	
284	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	
285	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	

- to be cont'd -

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5 5 5 5 5 5 5 5 4 5 5 5 5 5 5 5 5 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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2 0 2 - 2 - 2 - 2 1 2 1 4 2 2 0 1 0 1 0

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5 4 5 5 5 5 5 5 5 5 3 1 5 4 5 4 3 1 5 4

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5 3 5 4 5 4 5 4 5 5 5 4 5 3 3 2 4 3 5 4

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3 1 2 0 3 1 3 1 3 1 3 1 2 0 1 0 1 0 2 0

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3 1 4 2 4 2 4 2 3 1 3 1 3 1 3 1 3 1 3 1

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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286 287 288 289 290 291 292 293 294 295

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- to be cont'd -

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5	5	5	5	5	5	5	5	5	4	5	4	5	5	5	5	5	5	5	5	3
10	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
15	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
20	1	0	1	-	1	-	1	0	1	0	2	0	1	0	1	0	2	1	1	0
25	5	4	5	4	5	4	5	4	5	3	5	5	5	3	5	3	5	4	5	5
30	5	4	5	4	4	3	5	3	4	2	5	4	4	3	4	3	5	4	5	4
35	3	1	3	1	3	1	3	1	2	0	2	0	2	0	2	0	3	1	3	1
40	4	2	3	1	3	1	4	2	3	1	4	2	3	1	4	2	3	1	3	1
45	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8
50	296	297	298	299	300	302	301	303	304	305										
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- to be cont'd -

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5 5 5 5 5 4 5 5 5 5 5 5 5 5 5 5 5 5 5

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5 3 5 4 5 5 5 5 5 5 4 2 5 5 5 5 5 4 5 4

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4 3 5 4 5 4 5 4 5 3 3 2 5 4 4 2 5 3 5 3

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3 1 2 0 3 1 4 2 5 3 2 0 3 1 2 0 3 1 3 1

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4 2 3 1 4 2 4 2 4 2 3 1 3 1 3 1 3 1 4 2

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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306 307 308 309 310 312 313 314 315 316

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317	5	5	5	5	5	5	5	4	2	5	4	4	2	5	4	5	3	5	4	5	5
318	5	5	5	5	5	5	5	5	5	5	5	5	3	5	5	5	5	5	5	5	5
319	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
320	2	1	2	1	5	3	2	1	1	0	1	0	5	3	4	3	-	-	5	3	
321	5	4	3	1	5	5	5	5	5	3	4	1	5	4	5	4	5	5	5	5	5
322	5	3	5	4	5	5	5	4	4	2	3	0	5	5	5	5	5	5	4	5	4
323	3	1	2	0	3	1	3	0	2	0	1	0	4	2	4	2	4	2	3	1	
324	4	2	3	1	3	1	3	0	3	1	2	0	4	2	5	3	3	1	4	2	
325	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	
326	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	

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5 5 5 4 4 1 5 5 5 5 4 2 - 5 5 5 - 5 4 5 3

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 5 5 5 5

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5 2 4 2 1 0 1 5 5 5 3 1 1 1 0 1 0 2 1 1 0

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5 5 5 5 3 0 5 5 5 5 4 5 3 4 2 5 0 5 5 3 1

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5 4 5 4 3 1 4 5 5 5 3 4 3 3 2 4 0 5 4 4 3

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3 1 4 2 1 0 1 5 4 5 3 2 0 2 0 1 0 2 0 2 0

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4 2 3 1 2 0 3 5 3 5 3 3 1 2 0 2 0 2 0 2 0

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5 0.8 5 0.8 5 0.8 5 5 0.8 0.8 0.2 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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327 328 329 330 331 332 333 334 335 336 337

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- to be cont'd -

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5 5 4 5 5 4 3 4 2 4 2 4 2 4 2 4 2 4 1 4 1

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5 5 5 5 5 5 3 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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1 0 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0

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5 3 4 4 2 4 1 5 5 5 5 5 3 5 4 5 4 5 4 5 5

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4 3 4 3 2 3 0 5 4 5 4 5 3 5 3 5 3 5 3 5 3

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2 0 1 2 0 1 0 1 0 1 0 1 0 2 1 1 0 1 0 1 0

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3 1 2 2 0 3 1 3 1 3 1 3 1 3 1 3 1 3 1 3 1

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5 0.8 5 5 0.8 0.8 0.2 0.8 0.2 0.8 0.2 0.8 0.2 0.8 0.2 0.8 0.2 0.8 0.2 0.8 0.2

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338 339 340 341 342 343 344 345 346 347 348

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4	1	4	4	4	1	5	4	3	0	4	4	4	0	4	0	4	0	3	0
5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1	0	1	0	1	-	1	0	1	0	0	0	0	0	0	0	1	0	1	0
5	4	5	3	4	3	5	4	5	4	4	2	4	2	4	1	4	1	5	4
4	2	4	2	3	0	4	2	4	2	4	2	4	2	4	1	4	3	4	3
1	0	1	0	1	0	1	0	1	0	1	0	1	0	1	0	1	0	1	0
3	1	3	1	2	0	3	1	2	0	3	1	2	0	2	1	2	0	2	0
0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2
349	350	351	352	353	354	355	356	357	358										

- to be cont'd -

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4	1	5	3	4	0	5	3	5	4	4	2	4	1	3	0	3	0	5	5
5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	0	5	5
-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
1	0	1	-	1	0	1	0	2	1	1	0	1	0	1	0	1	0	2	1
5	5	5	5	4	3	5	5	5	5	5	5	5	5	4	2	4	2	5	5
5	3	5	4	4	3	4	3	5	4	5	3	5	4	4	3	4	3	5	5
1	0	1	0	1	0	1	0	3	1	1	0	1	0	1	0	1	0	3	1
3	1	3	1	3	1	3	1	5	3	3	1	2	0	3	1	2	0	4	2
0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	5	0.8	0.8	0.2	0.8	0.2	0.8	0.2	0.8	0.2	5	0.8
359	360	361	362	363	364	365	366	367	368										

- to be cont'd -

- to be cont'd -

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5 5 3 2 3 2 5 5 3 - 2 - 2 - 5 4 5 5 5 5

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5 5 5 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5

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2 1 1 0 1 0 1 0 1 0 1 0 1 0 1 0 1 0

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5 3 4 2 5 4 5 3 5 4 4 2 4 2 5 3 5 4 5 4

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4 3 3 2 3 2 5 4 5 3 4 3 4 3 4 2 4 3 4 3

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3 1 1 0 3 1 2 0 2 0 2 0 3 1 1 0 2 0 2 0

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3 1 3 1 2 0 2 0 3 1 3 1 4 2 3 0 3 1 3 1

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5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8 5 0.8

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379 380 381 382 383 384 385 386 388 389

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- to be cont'd -

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390	5	5	5	5	5	4	5	5	5	5	5	5	5	4	0	3	3	0	4	2	5	4
391	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5
392	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
393	1	0	0	0	1	0	1	0	0	0	1	0	1	0	1	1	0	1	0	5	4	
394	5	5	3	1	4	1	4	2	4	2	5	2	3	0	3	3	1	4	2	5	3	
395	4	3	4	3	4	2	4	2	4	2	4	2	4	2	3	4	2	4	3	5	4	
396	3	1	2	0	1	0	2	0	3	1	2	0	1	0	1	1	0	1	0	3	1	
398	3	1	3	1	2	0	3	0	3	1	3	1	2	0	2	2	0	2	0	3	1	
399	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	0.8	5	5	0.8	5	0.8	5	0.8	
400																						
401																						

- to be cont'd -

402	5	3	3	5	5	5	-	5	5
	0.8	1	1	4	4	4	-	5	5
403	0.8	4	3	5	5	2	-	5	5
	0.2	2	1	4	5	1	-	5	4
404	4	5	4	5	5	3	-	5	5
	0.8	3	2	5	5	1	-	5	5
405	5	4	3	4	5	1	-	5	5
	0.8	2	1	3	4	0	-	5	4
406	5	4	3	5	4	1	-	5	5
	0.8	2	1	4	4	0	-	5	4
A	5	0	0	0	0	0	0	0	0
B	5	0	0	0	0	0	0	0	0
C	5	0	0	0	0	0	0	0	0
D	5	0	0	0	0	0	0	0	0

As shown in table 7, the present 3-(substituted phenyl)-pyrazole derivatives or salt thereof, represented by the general formula (I), exhibit more excellent controlling effects of weeds than comparative compound A, B, C or D at pre-emergence treatment in paddy fields. Ever when the derivatives exhibit both herbicidal activity on weeds and phytotoxicity on crops at the same dosage, however, by the selection of a proper lower dosage, sufficient herbicidal activity is retained but phytotoxicity on crops is reduced remarkably.

Test Example 4 Herbicidal effect on upland field weeds of post-emergence stage.

Polyethylene vats of 10 cm W x 20 cm L x 5 cm depth were filled with soil, and planted with seeds of upland field weeds shown below and with seeds of soybean and wheat as upland field crops. These weeds and crops were grown to leaf stages shown below. Then these weeds and crops were sprayed separately with solutions containing compounds (listed in Table 1) of the present invention as active ingredient at a predetermined concentration. 14 Days later, the herbicidal effect was examined, the percentages of killed weeds were calculated in the same manner as in Test Example 1, and the phytotoxicity of soybean and wheat plants were examined and judged according to the criterion shown in Test Example 1.

Species of test weeds and leaf stages thereof and leaf stages of test soybean and wheat plants

Barnyardgrass	2-leaf stage
Velvetleaf	2 "
Cocklebur	1 "
Jimsonweed	1 "
Birdseye Speedwell	1 "
Cleavers	2 "
Wheat	2 "
Soybean	1 "

Results of the test are shown in Table 8.

Table 8

Com'd. No.	Amount of active ingredient Kg/ha	Phyto-toxicity		post-emergence treatment						
		wheat	soy-bean	banyard-grass	velvet-leaf	cock-lebur	jimson-weed	birdseye speedwell	cleavers	
2	5	1	2	2	5	4	5	-	-	
3	5	2	4	5	4	4	4	-	4	
4	5	5	5	5	5	5	5	-	5	
5	0.2	3	4	4	5	5	5	-	5	
	5	5	5	5	5	5	5	-	5	
	0.2	3	4	5	5	5	5	-	5	
	5	3	5	5	5	5	5	-	5	
6	5	3	5	5	5	5	5	-	5	
7	5	2	4	4	5	5	5	-	4	
8	5	1	4	2	5	2	5	-	1	
9	5	5	5	4	5	5	5	-	5	
10	0.8	3	5	2	5	5	5	-	5	
	0.8	5	5	5	5	5	5	-	5	
	0.1	4	5	4	5	5	5	-	5	
11	5	5	5	5	5	5	5	-	5	
	0.1	4	4	4	5	5	5	-	5	
	0.8	5	5	4	5	5	5	-	5	
12	0.2	3	4	3	5	5	5	-	5	

(To be continued)

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5 5 5 5 5 4 4 5 5 5 5 5 5 4 5 4 5 5 5 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 4 5 5 4 5 5 5 5

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5 5 5 5 5 4 4 4 5 4 5 5 5 3 5 3 5 5 5 5 5

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4 3 4 3 2 3 3 3 3 5 3 4 4 2 2 2 1 4 4 2 5 3

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5 4 5 4 3 3 2 5 5 3 5 5 3 2 2 2 2 5 5 5 5

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5 4 5 3 2 2 2 2 5 3 3 5 2 1 2 2 3 5 3 5 4

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0.8 0.1 0.8 0.2 5 5 5 5 5 0.8 5 5 0.8 5 5 5 5 0.8 0.2 5 0.8

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13 14 15 16 17 18 19 20 21 22 23 24 25 26 27

(To be continued)

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 3 5 5 5 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 5 5 4 0 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 5 5 5 5 5

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4 2 5 4 5 3 5 2 3 3 5 3 5 3 4 2 0 4 3 2 0 3

35

5 4 5 5 5 5 5 5 5 5 4 5 5 5 5 4 2 1 5 5 2 3 5

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5 3 5 4 5 4 5 3 5 3 5 3 5 3 4 2 0 2 2 0 0 2

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0.8
0.1
5
0.1
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0.8
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0.2
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28 29 30 31 32 33 34 35 36 37 38 40 41 42

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(To be continued)

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5 5 5 5 5 5 5 5 5 5 5 5 5 4 1 4 4 - 3 5 5 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 - - - -

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3 4 5 4 4 4 5 5 5 5 5 5 3 2 2 3 4 4 5 5 5 5

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5 5 5 5 5 5 5 5 5 5 5 5 5 5 4 4 4 3 5 5 5 5

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2 3 3 3 3 2 5 5 5 5 3 4 1 1 2 0 2 0 4 2 4 2

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3 3 5 4 2 5 5 5 5 5 4 5 2 3 2 2 2 2 5 3 5 4

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2 2 2 3 1 2 4 5 3 3 2 2 0 0 0 0 1 0 4 2 4 2

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5 5 5 5 5 5 5 5 0.2 5 5 5 5 5 5 5 5 5 0.8 5 0.8

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43 44 45 46 47 48 49 50 51 52 54 55 56 57 58 59 60 62 63

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(To be continued)

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|----|----|-----|----|-----|----|-----|-----|----|----|----|----|----|----|----|----|----|---|-----|-----|-----|-----|
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 0 | 5 | 5 | 2 | 2 | 1 | 2 | 4 | 3 | 3 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 15 | - | - | - | - | - | - | - | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 1 | 5 | 5 | 3 | 2 | 2 | 2 | 2 | 5 | 4 | 2 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 1 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 3 | 4 | 2 | 3 | 1 | 3 | 0 | 5 | 4 | 1 | 1 | 1 | 2 | 4 | 3 | 1 | 5 | 5 | 5 | 5 |
| 35 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 1 | 5 | 4 | 3 | 3 | 2 | 1 | 5 | 4 | 2 | 5 | 5 | 5 | 5 |
| 40 | 5 | 3 | 4 | 2 | 3 | 1 | 3 | 0 | 3 | 2 | 1 | 1 | 1 | 2 | 2 | 1 | 1 | 5 | 3 | 5 | 3 |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 0.8 | 0.2 | 0.8 | 0.2 |
| 50 | | | | | | | | | | | | | | | | | | | | | |
| 55 | 64 | 66 | 67 | 68 | 70 | 71 | 72 | 73 | 74 | 75 | 76 | 77 | 78 | 79 | 81 | 82 | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | - | - | - | - | 5 | 5 | 5 | 5 |
| 15 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | 5 | 5 | 5 | 5 | - | - | - | - |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 3 | 5 | 5 | 5 | 4 | 5 | 4 |
| 35 | 5 | 4 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 |
| 40 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 3 | 5 | 2 | 5 | 2 |
| 45 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.1 | 0.8 | 0.1 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |
| 50 | 83 | 84 | 85 | 86 | 87 | 88 | 89 | 90 | 91 | 92 | 93 | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | - | - |
| 15 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 3 | 5 | 4 | 5 | 4 | 5 | 3 | 5 | 3 | 3 |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | |
| 40 | 5 | 3 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 3 | 5 | 3 | |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | |
| 50 | | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 | 104 | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 15 | 5 | 5 | - | - | - | - | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | 5 | 4 | 5 | 5 | 5 | 4 | 5 | 4 |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 40 | 5 | 4 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 4 | 4 | 2 | 5 | 3 | 5 | 3 | 4 | 2 | 4 | 4 | 2 |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | |
| 50 | | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 | 113 | 114 | 115 | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 3 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | 5 | 5 | - | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | - | - | | |
| 15 | 5 | 5 | 5 | 5 | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | 5 | 5 | |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | |
| 30 | 5 | 4 | 5 | 3 | 5 | 3 | 5 | 3 | 4 | 3 | 5 | 4 | 5 | 3 | 5 | 4 | 2 | 0 | 5 | 4 | 5 | 4 | | |
| 35 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 1 | 5 | 5 | 5 | 5 | | |
| 40 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 4 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 3 | 0 | 5 | 3 | 5 | 3 | | |
| 45 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | | |
| 50 | | | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 116 | 117 | 118 | 119 | 120 | 121 | 122 | 123 | 124 | 125 | 126 | | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 |
| 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |
| 127 | 128 | 129 | 130 | 131 | 132 | 133 | 134 | 135 | 136 | 137 | | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 15 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 |
| 40 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 3 | 4 | 3 | 3 | 5 | 4 |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 5 | 5 | 5 | 5 | 0.8 |
| 50 | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 138 | 139 | 140 | 141 | 142 | 143 | 144 | 145 | 146 | 147 | 148 | 149 | 150 | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | - | - | - | - | - | - | - | - | - | - |
| 15 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 3 | 5 | 4 | 5 | 5 | 5 |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 3 | 5 | 4 | 5 | 5 | 5 | 5 | 5 |
| 40 | 5 | 4 | 5 | 4 | 5 | 4 | 3 | 1 | 5 | 3 | 4 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 5 | 4 |
| 45 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 |
| 50 | 151 | 152 | 153 | 154 | 155 | 156 | 157 | 158 | 159 | 160 | 161 | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | 5 | 5 | 5 | - | - | - | - | |
| 15 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | - | - | 5 | 5 | 5 | 5 | |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | |
| 30 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 4 | 2 | 5 | 4 | 5 | 4 | | |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 3 | 5 | 4 | 5 | 5 | 5 | 5 | | |
| 40 | 5 | 4 | 5 | 4 | 5 | 3 | 4 | 2 | 5 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 3 | 1 | 5 | 3 | 5 | 4 | | |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | | |
| 50 | 162 | 163 | 164 | 165 | 166 | 167 | 168 | 169 | 170 | 171 | 172 | | | | | | | | | | | | | |

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|----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|---|-----|---|-----|---|-----|---|-----|---|-----|-----|
| 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 15 | - | - | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 20 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 3 | 5 | 4 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 4 | 2 | 5 | 4 | 4 | 2 | 4 | 2 | 5 | 3 | 5 | 4 | 5 | 4 | 4 | 4 | 2 | 4 | 3 | 5 | 4 | 5 | 5 |
| 35 | 5 | 3 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | 5 | 4 | 5 | 5 | 5 | 5 |
| 40 | 3 | 1 | 5 | 3 | 5 | 3 | 3 | 1 | 5 | 3 | 5 | 3 | 5 | 3 | 3 | 1 | 4 | 2 | 5 | 3 | 5 | 5 | 3 |
| 45 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 |
| 50 | | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 173 | 174 | 175 | 176 | 177 | 178 | 179 | 180 | 181 | 182 | 183 | | | | | | | | | | | | |

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| 184 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | 5 | 5 |
| 185 | - | - | - | - | - | - | - | - | - | - | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | - | - |
| 186 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | - | - | - | - | - | - | - | - | 5 | 5 |
| 187 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | 5 | 5 |
| 188 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 |
| 189 | 5 | 3 | 4 | 2 | 5 | 3 | 4 | 2 | 4 | 3 | 5 | 3 | 5 | 4 | 5 | 3 | 4 | 2 | 3 | 1 | 5 | 3 | |
| 190 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 3 | 0 | 5 | 5 | |
| 191 | 4 | 2 | 4 | 2 | 5 | 3 | 3 | 0 | 4 | 2 | 5 | 3 | 5 | 4 | 5 | 3 | 3 | 1 | 3 | 0 | 5 | 2 | |
| 192 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | |
| 193 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | |
| 194 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | - | - | - | - | - | - | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 4 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 |
| 15 | 5 | 5 | 5 | 5 | 5 | 5 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 2 | 4 | 2 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 4 | 5 | 3 | 4 | 3 | 5 | 5 | 5 | 4 | 5 | 3 | 5 | 3 | 5 | 3 | 3 | 5 | 4 | 5 | 4 | 5 | 4 |
| 35 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 2 | 5 | 2 | 5 | 3 | 3 | 3 | 3 | 5 | 2 | 5 | 5 |
| 40 | 5 | 4 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 2 | 5 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 5 | 2 |
| 45 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.1 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | |
| 50 | | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 220 | 221 | 222 | 223 | 224 | 225 | 226 | 227 | 228 | 229 | 230 | | | | | | | | | | | | |

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| 242 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 243 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 244 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 245 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 246 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 247 | 5 | 3 | 4 | 2 | 5 | 3 | 5 | 2 | 4 | 2 | 2 | 4 | 3 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 |
| 248 | 5 | 5 | 4 | 2 | 5 | 3 | 5 | 3 | 4 | 2 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 |
| 249 | 5 | 3 | 3 | 1 | 5 | 3 | 5 | 2 | 4 | 2 | 4 | 2 | 5 | 2 | 4 | 2 | 2 | 3 | 5 | 3 | 5 | 5 | 3 |
| 250 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |
| 251 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |
| 252 | 5 | 0.8 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 15 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 35 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 |
| 40 | 4 | 2 | 4 | 2 | 4 | 2 | 4 | 2 | 5 | 2 | 5 | 2 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 3 | 5 | 5 | 4 |
| 45 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.2 | 0.1 | 0.2 | 0.1 | 0.8 | 0.2 | 0.2 | 0.2 | 0.1 |
| 50 | 264 | 265 | 266 | 267 | 268 | 269 | 270 | 271 | 272 | 273 | 274 | | | | | | | | | | | | |
| 55 | | | | | | | | | | | | | | | | | | | | | | | |

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| 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 15 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 20 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 5 | 2 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 3 | 4 | 3 |
| 35 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 |
| 40 | 5 | 2 | 5 | 3 | 5 | 3 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 5 | 2 | 4 | 3 |
| 45 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.2 |
| 50 | | | | | | | | | | | | | | | | | | | | | | |
| 55 | 286 | 287 | 288 | 289 | 290 | 291 | 292 | 293 | 294 | 295 | 296 | | | | | | | | | | | |

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| 319 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 320 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 321 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 322 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 323 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 324 | 5 | 3 | 5 | 4 | 4 | 3 | 5 | 4 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 3 |
| 325 | 5 | 3 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 |
| 326 | 5 | 3 | 5 | 3 | 4 | 2 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 2 |
| 327 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.1 | 0.8 | 0.1 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 5 | 0.8 | 5 | 0.8 |
| 328 | | | | | | | | | | | | | | | | | | | | | | |
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| 5 | 5 | 4 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 10 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 15 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 20 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 25 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 30 | 3 | 2 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 3 | 5 | 4 | 5 | 3 | 5 | 4 | 4 |
| 35 | 3 | 1 | 5 | 5 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 3 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 40 | 3 | 1 | 5 | 3 | 5 | 3 | 5 | 4 | 5 | 3 | 5 | 4 | 5 | 4 | 4 | 2 | 5 | 4 | 5 | 4 | 5 | 4 | 4 |
| 45 | 0.8 | 0.2 | 0.2 | 0.1 | 0.2 | 0.1 | 0.2 | 0.1 | 0.2 | 0.1 | 0.2 | 0.1 | 0.2 | 0.1 | 0.8 | 0.2 | 0.8 | 0.2 | 0.8 | 0.2 | 0.2 | 0.2 | 0.1 |
| 50 | 341 | 342 | 343 | 344 | 345 | 346 | 347 | 348 | 349 | 350 | 351 | | | | | | | | | | | | |

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| 385 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 2 | 0 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 386 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 387 | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - | - |
| 388 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 389 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 390 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 391 | 5 | 4 | 5 | 4 | 5 | 4 | 4 | 3 | 4 | 4 | 4 | 3 | 4 | 2 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 3 | | |
| 392 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 | 5 |
| 393 | 5 | 2 | 5 | 2 | 5 | 3 | 5 | 4 | 5 | 4 | 5 | 4 | 5 | 3 | 5 | 2 | 5 | 3 | 5 | 3 | 5 | 3 | 5 | 2 |
| 394 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 0.8 | 0.1 | 0.8 | 0.2 | 0.8 | 0.1 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 | 5 | 0.8 |
| 395 | | | | | | | | | | | | | | | | | | | | | | | | |

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|---|---|---|---|---|---|---|---|
| A | 5 | 0 | 2 | 0 | 3 | 1 | 3 |
| B | 5 | 0 | 1 | 0 | 2 | 1 | 2 |
| C | 5 | 0 | 1 | 0 | 1 | 0 | 1 |
| D | 5 | 0 | 2 | 2 | 3 | 3 | 3 |

As shown in table 8, the present invented 3-(substituted phenyl)pyrazole derivatives or salt thereof, represented

by the general formula (I), exhibit more excellent controlling effect on weeds than comparative compound A, B, C or D on post-emergence treatment in upland fields. Even when the invented derivatives exhibit both herbicidal activity on weeds and phytotoxicity on crops at same dosage, by the selection of proper lower dosage, sufficient herbicidal activity is retained and chemical injury on crops is diminished remarkably.

Formulation Example 1

A wettable powder composition was prepared by uniform mixing and grinding of the following ingredients:

| | |
|---|----------|
| Compound No. 1 | 50 parts |
| Clay-white carbon mixture (clay is the major component) | 45 " |
| Polyoxyethylene nonylphenyl ether | 5 " |

Formulation Example 2

A granular composition was prepared by uniform mixing and grinding of the following ingredients, kneading the mixture with a suitable amount of water, and granulating the kneaded mixture.

| | |
|-------------------------|---------|
| Compound No. 7 | 5 parts |
| Bentonite-clay mixture | 90 " |
| Calcium ligninsulfonate | 5 " |

Formulation Example 3

An emulsifiable concentrate was prepared by uniform mixing of the following ingredients:

| | |
|--|----------|
| Compound No. 12 | 50 parts |
| Xylene | 40 " |
| Mixture of polyoxyethylene nonylphenyl ether and calcium alkylbenzenesulfonate | 10 " |

Formulation Example 4

A wettable powder composition was prepared by uniform mixing and grinding of the following ingredients:

| | |
|---|----------|
| Compound No. 33 | 50 parts |
| Clay-white carbon mixture (clay is the major component) | 45 " |
| Polyoxyethylene nonylphenyl ether | 5 " |

Formulation Example 5

A granular composition was prepared by uniform mixing and grinding of the following ingredients, kneading the mixture with a suitable amount of water, and granulating the kneaded mixture.

| | |
|-------------------------|---------|
| Compound No. 122 | 5 parts |
| Bentonite-clay mixture | 90 " |
| Calcium ligninsulfonate | 5 " |

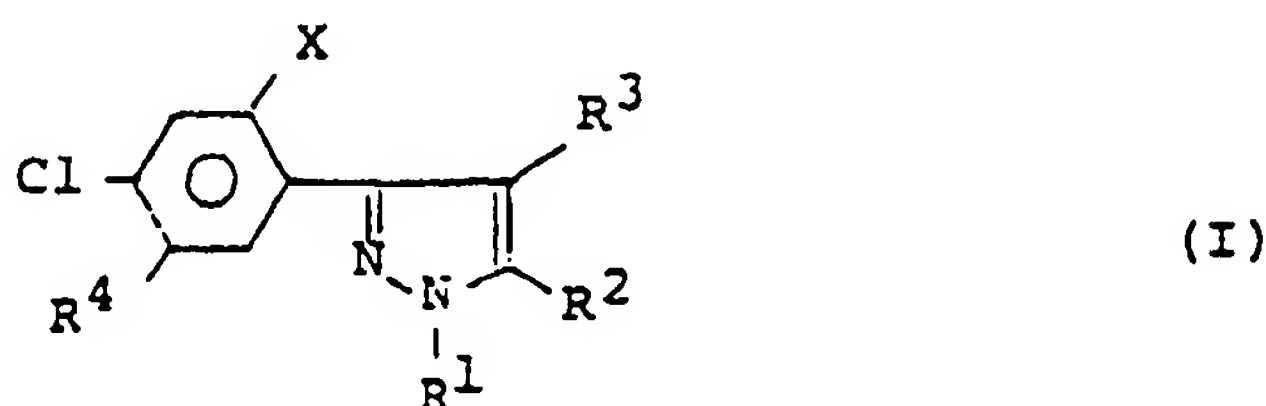
Formulation Example 6

An emulsifiable concentrate was prepared by uniform mixing the following ingredients:

| | |
|--|----------|
| Compound No. 381 | 50 parts |
| Xylene | 40 " |
| Mixture of polyoxyethylene nonylphenyl ether and calcium alkylbenzenesulfonate | 10 " |

Claims

1. A 3-(substituted phenyl)pyrazole derivative or a salt thereof, the derivative being represented by the general formula



wherein

X denotes halogen,

R¹ denotes CH₃,

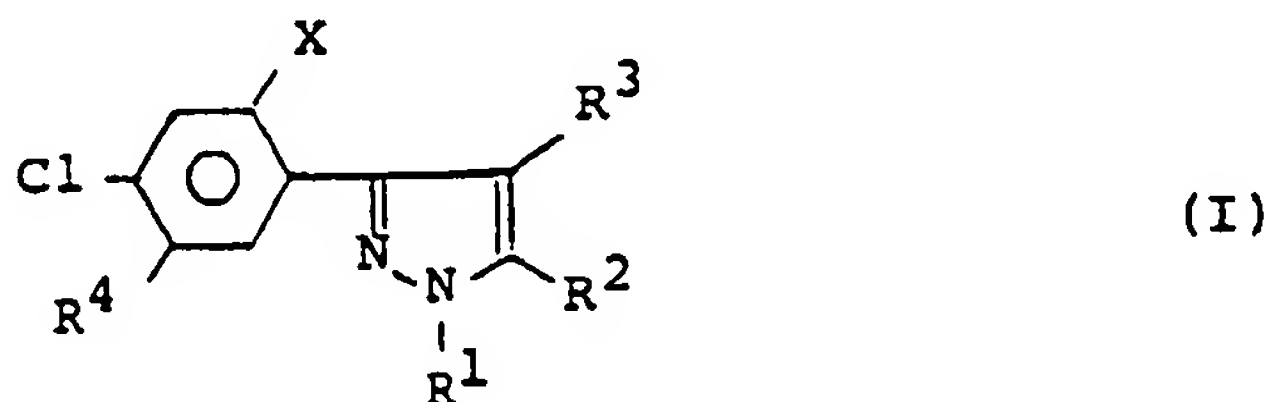
R² denotes hydroxy, mercapto, alkylthio, haloalkoxy or haloalkylthio,

R³ denotes hydrogen or halogen, and

R⁴ denotes (i) formyl, (ii) nitro, (iii) -CO-B-R⁶ wherein B denotes -O-, -S-, or -NR⁷, R⁶ and R⁷ are the same or different and each denote hydrogen or cycloalkyl, and when B is -O-, R⁶ can be an alkali metal atom or -CO-B-R⁶ denotes methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, methoxycarbonylmethoxycarbonyl, ethoxycarbonylmethoxycarbonyl, propoxycarbonylmethoxycarbonyl, methoxycarbonylethoxycarbonyl, ethoxycarbonylethoxycarbonyl, or propoxycarbonylethoxycarbonyl,

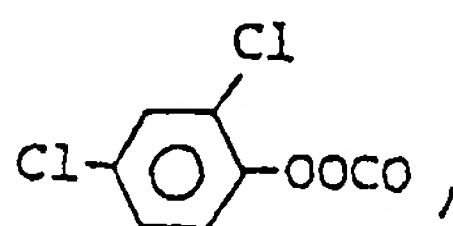
(iv) -D-R⁸ wherein D denotes -O-, -S(O)_n, n being an integer of 0 to 2, or -NR⁹-, R⁸ and R⁹, which are the same or different, each denotes hydrogen; alkyl; haloalkyl; aminosulfonyl; phenylalkyl or phenoxy-alkyl optionally having on the phenyl ring one or more halogen substituents which are the same or different; or -D-R⁸ denotes an alkenyloxy group containing propenyl, butenyl, or pentenyl; an alkynyloxy group containing propynyl, butynyl or pentynyl; an alkenylamino group containing propenyl, butenyl or pentenyl; an alkoxy-carbonylalkoxy or an alkylthiocarbonylalkoxy group containing each methyl, ethyl, propyl or butyl; or (v) -(CHR¹⁰)_m-CO-E-R¹² wherein E denotes -O-, -S-, or -NR¹¹, R¹⁰ denotes hydrogen and R¹¹ and R¹², which are the same or different, each denotes hydrogen; alkyl; haloalkyl; phenyl optionally having on the phenyl ring one or more halogen substituents; phenylalkyl optionally having on the phenyl ring one or more halogen substituents; or R¹¹, jointly with R¹², forms a piperidino or morpholinyl group, and when E is -O-, R¹² can be an alkali metal atom; and m denotes an integer of 0 to 3.

2. A 3-(substituted phenyl)-pyrazole derivative or a salt thereof, the derivative being represented by the general formula

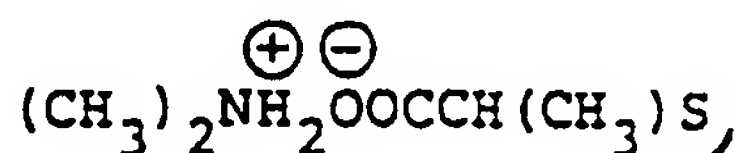
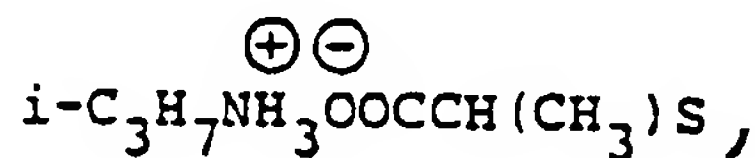


wherein

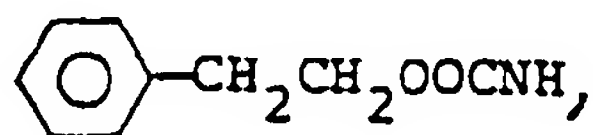
a) X is Cl, R¹ is CH₃, R² is CH₃S, R³ is Cl and R⁴ is selected from



HOOCCH(CH₃)O, CH₃OOCCH(CH₃)O, C₂H₅OOCCH(CH₃)O, i-C₃H₇OOCCH(CH₃)O, CH₂=CHCH₂OOCCH(CH₃)O, CH≡CCH₂OOCCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(Cl)CH₂S, CH≡CCH₂S, HO₂CCH(CH₃)S,



Cl(CH₂)₃OOCCH(CH₃)S, CH₂=CHCH₂OOCCH(CH₃)S, CH≡CCH₂OOCCH(CH₃)S, C₂H₅CH(CH₃)CH₂OOCCH(CH₃)S, (C₂H₅)₂NCOCH(CH₃)S, CH₃NHSO₂, (CH₃)₂NSO₂, n-C₃H₇NHSO₂, CH₃SO₂NHSO₂, Cl⁻H₃N⁺, CH≡CCH₂NH, CH₃OOCNH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃NHSO₂N(CH₃), (CH₃)₂NSO₂N(CH₃), i-C₃H₇NHSO₂NH, CH≡CCH₂N(CH₃)SO₂N(CH≡CCH₂), CH₃OCH₂CH₂OOCNH, BrCH₂CH₂OOCNH, NCCH₂CH₂OOCNH, CH₂=C(CH₃)CH₂OOCNH,



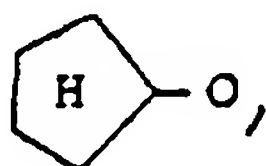
CH₂=CHCH₂OOCCH₂NH, CH≡CCH₂OOCCH₂NH, C₂H₅OOCCH(CH₃)NH, HOOCCH₂NH, CH₃OOCCH₂NH, HOOCCH(CH₃)NH and CH₃OOCCH(CH₃)NH;

b) X is Cl, R¹ is CH₃, R² is CH₃SO, R³ is Cl and R⁴ is selected from NO₂, CH₃SO₂NH, CH₃NHSO₂NH, i-C₃H₇NHSO₂NH and CH₃NHSO₂(CH≡CCH₂);

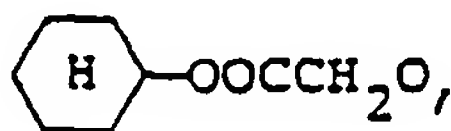
c) X is Cl, R¹ is CHF₂, R² is CH₃S, R³ is Cl and R⁴ is selected from NH₂, CH₃NHSO₂NH, CH₃NHSO₂N(CH≡CCH₂) and CH₃N(CH≡CCH₂)SO₂N(CH≡CCH₂);

d) X is Cl, R¹ is CHF₂, R² is CH₃SO, R³ is Cl and R⁴ is selected from NH₂ and NO₂;

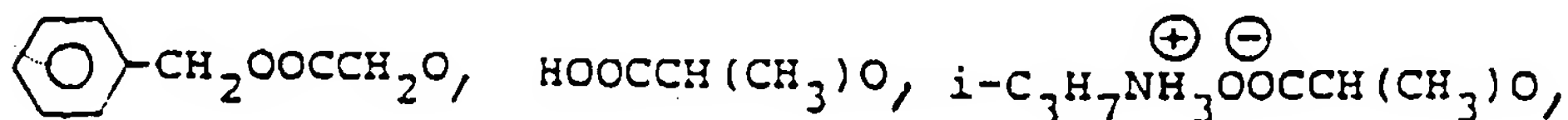
e) X is Cl, R¹ is CH₃, R² is CHF₂O, R³ is Cl and R⁴ is selected from



C₂H₅OCH₂CH₂O, NCCH₂O, CH₃OOCO, n-C₆H₁₃OOCCH₂O, CH₂=CHCH₂OOCCH₂O, CH≡CCH₂OOCCH₂O,

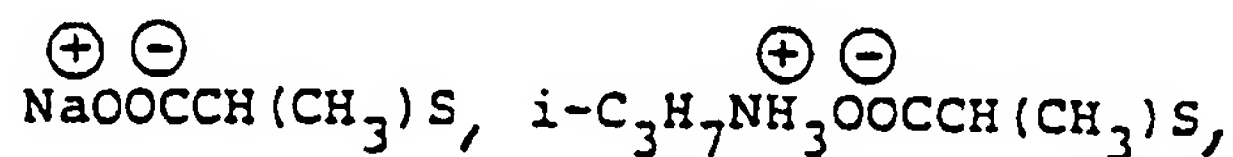


C₂H₅O(CH₂)₂OOCCH₂O, C₂H₅O(CH₂)₂O(CH₂)₂OOCCH₂O,



CH₂=CHCH₂OOCCH(CH₃)O, CH≡CCH₂OOCCH(CH₃)O, CH₃NHCOCH(CH₃)O, (CH₃)₂NCOCH(CH₃)O,
CH₂=CHCH₂S, CH₂=CHCH₂SO, CH≡CCH₂S, (C₂H₅O)₂P(O)CH₂S, CH₃OOCs, CH₃OOCCH₂S,
C₂H₅OOCCH₂S, CH₂=CHCH₂OOCCH₂S, HOOCCH(CH₃)S,

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CH₃OOCCH(CH₃)S, C₂H₅OOCCH(CH₃)S, C₂H₅OOCCH(CH₃)S(O), C₂H₅OOCCH(CH₃)SO₂, i-C₃H₇OOCCH
(CH₃)S, n-C₄H₉OOCCH(CH₃)S, (CH₃)₂CH(CH₂)₂OOCCH(CH₃)S, CH₂=CHCH₂OOCCH(CH₃)S, CH≡
CCH₂OOCCH(CH₃)S, CH₃O(CH₂)₂O(CH₂)₂OOCCH(CH₃)S, (C₂H₅O)₂P(O)CH₂OOCCH(CH₃)S,
CH₃NHCOCH(CH₃)S, (CH₃)₂NCOCH(CH₃)S, (C₂H₅)₂NCOCH(CH₃)S, i-C₃H₇NHCOCH(CH₃)S,

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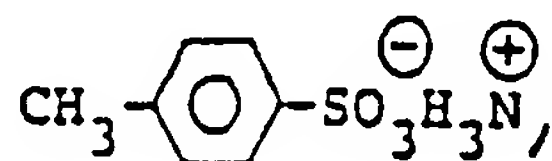
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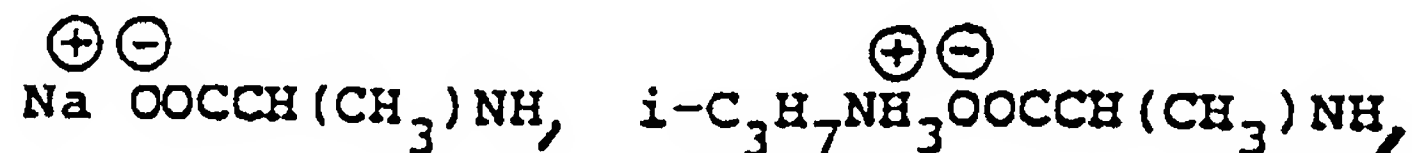
CH₃SO₂NH, CH₃NHSO₂NH, CH₂=CHCH₂NH, CH≡CCH₂NH, CH₃OOCNH, Cl[⊖]H₃N[⊕],

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CH₃OOCCH₂NH, HOOCCH(CH₃)NH,

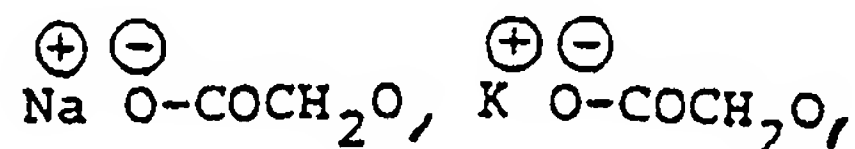
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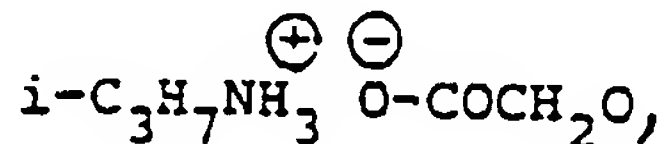
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C₂H₅OOCCH(CH₃)N, CH₃NHOCCH(CH₃)NH, (CH₃)₂NOCCH(CH₃)NH, CH₃NHSO₂, (CH₃)₂NSO₂,
(C₂H₅)₂NCOS, (CH₃)₃SiCH₂S, (CH₃)₃SiCH₂O, HO-COCH₂O,

45



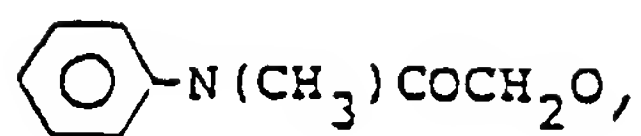
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CH₂=CHC(CH₃)₂O-COCH₂O, CH₃C≡CCH₂O-COCH₂O, CH₃NHCOCH₂O, (CH₃)₂NCOCH₂O, n-
C₄H₉NHCOCH₂O, H₂C=CHCH₂NHCOCH₂O, (H₂C=CHCH₂)₂NCOCH₂O,

55

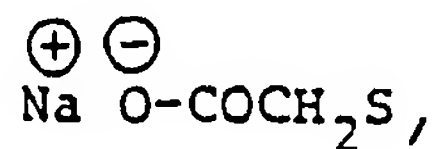




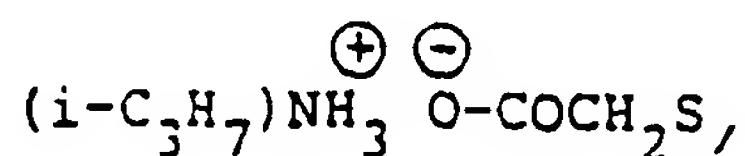
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$\text{CH}_3\text{SO}_2\text{NHCOCH}_2\text{O}$, $\text{CH}_3\text{SO}_2\text{NHCOCH}(\text{CH}_3)\text{O}$, $(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{S}$, $\text{CH}_2=\text{C}(\text{Cl})\text{CH}_2\text{S}$, $\text{ClCH}=\text{CHCH}_2\text{S}$, $\text{ClCH}=\text{C}(\text{Cl})\text{CH}_2\text{S}$, NCCH_2S ,

10



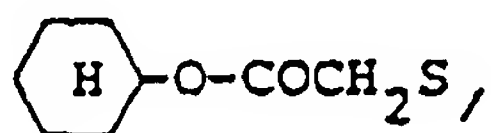
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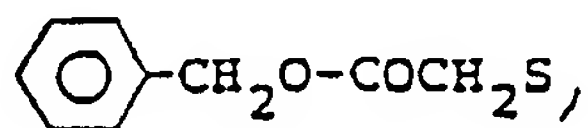
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$\text{HO-COCH}_2\text{S}$, $\text{i-C}_3\text{H}_7\text{O-COCH}_2\text{S}$, $\text{n-C}_4\text{H}_9\text{O-COCH}_2\text{S}$, $\text{C}_{12}\text{H}_{25}\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{S-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{S-COCH}_2\text{S}$, $\text{i-C}_3\text{H}_7\text{S-COCH}_2\text{S}$, $\text{CH}\equiv\text{CCH}_2\text{O-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}_2\text{S}$,

25

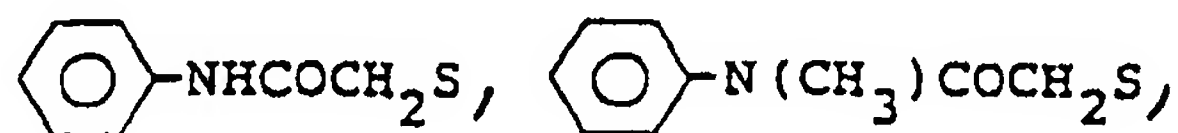


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$(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{NHCOCH}_2\text{S}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{S}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{S}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{S}$, $\text{HC}\equiv\text{CCH}_2\text{NHCOCH}_2\text{S}$,

35



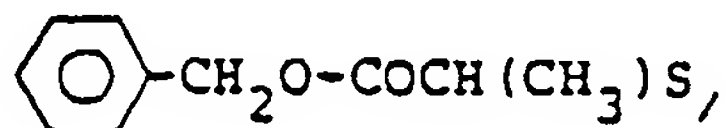
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$\text{i-C}_4\text{H}_9\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{t-C}_4\text{H}_9\text{CH}_2\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{n-C}_6\text{H}_{13}\text{O-COCH}(\text{CH}_3)\text{S}$,

50



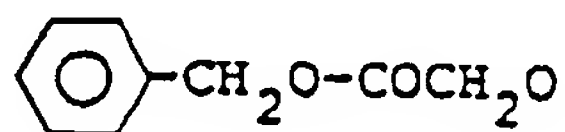
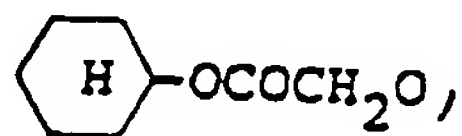
$\text{C}_2\text{H}_5\text{S-COCH}(\text{CH}_3)\text{S}$, $\text{i-C}_3\text{H}_7\text{S-COCH}(\text{CH}_3)\text{S}$, $\text{NCCH}(\text{CH}_3)\text{O-COCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{SC}_2\text{H}_4\text{O-COCH}(\text{CH}_3)\text{S}$,

55



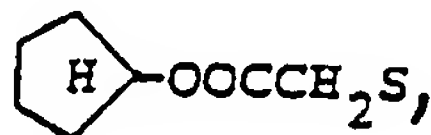


(CH₃)₃SiCH₂O-COCH(CH₃)S, CH₂=CHCH₂NHCOCH(CH₃)S, (H₂C=CHCH₂)₂NCOCH(CH₃)S, HC≡CCH₂NHCOCH(CH₃)S, CH₃CO-O-(CH₂)₂S, C₂H₅O-COC(CH₃)₂S, CH₃O-COCH(i-C₃H₇)S, C₂H₅O-COCH(CH₃)N(CH₃), CH₂=CHCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, CH≡CCH₂O-CO,

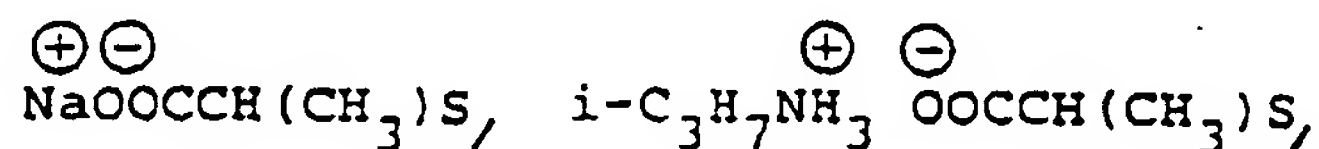


and CH₃O-CO-OCH₂S-CO;

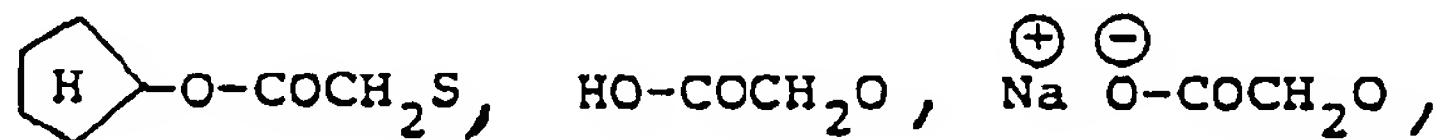
I) X is Cl, R¹ is CH₃, R² is CHF₂S, R³ is Cl and R⁴ is selected from CH₂=CHCH₂S, CH≡CCH₂S, C₂H₅OCH₂CH₂S, CH₃SCH₂S, NCCH₂S, C₂H₅OOCs, CH₃OOCCH₂S, C₂H₅OOCCH₂S,



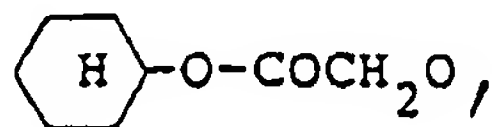
CH₃NHOCCH₂S, (CH₃)₂NOCCH₂S, HOOCCH(CH₃)S,



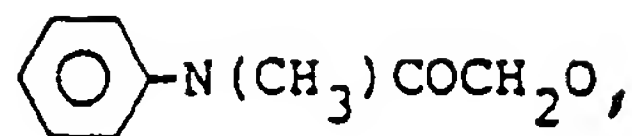
C₂H₅OOCCH(CH₃)S, n-C₃H₇OOCCH(CH₃)S, i-C₃H₇OOCCH₂S, (CH₃)₂CHCH₂OOCCH₂S, CH₂=CHCH₂OOCCH(CH₃)S, C₂H₅OOC(CH₂)₃S, CH₃NHOCCH(CH₃)S, (CH₃)₂NOCCH(CH₃)S, CH≡CCH₂NH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃OOCNH, (C₂H₅)₂NCOS, NCCH₂O, ClCH=CHCH₂O, CH₃OCH₂O, C₂H₅OC₂H₄O, CH₃OCH₂S, ClCH=CHCH₂S, CH₂=C(Cl)CH₂S,



CF₃CH₂O-COCH₂O, CH₂=CHCH₂O-COCH₂O,



CH₃NHCOCH₂O, CH₂=CHCH₂O-COCH(CH₃)O, CH₃NHCOCH(CH₃)O, NCCH₂O-COCH₂O, CH₃O-COCH(CH₃)S, NCCH₂O-COCH(CH₃)S, C₂H₅O-CO(CH₂)₃S,



HO-COCH₂NH, C₂H₅O-COCH₂NH and C₂H₅O-COCH(CH₃)NH;

g) X is F, R¹ is CH₃, R² is CH₃SO, R³ is Cl and R⁴ is 5-NO₂;

h) X is F, R¹ is CH₃, R² is CH₃S, R³ is Cl and R⁴ is selected from 5-NH₂, 5-SH, 5-CH≡CCH₂O, 5-CH≡CCH₂S and 5-C₂H₅OOCCH(CH₃)S;

i) X is F, R¹ is CH₃, R² is CHF₂O, R³ is Cl and R⁴ is selected from HO-COCH₂O, CH₃O-COCH₂O, CH₂-CHCH₂O-COCH₂O, CH≡CCH₂O-COCH₂O, HO-COCH(CH₃)O, CH₂=CHCH₂O-COCH(CH₃)O, CH≡CCH₂O-COCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(CH₃)CH₂S, CH≡CCH₂S, C₂H₅OC₂H₄S, HO-COCH₂S, CH₃O-COCH₂S, C₂H₅O-COCH₂S, C₂H₅O-COCH₂S(O), n-C₄H₉O-COCH₂S, CH₂=CHCH₂O-COCH₂S, CH₃SCH₂O-COCH₂S,

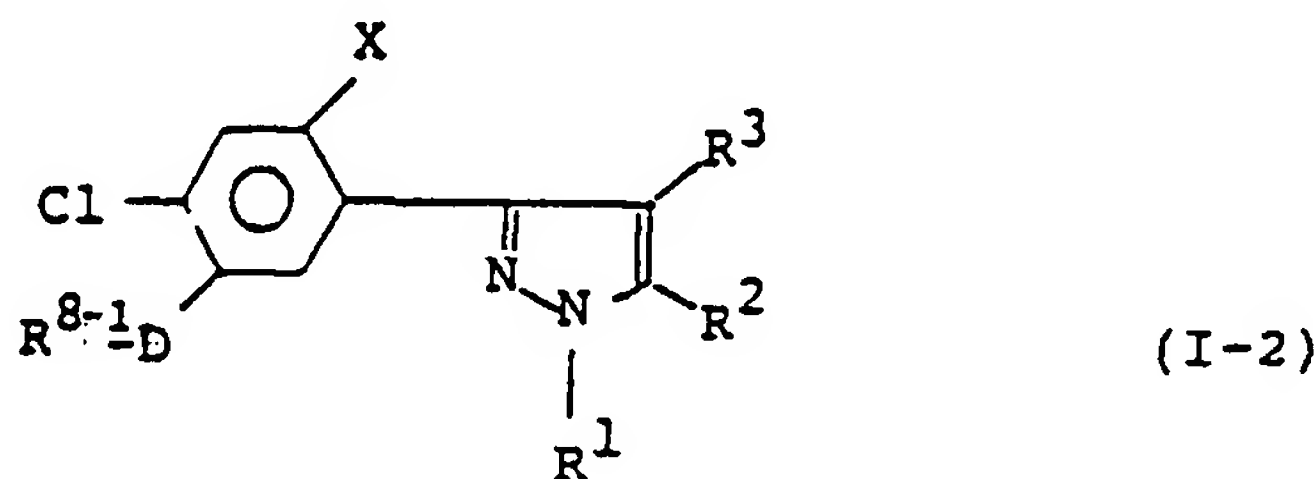


CH₃NHCOCH₂S, (CH₃)₂NCOCH₂S, HO-COCH(CH₃)S, CH₃O-COCH(CH₃)S, C₂H₅O-COCH(CH₃)S, n-C₄H₉O-COCH(CH₃)S, CH₂=CHCH₂O-COCH(CH₃)S, CH≡CH₂O-COCH(CH₃)S, CH₃OC₂H₄OC₂H₄O-COCH(CH₃)S, CH₃SO₂NH, CH₃NHSO₂NH, C₂H₅O-COCH₂NH, C₂H₅O-COCH(CH₃)NH, CH₂=CHCH₂O-CO, CH≡CCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, (CH₃)₂NCO and



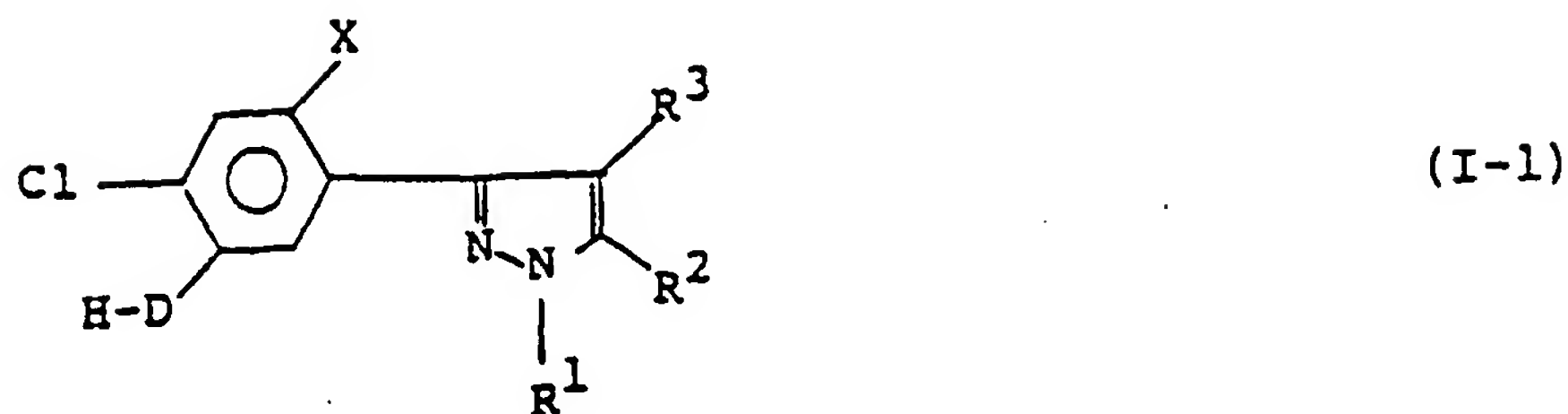
j) X is Cl, R¹ is CH₃, R² is CHF₂O, R³ is Br and R⁴ is BrCH₂CH(Br)CH₂O or BrC≡CCH₂O.

3. A process for producing 3-(substituted phenyl)-pyrazole derivatives represented by the general formula



wherein X, R¹, R² and R³ are as defined in claim 1, and D-R⁸⁻¹ is defined as D-R⁸ in claim 1 except that R⁸⁻¹ is not hydrogen,

the process comprising reacting a pyrazole derivative represented by the general formula

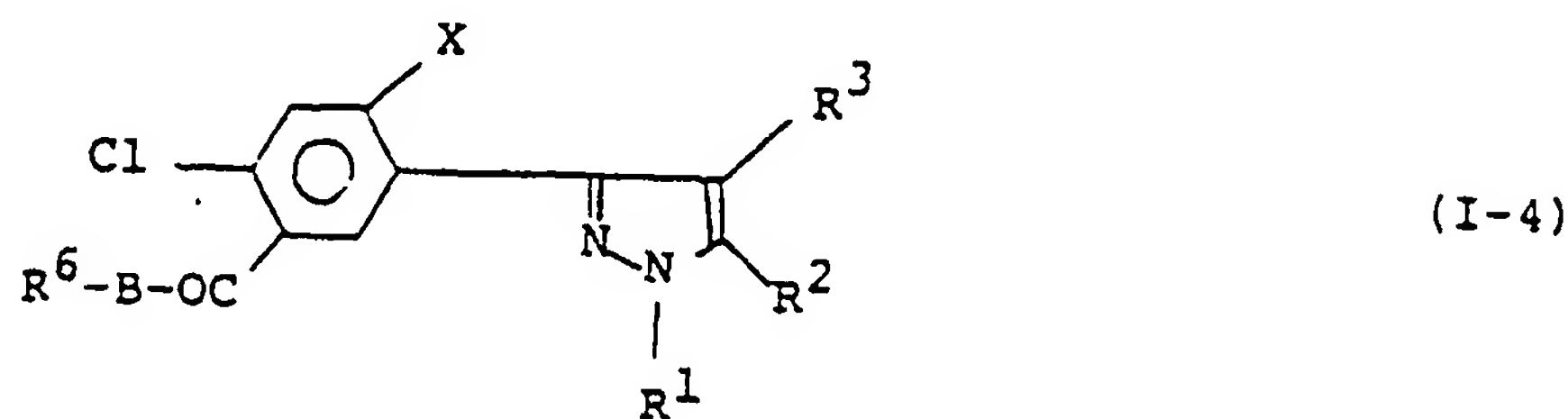


10 wherein X, R¹, R², R³, and D are as defined above, with a halide represented by the general formula

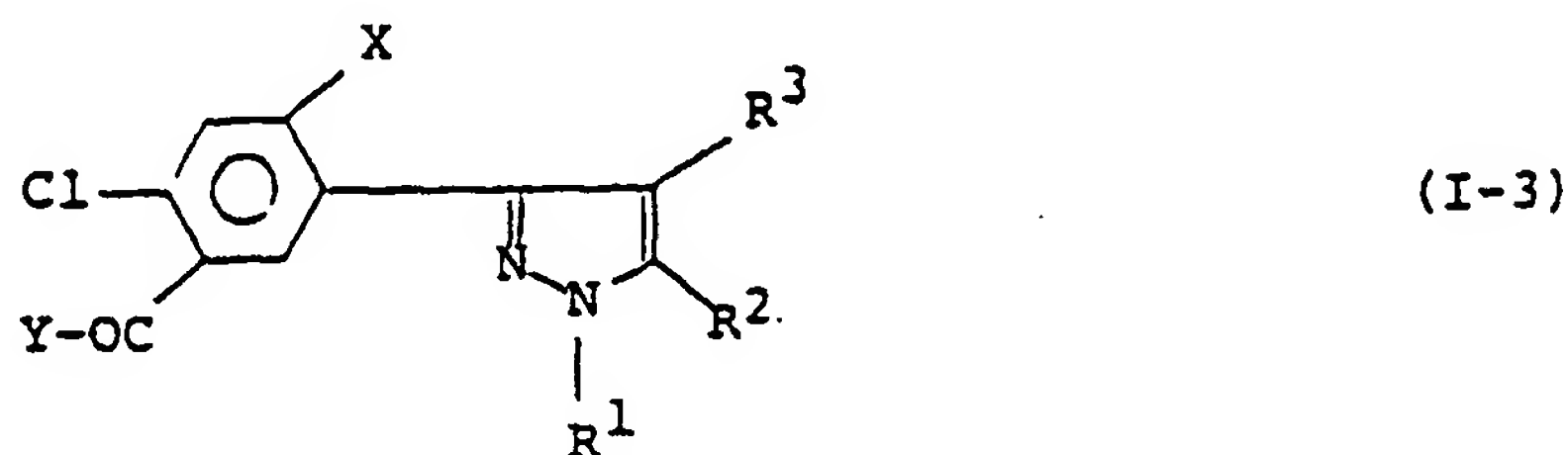


wherein R⁸⁻¹ is defined above and Z is halogen.

4. A process for producing 3-(substituted phenyl)pyrazole derivatives represented by the general formula



30 wherein X, R¹, R², R³ and -CO-B-R⁶ are as defined in claim 1, the process comprising reacting a pyrazole derivative represented by the general formula



40 wherein X, R¹, R² and R³ are as defined above and Y denotes halogen or OH, with a compound represented by the general formula



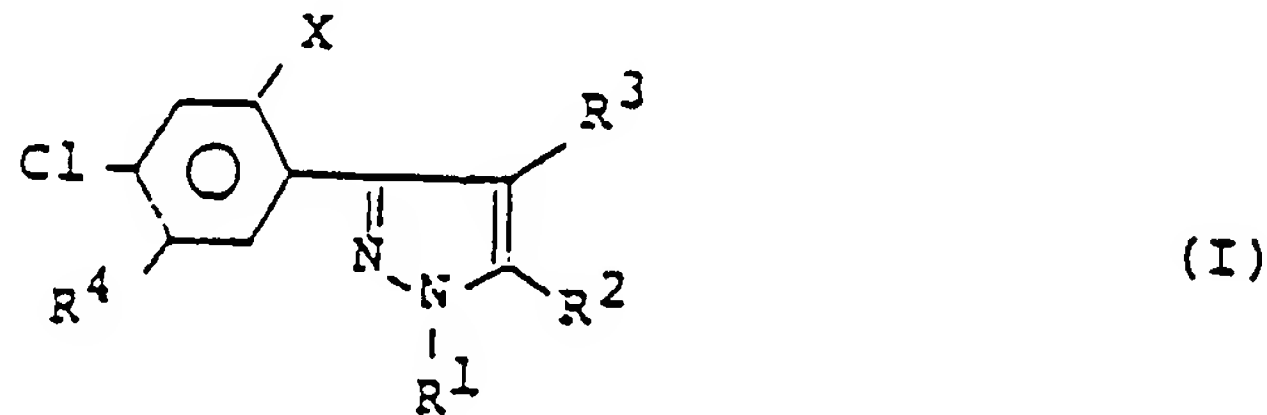
50 wherein R⁶ and B are as defined above.

- 55
5. A herbicidal composition which contains a 3-(substituted phenyl)pyrazole derivative or a salt thereof according to claim 1 or 2 as an active ingredient.
 6. Use of the herbicidal composition of claim 5, for paddy fields, upland fields, or non-farming area purposes.

7. A method for controlling undesirable plants which comprises applying a herbicidal composition containing a 3-(substituted phenyl)pyrazole derivative or a salt thereof according to claim 1 or 2 as an active ingredient, in a dose of 1.0 g to 10 Kg in terms of the quantity of active ingredient per hectare.

Patentansprüche

1. 3-(substituiertes Phenyl)pyrazol-Derivat oder ein Salz davon, wobei das Derivat dargestellt wird durch die allgemeine Formel (I)



worin

X Halogen bedeutet,

R¹ CH₃ bedeutet,

R² Hydroxy, Mercapto, Alkylthio, Haloalkoxy oder Haloalkylthio bedeutet,

R³ Wasserstoff oder Halogen bedeutet, und

R⁴

(i) Formyl bedeutet, (ii) Nitro bedeutet,

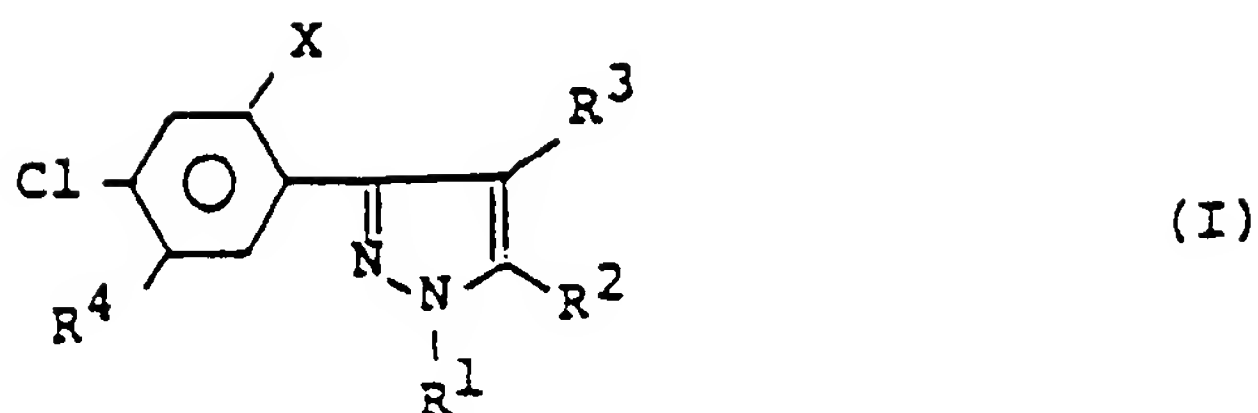
(iii) -CO-B-R⁶ bedeutet, worin B -O-, -S- oder -NR⁷- bedeutet, wobei R⁶ und R⁷ gleich oder verschieden sind und jeweils Wasserstoff oder Cycloalkyl bedeuten, und wenn B -O- ist, R⁶ ein Alkalimetallatom sein kann oder -CO-B-R⁶ Methoxycarbonyl, Ethoxycarbonyl, Propoxycarbonyl, Butoxycarbonyl, Methoxycarbonylmethoxycarbonyl, Ethoxycarbonylmethoxycarbonyl, Propoxycarbonylmethoxycarbonyl, Methoxycarbonylethoxycarbonyl, Ethoxycarbonylethoxycarbonyl oder Propoxycarbonylethoxycarbonyl bedeutet,

(iv) -D-R⁸ bedeutet, worin D -O-, -S(O)_n-, worin n eine ganze Zahl von 0 bis 2 ist, oder -NR⁹- bedeutet, worin R⁸ und R⁹, die gleich oder verschieden sind, jeweils Wasserstoff; Alkyl; Haloalkyl; Aminosulfonyl; Phenylalkyl oder Phenoxyalkyl bedeuten, worin der Phenylring gegebenenfalls einen oder mehrere Halogensubstituenten haben kann, die gleich oder verschieden sein können; oder -D-R⁸ eine Alkenyloxygruppe, enthaltend Propenyl, Butenyl oder Pentenyl; eine Alkinyloxygruppe, enthaltend Propinyl, Butinyl oder Pentinyl; eine Alkenylaminogruppe, enthaltend Propenyl, Butenyl oder Pentenyl; eine Alkoxy-carbonylalkoxy- oder eine Alkylthiocarbonylalkoxygruppe, enthaltend jeweils Methyl, Ethyl, Propyl oder Butyl bedeutet; oder

(v) -(CHR¹⁰)_m-CO-E-R¹² bedeutet, worin E -O-, -S- oder -NR¹¹ bedeutet; R¹⁰ Wasserstoff bedeutet und R¹¹ und R¹², die gleich oder verschieden sein können, jeweils Wasserstoff; Alkyl; Haloalkyl; Phenyl, gegebenenfalls mit einem oder mehreren Halogensubstituenten an dem Phenylring; Phenylalkyl, gegebenenfalls mit einem oder mehreren Halogensubstituenten am Phenylring, bedeutet; oder R¹¹, verbunden mit R¹², eine Piperidin- oder Morpholinylgruppe bedeutet; und wenn E -O- ist, R¹² ein Alkalimetallatom sein kann; und

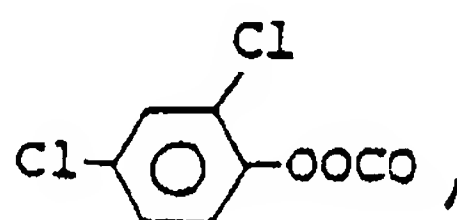
m eine ganze Zahl von 0 bis 3 bedeutet.

2. 3-(substituiertes Phenyl)pyrazol-Derivat oder ein Salz davon, wobei das Derivat dargestellt wird durch die allgemeine Formel (I)

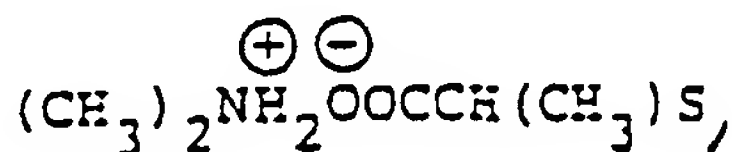
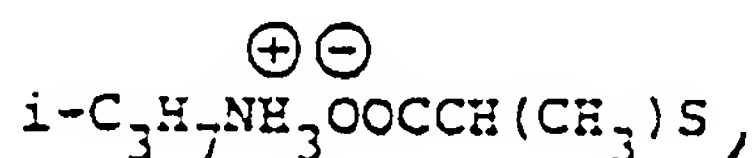


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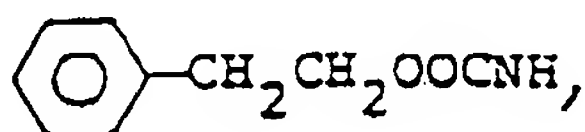
a) X Cl ist, R¹ CH₃ ist, R² CH₃S ist, R³ Cl ist und R⁴ gewählt wird aus



20 HOOCCH(CH₃)O, CH₃OOCCH(CH₃)O, C₂H₅OOCCH(CH₃)O, i-C₃H₇OOCCH(CH₃)O, CH₂=CHCH₂OOCCH(CH₃)O, CH≡CCH₂OOCCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(Cl)CH₂S, CH≡CCH₂S, HO₂CCH(CH₃)S,



35 Cl(CH₂)₃OOCCH(CH₃)S, CH₂=CHCH₂OOCCH(CH₃)S, CH≡CCH₂OOCCH(CH₃)S, C₂H₅CH(CH₃)CH₂OOCCH(CH₃)S, (C₂H₅)₂NCOCH(CH₃)S, CH₃NHSO₂, (CH₃)₂NSO₂, n-C₃H₇NHSO₂, CH₃SO₂NHSO₂, Cl⁻H₃N⁺, CH≡CCH₂NH, CH₃OOCNH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃NHSO₂N(CH₃), (CH₃)₂NSO₂N(CH₃), i-C₃H₇NHSO₂NH, CH≡CCH₂N(CH₃)SO₂N(CH≡CCH₂), CH₃OCH₂CH₂OOCNH, BrCH₂CH₂OOCNH, NCCH₂CH₂OOCNH, CH₂=C(CH₃)CH₂OOCNH,



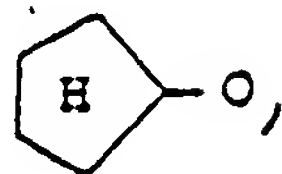
45 CH₂=CHCH₂OOCCH₂NH, CH≡CCH₂OOCCH₂NH, C₂H₅OOCCH(CH₃)NH, HOOCCH₂NH, CH₃OOCCH₂NH, HOOCCH(CH₃)NH und CH₃OOCCH(CH₃)NH;

b) X Cl ist, R¹ CH₃ ist, R² CH₃SO ist, R³ Cl ist und R⁴ gewählt wird aus NO₂, CH₃SO₂NH, CH₃NHSO₂NH, i-C₃H₇NHSO₂NH und CH₃NHSO₂(CH≡CCH₂);

c) X Cl ist, R¹ CHF₂ ist, R² CH₃S ist, R³ Cl ist und R⁴ gewählt wird aus NH₂, CH₃NHSO₂NH, CH₃NHSO₂N(CH≡CCH₂) und CH₃N(CH≡CCH₂)SO₂N(CH≡CCH₂);

d) X Cl ist, R¹ CHF₂ ist, R² CH₃SO ist, R³ Cl ist und R⁴ gewählt wird aus NH₂ und NO₂;

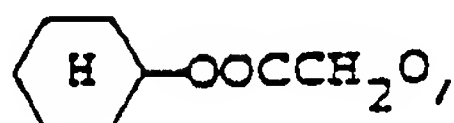
e) X Cl ist, R¹ CH₃ ist, R² CHF₂O ist, R³ Cl ist und R⁴ gewählt wird aus



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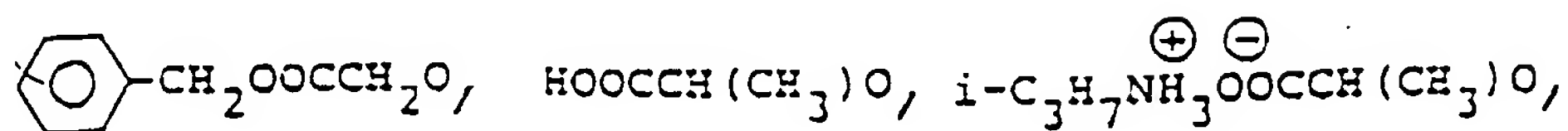
$C_2H_5OCH_2CH_2O$, $NCCH_2O$, CH_3OOCO , $n-C_6H_{13}OOCCH_2O$, $CH_2=CHCH_2OOCCH_2O$, $CH\equiv$
 CCH_2OOCCH_2O ,

10



$C_2H_5O(CH_2)_2OOCCH_2O$, $C_2H_5O(CH_2)_2O(CH_2)_2OOCCH_2O$,

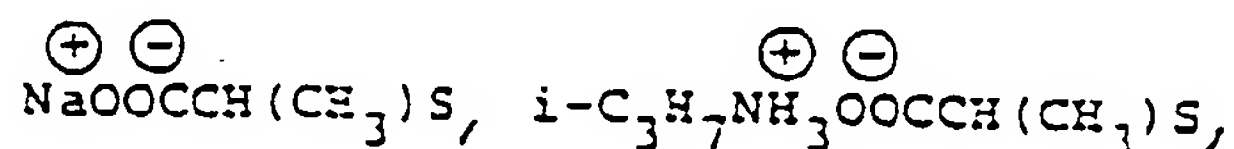
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$CH_2=CHCH_2OOCCH(CH_3)O$, $CH\equiv CCH_2OOCCH(CH_3)O$, $CH_3NHCOCH(CH_3)O$, $(CH_3)_2NCOCH(CH_3)O$,
 $CH_2=CHCH_2S$, $CH_2=CHCH_2SO$, $CH\equiv CCH_2S$, $(C_2H_5O)_2P(O)CH_2S$, CH_3OOC , CH_3OOCCH_2S ,
 $C_2H_5OOCCH_2S$, $CH_2=CHCH_2OOCCH_2S$, $HOOCCH(CH_3)S$,

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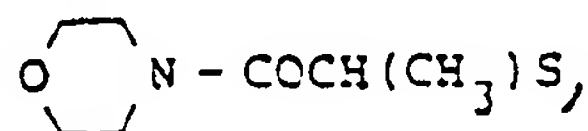
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$CH_3OOCCH(CH_3)S$, $C_2H_5OOCCH(CH_3)S$, $C_2H_5OOCCH(CH_3)S(O)$, $C_2H_5OOCCH(CH_3)SO_2$, $i-C_3H_7OOCCH$
 $(CH_3)S$, $n-C_4H_9OOCCH(CH_3)S$, $(CH_3)_2CH(CH_2)_2OOCCH(CH_3)S$, $CH_2=CHCH_2OOCCH(CH_3)S$, $CH\equiv$
 $CCH_2OOCCH(CH_3)S$, $CH_3O(CH_2)_2O(CH_2)_2OOCCH(CH_3)S$, $(C_2H_5O)_2P(O)CH_2OOCCH(CH_3)S$, CH_3
 $NHCOCH(CH_3)S$, $(CH_3)_2NCOCH(CH_3)S$, $(C_2H_5)_2NCOCH(CH_3)S$, $i-C_3H_7NHCOCH(CH_3)S$,

35

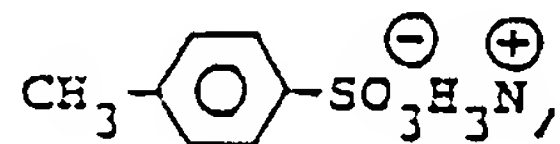


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CH_3SO_2NH , CH_3NHSO_2NH , $CH_2=CHCH_2NH$, $CH\equiv CCH_2NH$, CH_3OOCNH , $CH_3H_3N^{\oplus}$,

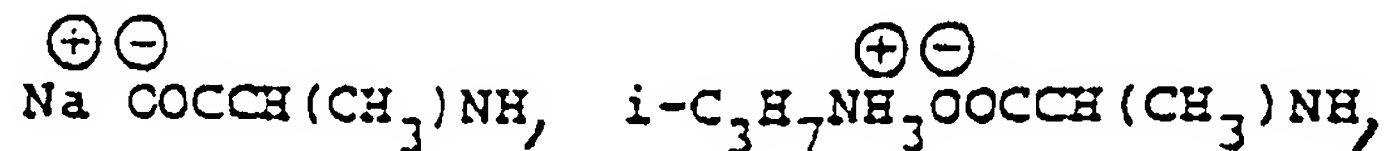
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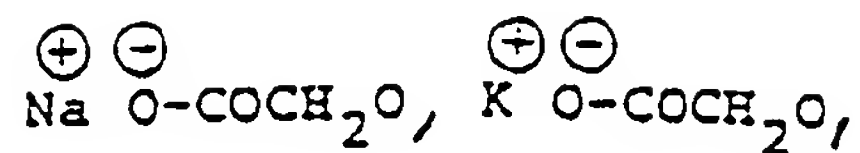
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CH_3OOCCH_2NH , $HOOCCH(CH_3)NH$,

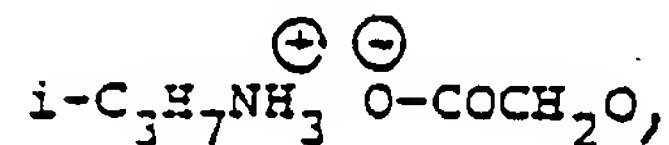
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$C_2H_5OOCCH(CH_3)NH$, $CH_3NHOCCH(CH_3)NH$, $(CH_3)_2NOCCH(CH_3)NH$, CH_3NHSO_2 , $(CH_3)_2NSO_2$,
 $(C_2H_5)_2NCOS$, $(CH_3)_3SiCH_2S$, $(CH_3)_3SiCH_2O$, $HO-COCH_2O$,



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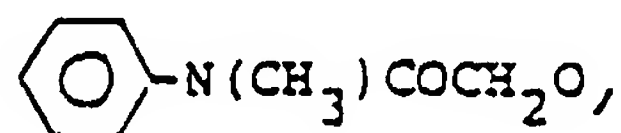
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$\text{CH}_2=\text{CHC}(\text{CH}_3)_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{C}\equiv\text{CCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{NHCOCH}_2\text{O}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{O}$, $\text{n-C}_4\text{H}_9\text{NHCOCH}_2\text{O}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{O}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{O}$,

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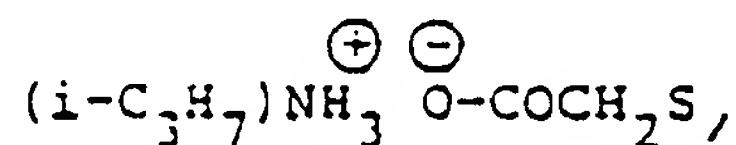
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$\text{CH}_3\text{SO}_2\text{NHCOCH}_2\text{O}$, $\text{CH}_3\text{SO}_2\text{NHCOCH}(\text{CH}_3)\text{O}$, $(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{S}$, $\text{CH}_2=\text{C}(\text{Cl})\text{CH}_2\text{S}$, $\text{ClCH}=\text{CHCH}_2\text{S}$, $\text{ClCH}=\text{C}(\text{Cl})\text{CH}_2\text{S}$, NCCH_2S ,

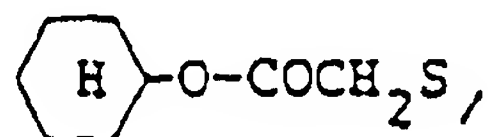
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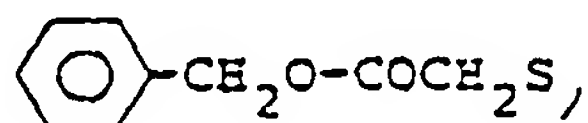
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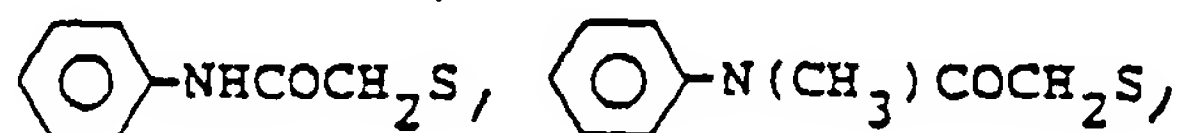
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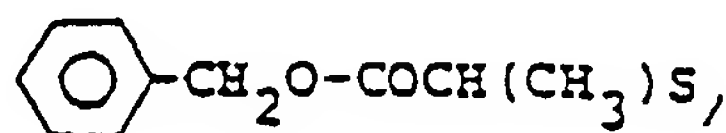
$(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{NHCOCH}_2\text{S}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{S}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{S}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{S}$, $\text{HC}\equiv\text{CCH}_2\text{NHCOCH}_2\text{S}$,

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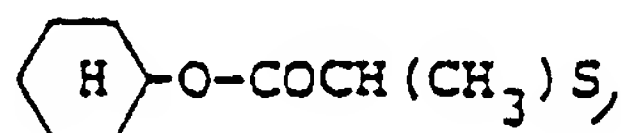




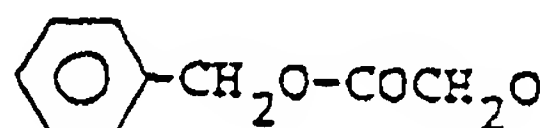
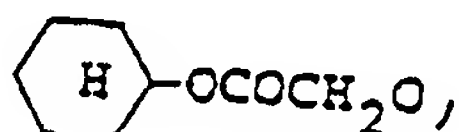
i-C₄H₉O-COCH(CH₃)S, t-C₄H₉CH₂O-COCH(CH₃)S, n-C₆H₁₃O-COCH(CH₃)S,



C₂H₅S-COCH(CH₃)S, i-C₃H₇S-COCH(CH₃)S, NCCH(CH₃)O-COCH(CH₃)S, CH₃SC₂H₄O-COCH(CH₃)S,

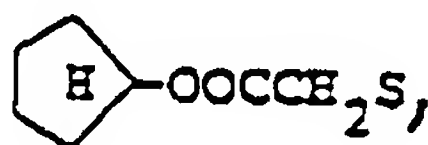


(CH₃)₃SiCH₂O-COCH(CH₃)S, CH₂=CHCH₂NHCOCH(CH₃)S, (H₂C=CHCH₂)₂NCOCH(CH₃)S, HC≡CCH₂NHCOCH(CH₃)S, CH₃CO-O-(CH₂)₂S, C₂H₅O-COC(CH₃)₂S, CH₃O-COCH(i-C₃H₇)S, C₂H₅O-COCH(CH₃)N(CH₃), CH₂=CHCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, CH≡CCH₂O-CO,

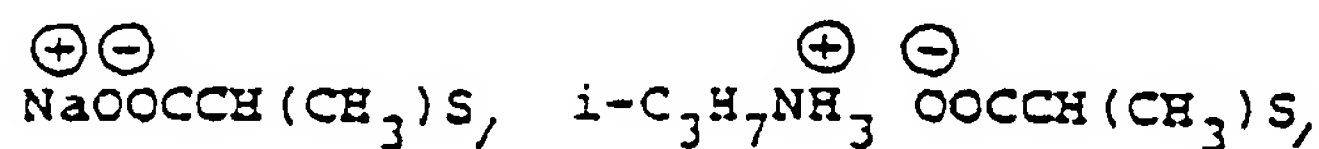


und CH₃O-CO-OCH₂S-CO;

f) X Cl ist, R¹ CH₃ ist, R² CHF₂S ist, R³ Cl ist und R⁴ gewählt wird aus CH₂=CHCH₂S, CH≡CCH₂S, C₂H₅OCH₂CH₂S, CH₃SCH₂S, NCCH₂S, C₂H₅OOCs, CH₃OOCCH₂S, C₂H₅OOCCH₂S,



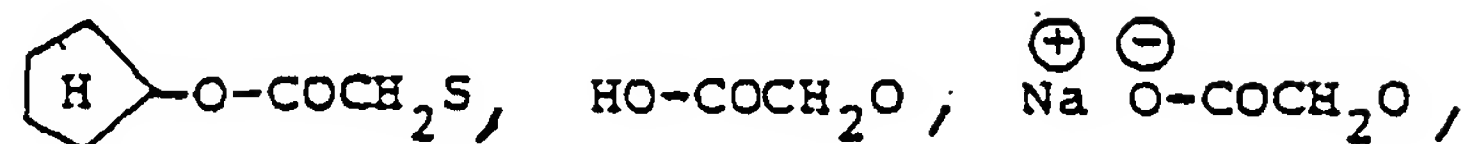
CH₃NHOCCH₂S, (CH₃)₂NOCCH₂S, HOOCCH(CH₃)S,



C₂H₅OOCCH(CH₃)S, n-C₃H₇OOCCH(CH₃)S, i-C₃H₇OOCCH₂S, (CH₃)₂CHCH₂OOCCH₂S, CH₂=CHCH₂OOCCH(CH₃)S, C₂H₅OOC(CH₂)₃S, CH₃NHOCCH(CH₃)S, (CH₃)₂NOCCH(CH₃)S, CH≡CCH₂NH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃OOCNH, (C₂H₅)₂NCOS, NCCH₂O, ClCH=CHCH₂O, CH₃OCH₂O,

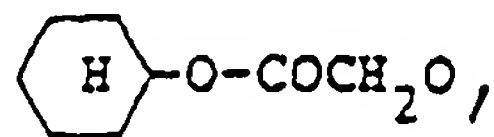
$C_2H_5OC_2H_4O$, CH_3OCH_2S , $ClCH=CHCH_2S$, $CH_2=C(Cl)CH_2S$,

5



$CF_3CH_2O-COCH_2O$, $CH_2=CHCH_2O-COCH_2O$,

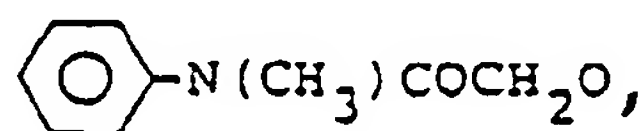
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$CH_3NHCOCH_2O$, $CH_2=CHCH_2O-COCH(CH_3)O$, $CH_3NHCOCH(CH_3)O$, $NCCH_2O-COCH_2O$, $CH_3O-COCH(CH_3)S$, $NCCH_2O-COCH(CH_3)S$, $C_2H_5O-CO(CH_2)_3S$,

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$HO-COCH_2NH$, $C_2H_5O-COCH_2NH$ und $C_2H_5O-COCH(CH_3)NH$;

25

g) X F ist, $R^1 CH_3$ ist, $R^2 CH_3SO$ ist, $R^3 Cl$ ist und $R^4 5-NO_2$ ist;

h) X F ist, $R^1 CH_3$ ist, $R^2 CH_3S$ ist, $R^3 Cl$ ist und R^4 gewählt wird aus $5-NH_2$, $5-SH$, $5-CH\equiv CCH_2O$, $5-CH\equiv CCH_2S$ und $5-C_2H_5OOCCH(CH_3)S$;

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i) X F ist, $R^1 CH_3$ ist, $R^2 CHF_2O$ ist, $R^3 Cl$ ist und R^4 gewählt wird aus $HO-COCH_2O$, $CH_3O-COCH_2O$, $CH_2-CHCH_2O-COCH_2O$, $CH\equiv CCH_2O-COCH_2O$, $HO-COCH(CH_3)O$, $CH_2=CHCH_2O-COCH(CH_3)O$, $CH\equiv CCH_2O-COCH(CH_3)O$, $CH_2=CHCH_2S$, $CH_2=C(CH_3)CH_2S$, $CH\equiv CCH_2S$, $C_2H_5OC_2H_4S$, $HO-COCH_2S$, $CH_3O-COCH_2S$, $C_2H_5O-COCH_2S$, $C_2H_5O-COCH_2S(O)$, $n-C_4H_9O-COCH_2S$, $CH_2=CHCH_2O-COCH_2S$, $CH_3SCH_2O-COCH_2S$,

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$CH_3NHCOCH_2S$, $(CH_3)_2NCOCH_2S$, $HO-COCH(CH_3)S$, $CH_3O-COCH(CH_3)S$, $C_2H_5O-COCH(CH_3)S$, $n-C_4H_9O-COCH(CH_3)S$, $CH_2=CHCH_2O-COCH(CH_3)S$, $CH\equiv CH_2O-COCH(CH_3)S$, $CH_3OC_2H_4OC_2H_4O-COCH(CH_3)S$, CH_3SO_2NH , CH_3NHSO_2NH , $C_2H_5O-COCH_2NH$, $C_2H_5O-COCH(CH_3)NH$, $CH_2=CHCH_2O-CO$, $CH\equiv CCH_2O-CO$, $CH_3O-COCH(CH_3)O-CO$, $C_2H_5O-COCH(CH_3)O-CO$, $(CH_3)_2NCO$ und

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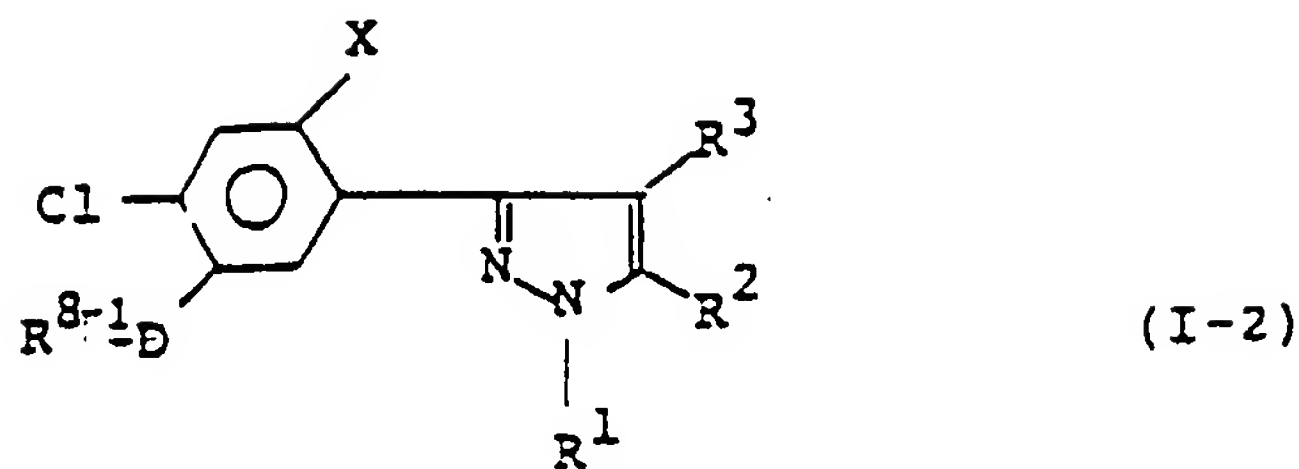


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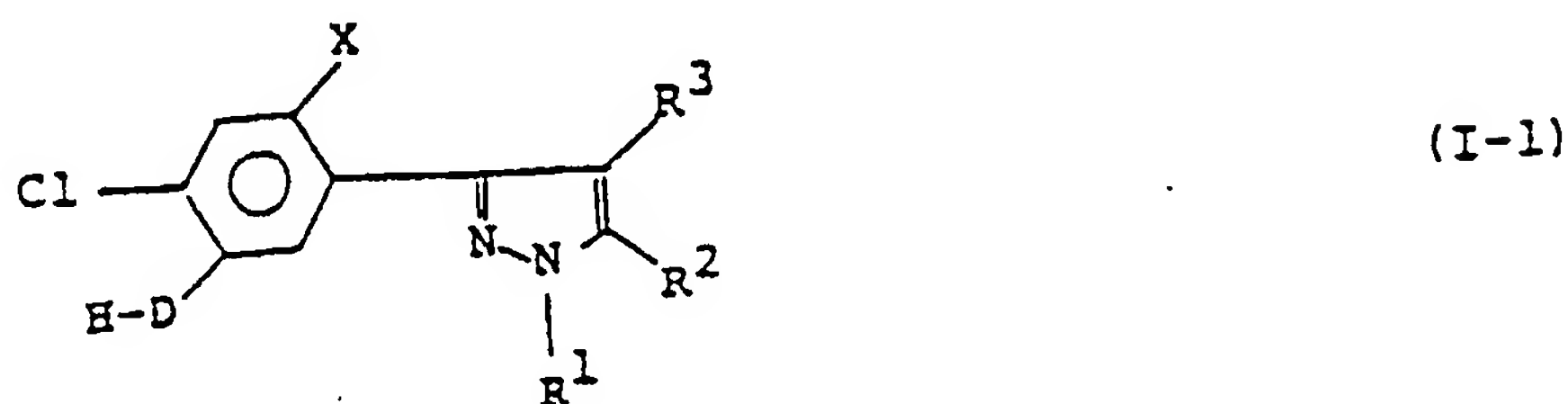
j) X Cl ist, $R^1 CH_3$ ist, $R^2 CHF_2O$ ist, $R^3 Br$ ist und $R^4 BrCH_2CH(Br)CH_2O$ oder $BrC\equiv CCH_2O$ ist.

3. Verfahren zur Herstellung von 3-(substituiertes Phenyl)prazol-Derivaten der allgemeinen Formel (I-2)

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10
 worin X, R¹, R² und R³ definiert sind wie in Anspruch 1, und D-R⁸⁻¹ definiert ist wie D-R⁸ in Anspruch 1 mit dem Unterschied, daß R⁸⁻¹ kein Wasserstoff ist, wobei das Verfahren das Umsetzen eines Pyrazol-Derivats der allgemeinen Formel (I-1)

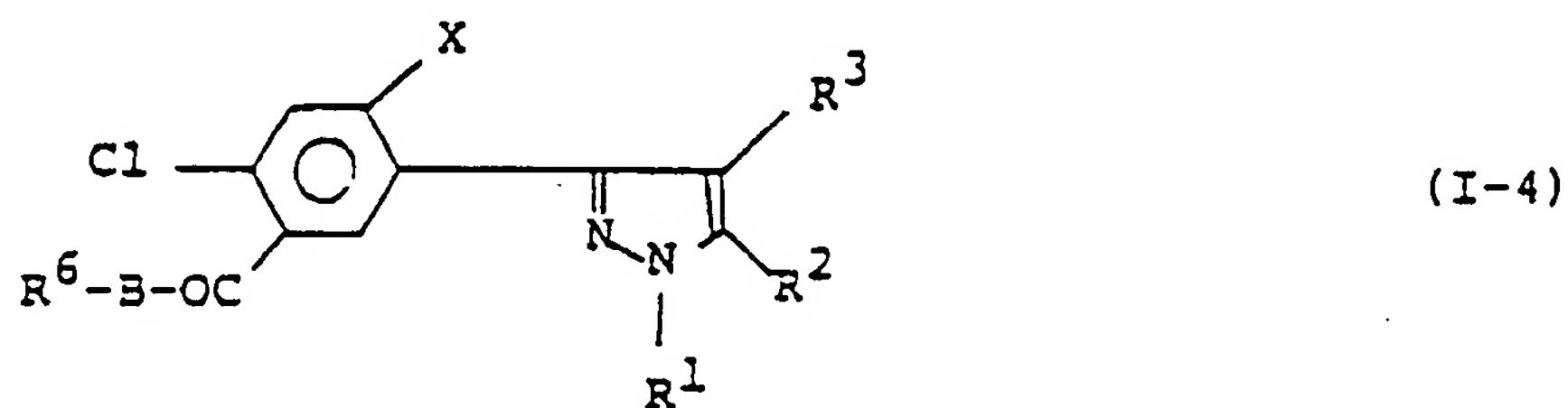


25
 worin X, R¹, R², R³ und D wie oben definiert sind, mit einem Halogenid der allgemeinen Formel (II)

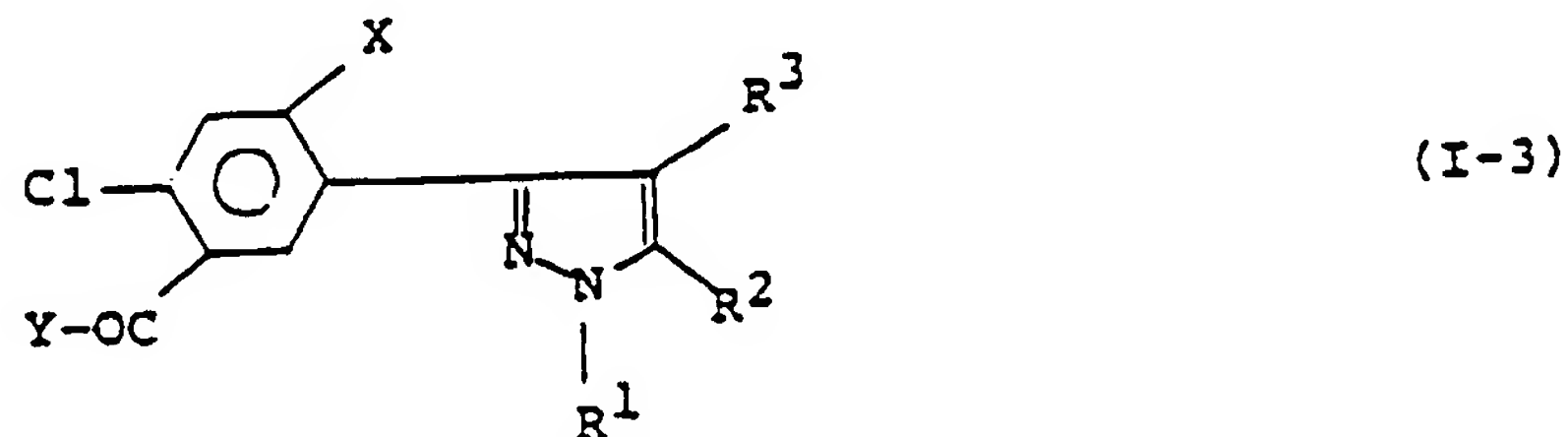


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 worin R⁸⁻¹ definiert ist wie oben und Z Halogen ist, umfaßt.

4. Verfahren zur Herstellung von 3-(substituiertes Phenyl)pyrazol-Derivaten der allgemeinen Formel (I-4)



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 worin X, R¹, R², R³ und -CO-B-R⁶ definiert sind wie in Anspruch 1, wobei das Verfahren das Umsetzen eines Pyrazol-Derivats der allgemeinen Formel (I-3)



worin X, R¹, R² und R³ wie oben definiert sind und Y Halogen oder OH bedeutet, mit einer Verbindung der allge-

meinen Formel (II-1)

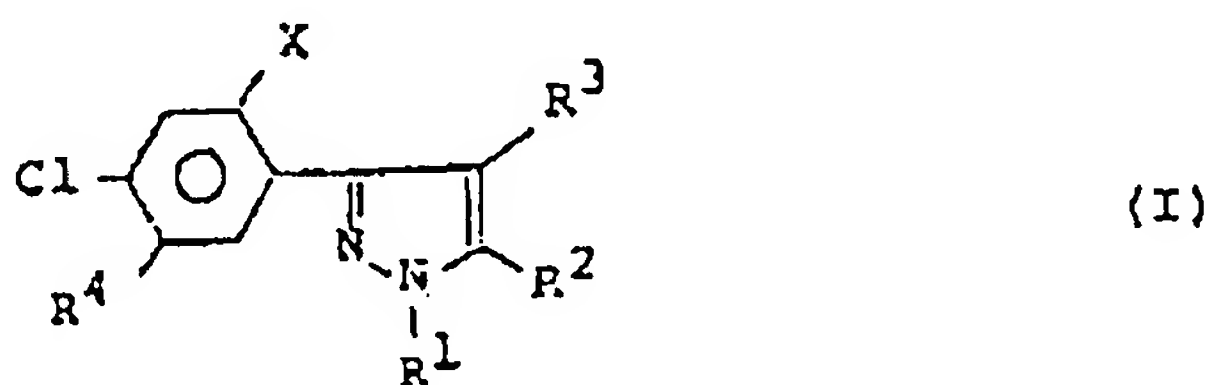


worin R^6 und B wie oben definiert sind, umfaßt.

5. Herbizide Zusammensetzung, die ein 3-(substituiertes Phenyl)pyrazol-Derivat oder ein Salz davon nach Anspruch 1 oder 2 als aktiven Inhaltsstoff enthält.
6. Verwendung der herbiziden Zusammensetzung nach Anspruch 5 für Reisfelder, Hochlandfelder oder für Zwecke in nicht-landwirtschaftlichen Gebieten.
7. Verfahren zur Kontrolle von unerwünschten Pflanzen, welches das Aufbringen einer herbiziden Zusammensetzung, welche ein 3-(substituiertes Phenyl)pyrazol-Derivat oder ein Salz davon nach Anspruch 1 oder 2 als aktiven Inhaltsstoff enthält, in einer Dosis von 1,0 g bis 10 kg, bezogen auf die Menge des aktiven Inhaltsstoffs, pro Hektar umfaßt.

Revendications

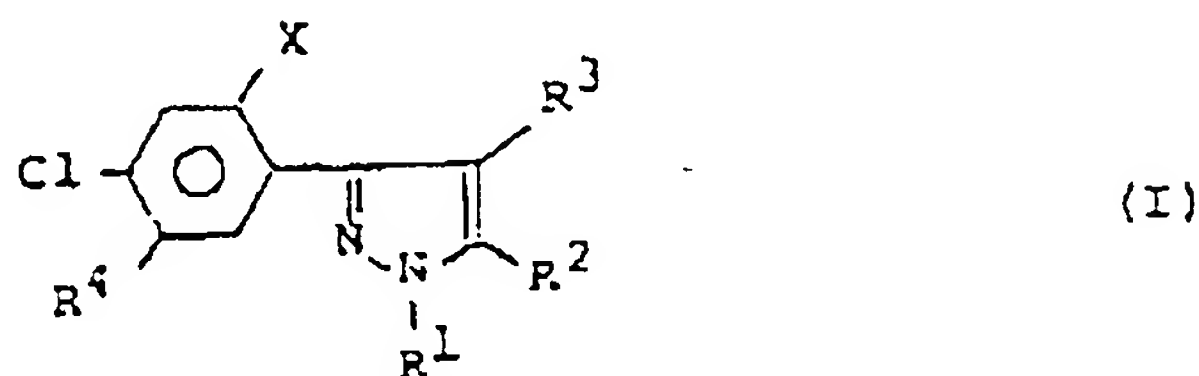
1. Dérivé de 3-(phényl substitué) pyrazole ou sel de celui-ci, le dérivé étant représenté par la formule générale



dans laquelle

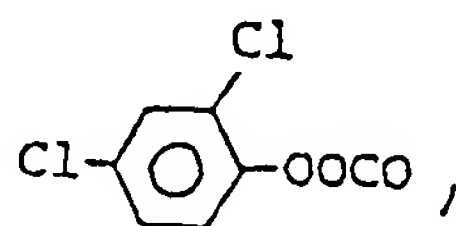
- X signifie un halogène,
 R^1 signifie CH_3 ,
 R^2 signifie un hydroxy, un mercapto, un alkylthio, un haloalcoxy ou un haloalkylthio,
 R^3 signifie l'hydrogène ou un halogène, et
 R^4 signifie (i) le formyle, (ii) le nitro, (iii) $-CO-B-R^6$ dans laquelle B signifie $-O-$, $-S-$, ou $-NR^7$, R^6 et R^7 sont les mêmes ou différents et signifient chacun l'hydrogène ou un cycloalkyle, et lorsque B est $-O-$, R^6 peut être un atome de métal alcalin ou $-CO-B-R^6$ signifie le méthoxycarbonyl, l'éthoxycarbonyl, le propoxycarbonyl, le butoxycarbonyl, le méthoxycarbonylméthoxycarbonyl, l'éthoxycarbonylméthoxycarbonyl, le propoxycarbonylméthoxycarbonyl, le méthoxycarbonyléthoxycarbonyl, l'éthoxycarbonyléthoxycarbonyl, ou le propoxycarbonyléthoxycarbonyl,
 (iv) $-D-R^8$ dans laquelle D signifie $-O-$, $-S(O)_n$, n étant un nombre entier de 0 à 2, ou $-NR^9$, R^8 et R^9 , qui sont les mêmes ou différents, signifient chacun l'hydrogène ; un alkyle ; un haloalkyle ; un aminosulfonyl ; un phénylalkyle ou un phénoxy-alkyle ayant éventuellement sur l'anneau phényle un ou plusieurs substituants halogène qui sont les mêmes ou différents ; ou $-D-R^8$ signifie un groupe alcényloxy contenant le groupe propényle, butényle ou pentényle ; un groupe alcényloxy contenant le groupe propynyle, butynyle ou pentynyle ; un groupe alcénylamino contenant le groupe propényle, butényle ou pentényle ; un groupe alcoxy-carbonylalcoxy ou alkylthiocarbonylalcoxy contenant chacun le méthyle, l'éthyle, le propyle ou le butyle ; ou (v) $-(CH_2R^{10})_m-CO-E-R^{12}$ dans laquelle E signifie $-O-$, $-S-$ ou $-NR^{11}$; R^{10} signifie l'hydrogène et R^{11} et R^{12} , qui sont les mêmes ou différents, signifient chacun l'hydrogène ; un alkyle ; un haloalkyle ; un phényle ayant éventuellement sur l'anneau phényle un ou plusieurs substituants halogène ; un phénylalkyle ayant éventuellement sur l'anneau phényle un ou plusieurs substituants halogène ; ou R^{11} , conjointement avec R^{12} forme un groupe pipéridino ou morpholinyle, et lorsque E est $-O-$, R^{12} peut être un atome de métal alcalin ; et m signifie un nombre entier de 0 à 3.

2. Dérivé de 3-(phényl substitué) pyrazole ou sel de celui-ci, le dérivé étant représenté par la formule générale

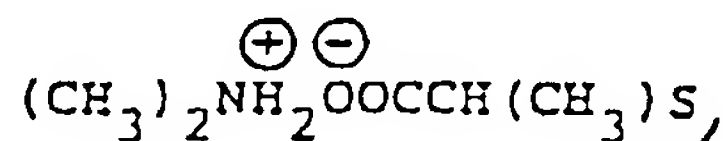
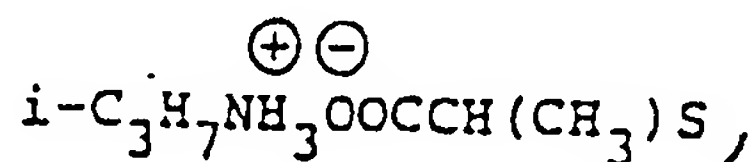


dans laquelle

a) X est Cl, R¹ est CH₃, R² est CH₃S, R³ est Cl et R⁴ est sélectionné parmi

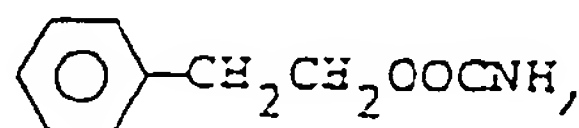


HOOCCH(CH₃)O, CH₃OOCCH(CH₃)O, C₂H₅OOCCH(CH₃)O, i-C₃H₇OOCCH(CH₃)O, CH₂=CHCH₂OOCCH(CH₃)O, CH≡CCH₂OOCCH(CH₃)O, CH₂=CHCH₂S, CH₂=C(Cl)CH₂S, CH≡CCH₂S, HO₂CCH(CH₃)S,



35 Cl(CH₂)₃OOCCH(CH₃)S, CH₂=CHCH₂OOCCH(CH₃)S, CH≡CCH₂OOCCH(CH₃)S, C₂H₅CH(CH₃)CH₂OOCCH(CH₃)S, (C₂H₅)₂NCOCH(CH₃)S, CH₃NHSO₂, (CH₃)₂NSO₂, n-C₃H₇NHSO₂, CH₃SO₂NHSO₂, Cl⁻H₃N⁺, CH≡CCH₂NH, CH₃OOCNH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃NHSO₂N(CH₃), (CH₃)₂NSO₂N(CH₃), i-C₃H₇NHSO₂NH, CH≡CCH₂N(CH₃)SO₂N(CH≡CCH₂), CH₃OCH₂CH₂OOCNH, BrCH₂CH₂OOCNH, NCCH₂CH₂OOCNH, CH₂=C(CH₃)CH₂OOCNH,

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45 CH₂=CHCH₂OOCCH₂NH, CH≡CCH₂OOCCH₂NH, C₂H₅OOCCH(CH₃)NH, HOOCCH₂NH, CH₃OOCCH₂NH, HOOCCH(CH₃)NH et CH₃OOCCH(CH₃)NH;

b) X est Cl, R¹ est CH₃, R² est CH₃SO, R³ est Cl et R⁴ est sélectionné parmi NO₂, CH₃SO₂NH, CH₃NHSO₂NH, i-C₃H₇NHSO₂NH, et CH₃NHSO₂(CH≡CCH₂);

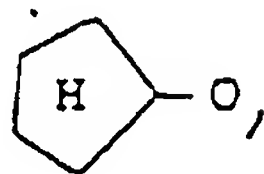
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c) X est Cl, R¹ est CHF₂, R² est CH₃S, R³ est Cl et R⁴ est sélectionné parmi NH₂, CH₃NHSO₂NH, CH₃NHSO₂N(CH≡CCH₂), et CH₃N(CH≡CCH₂)SO₂N(CH≡CCH₂);

d) X est Cl, R¹ est CHF₂, R² est CH₃SO, R³ est Cl et R⁴ est sélectionné parmi NH₂ et NO₂;

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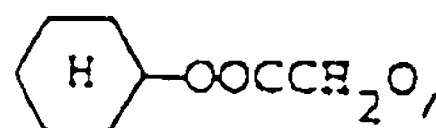
e) X est Cl, R¹ est CH₃, R² est CHF₂O, R³ est Cl et R⁴ est sélectionné parmi



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$C_2H_5OCH_2CH_2O$, $NCCH_2O$, CH_3OOCO , $n-C_6H_{13}OOCCH_2O$, $CH_2=CHCH_2OOCCH_2O$,
 $CH\equiv CCH_2OOCCH_2O$,

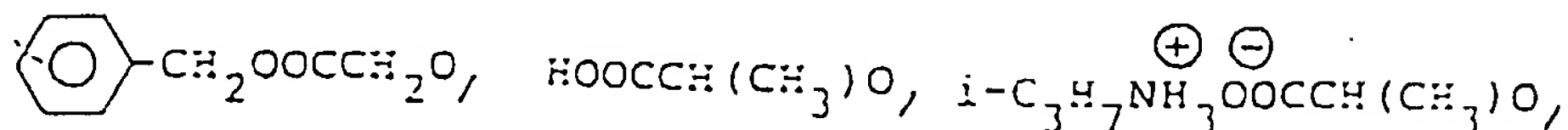
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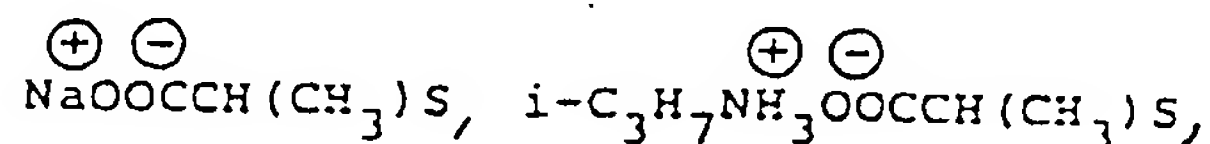
$C_2H_5O(CH_2)_2OOCCH_2O$, $C_2H_5O(CH_2)_2O(CH_2)_2OOCCH_2O$,

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$CH_2=CHCH_2OOCCH(CH_3)O$, $CH\equiv CCH_2OOCCH(CH_3)O$, $CH_3NHCOCH(CH_3)O$, $(CH_3)_2NCOCH(CH_3)O$,
 $CH_2=CHCH_2S$, $CH_2=CHCH_2SO$, $CH\equiv CCH_2S$, $(C_2H_5O)_2P(O)CH_2S$, CH_3OOC , CH_3OOCCH_2S ,
 $C_2H_5OOCCH_2S$, $CH_2=CHCH_2OOCCH_2S$, $HOOCCH(CH_3)S$,

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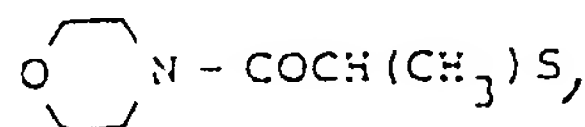
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$CH_3OOCCH(CH_3)S$, $C_2H_5OOCCH(CH_3)S$, $C_2H_5OOCCH(CH_3)S(O)$, $C_2H_5OOCCH(CH_3)SO_2$, $i-C_3H_7OOCCH$
 $(CH_3)S$, $n-C_4H_9OOCCH(CH_3)S$, $(CH_3)_2CH(CH_2)_2OOCCH(CH_3)S$, $CH_2=CHCH_2OOCCH(CH_3)S$,
 $CH\equiv CCH_2OOCCH(CH_3)S$, $CH_3O(CH_2)_2O(CH_2)_2OOCCH(CH_3)S$, $(C_2H_5O)_2P(O)CH_2OOCCH(CH_3)S$,
 $CH_3NHCOCH(CH_3)S$, $(CH_3)_2NCOCH(CH_3)S$, $(C_2H_5)_2NCOCH(CH_3)S$, $i-C_3H_7NHCOCH(CH_3)S$,

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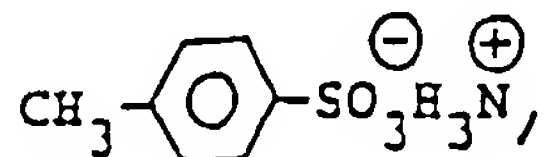


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CH_3SO_2NH , CH_3NHSO_2NH , $CH_2=CHCH_2NH$, $CH\equiv CCH_2NH$, CH_3OOCNH , $CH_3H_3N^+$,

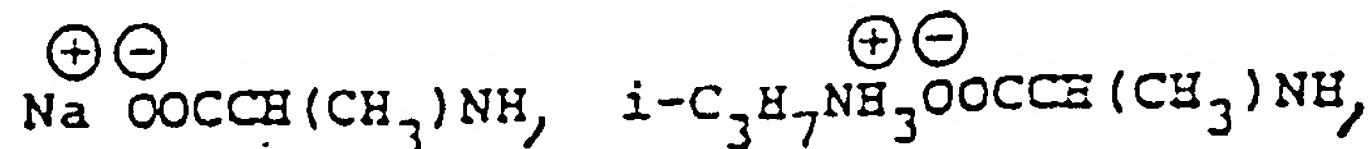
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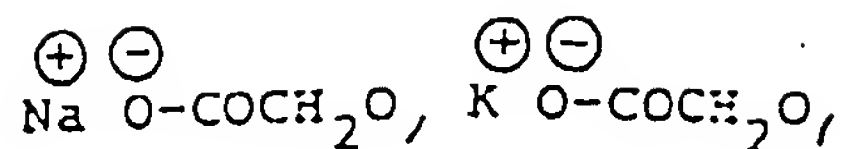
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CH_3OOCCH_2NH , $HOOCCH(CH_3)NH$,

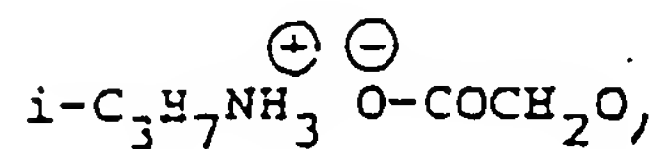
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$C_2H_5OOCCH(CH_3)N$, $CH_3NHOCCH(CH_3)NH$, $(CH_3)_2NOCCH(CH_3)NH$, CH_3NHSO_2 , $(CH_3)_2NSO_2$,
 $(C_2H_5)_2NCOS$, $(CH_3)_3SiCH_2S$, $(CH_3)_3SiCH_2O$, $HO-COCH_2O$,



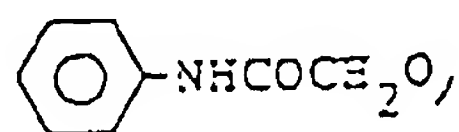
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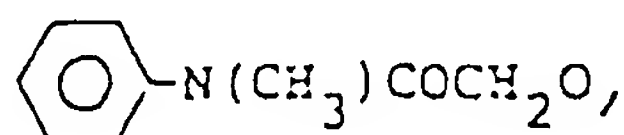
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$\text{CH}_2=\text{CHC}(\text{CH}_3)_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{C}\equiv\text{CCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{NHCOCH}_2\text{O}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{O}$, n-
 $\text{C}_4\text{H}_9\text{NHCOCH}_2\text{O}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{O}$, $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{O}$,

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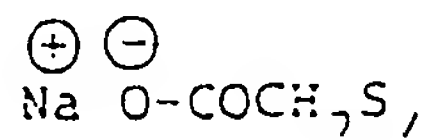


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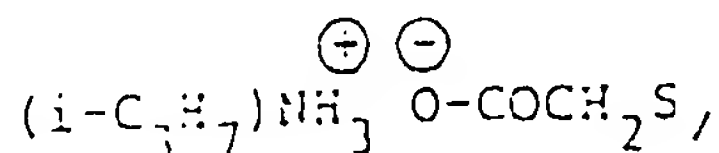


$\text{CH}_3\text{SO}_2\text{NHCOCH}_2\text{O}$, $\text{CH}_3\text{SO}_2\text{NHCOCH}(\text{CH}_3)\text{O}$, $(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{O}$, $\text{CH}_3\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}$
 $(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{S}$, $\text{CH}_2=\text{C}(\text{Cl})\text{CH}_2\text{S}$, $\text{ClCH}=\text{CHCH}_2\text{S}$, $\text{ClCH}=\text{C}(\text{Cl})\text{CH}_2\text{S}$, NCCH_2S ,

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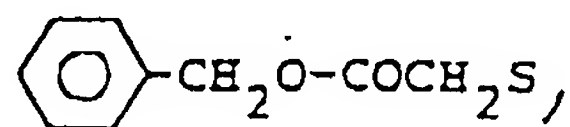
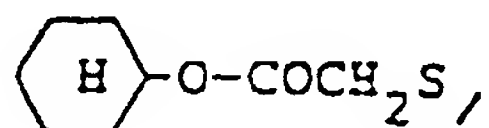
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$\text{HO-COCH}_2\text{S}$, $\text{i-C}_3\text{H}_7\text{O-COCH}_2\text{S}$, $\text{n-C}_4\text{H}_9\text{O-COCH}_2\text{S}$, $\text{C}_{12}\text{H}_{25}\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{S-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{S-COCH}_2\text{S}$,
 $\text{i-C}_3\text{H}_7\text{S-COCH}_2\text{S}$, $\text{CH}\equiv\text{CCH}_2\text{O-COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O-COCH}_2\text{S}$,

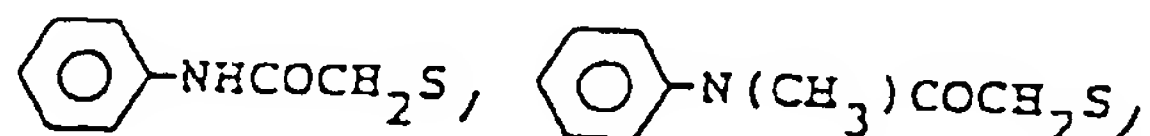
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$(\text{CH}_3)_3\text{SiCH}_2\text{O-COCH}_2\text{S}$, $\text{CH}_3\text{NHCOCH}_2\text{S}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{S}$, $\text{H}_2\text{C}=\text{CHCH}_2\text{NHCOCH}_2\text{S}$,
 $(\text{H}_2\text{C}=\text{CHCH}_2)_2\text{NCOCH}_2\text{S}$, $\text{HC}\equiv\text{CCH}_2\text{NHCOCH}_2\text{S}$,

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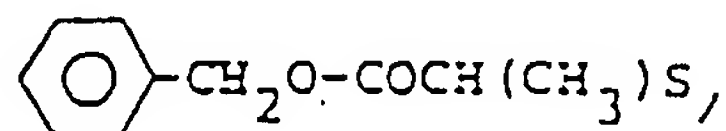


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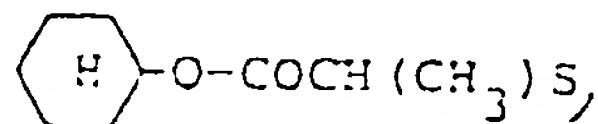
i-C₄H₉O-COCH(CH₃)S, t-C₄H₉CH₂O-COCH(CH₃)S, n-C₆H₁₃O-COCH(CH₃)S,

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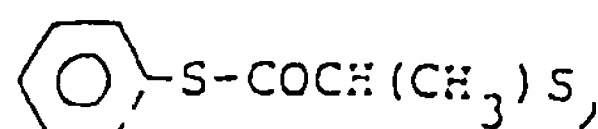


C₂H₅S-COCH(CH₃)S, i-C₃H₇S-COCH(CH₃)S, NCCH(CH₃)O-COCH(CH₃)S, CH₃SC₂H₄O-COCH(CH₃)S,

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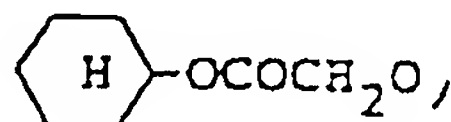
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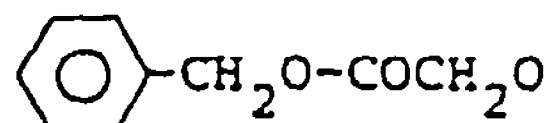
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(CH₃)₃SiCH₂O-COCH(CH₃)S, CH₂=CHCH₂NHCOCH(CH₃)S, (H₂C=CHCH₂)₂NCOCH(CH₃)S, HC≡CCH₂NHCOCH(CH₃)S, CH₃CO-O-(CH₂)₂S, C₂H₅O-COC(CH₃)₂S, CH₃O-COCH(i-C₃H₇)S, C₂H₅O-COCH(CH₃)N(CH₃), CH₂=CHCH₂O-CO, CH₃O-COCH(CH₃)O-CO, C₂H₅O-COCH(CH₃)O-CO, CH≡CCH₂O-CO,

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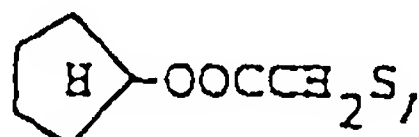


et CH₃O-CO-OCH₂S-CO;

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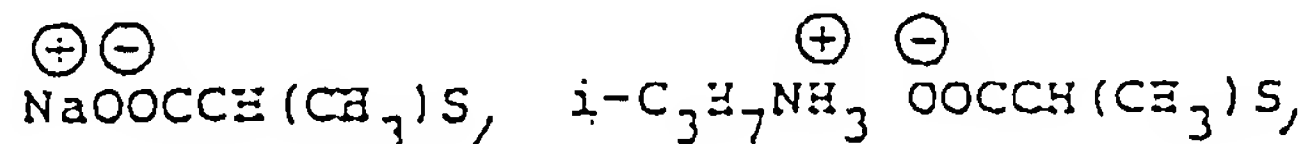
f) X est Cl, R¹ est CH₃, R² est CHF₂S, R³ est Cl et R⁴ est sélectionné parmi CH₂=CHCH₂S, CH≡CCH₂S, C₂H₅OCH₂CH₂S, CH₃SCH₂S, NCCH₂S, C₂H₅OOCs, CH₃OOCCH₂S, C₂H₅OOCCH₂S,

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CH₃NHOCCH₂S, (CH₃)₂NOCCH₂S, HOOCCH(CH₃)S,

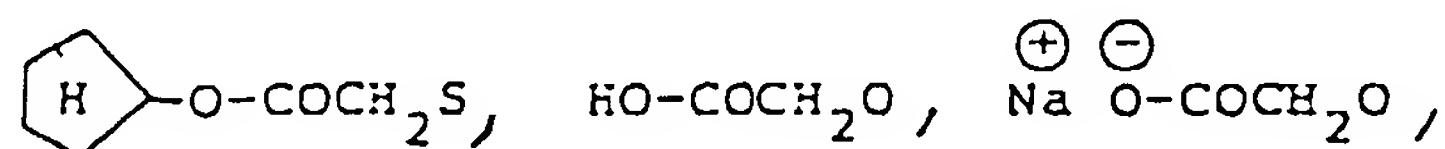
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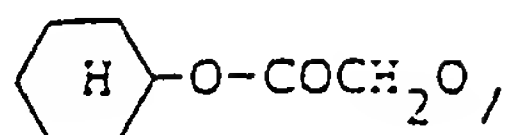
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C₂H₅OOCCH(CH₃)S, n-C₃H₇OOCCH(CH₃)S, i-C₃H₇OOCCH₂S, (CH₃)₂CHCH₂OOCCH₂S, CH₂=CHCH₂OOCCH(CH₃)S, C₂H₅OOC(CH₂)₃S, CH₃NHOCCH(CH₃)S, (CH₃)₂NOCCH(CH₃)S, CH≡CCH₂NH, CH₃SO₂NH, CH₃NHSO₂NH, CH₃OOCNH, (C₂H₅)₂NCOS, NCCH₂O, ClCH=CHCH₂O, CH₃OCH₂O, C₂H₅OC₂H₄O, CH₃OCH₂S, ClCH=CHCH₂S, CH₂=C(Cl)CH₂S,

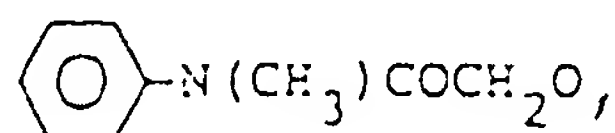
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$\text{CF}_3\text{CH}_2\text{O}-\text{COCH}_2\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}_2\text{O}$,



$\text{CH}_3\text{NHCOCH}_2\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{O}$, $\text{CH}_3\text{NHCOCH}(\text{CH}_3)\text{O}$, $\text{NCCH}_2\text{O}-\text{COCH}_2\text{O}$, $\text{CH}_3\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{NCCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{C}_2\text{H}_5\text{O}-\text{CO}(\text{CH}_2)_3\text{S}$,



$\text{HO}-\text{COCH}_2\text{NH}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}_2\text{NH}$ et $\text{C}_2\text{H}_5\text{O}-\text{COCH}(\text{CH}_3)\text{NH}$;

g) X est F, R^1 est CH_3 , R^2 est CH_3SO , R^3 est Cl et R^4 est 5- NO_2 ;

h) X est F, R^1 est CH_3 , R^2 est CH_3S , R^3 est Cl et R^4 est sélectionné parmi 5- NH_2 , 5- SH , 5- $\text{CH}\equiv\text{CCH}_2\text{O}$, 5- $\text{CH}\equiv\text{CCH}_2\text{S}$ et 5- $\text{C}_2\text{H}_5\text{OOCCH}(\text{CH}_3)\text{S}$;

i) X est F, R^1 est CH_3 , R^2 est CHF_2O , R^3 est Cl et R^4 est sélectionné parmi $\text{HO}-\text{COCH}_2\text{O}$, $\text{CH}_3\text{O}-\text{COCH}_2\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}_2\text{O}$, $\text{CH}\equiv\text{CCH}_2\text{O}-\text{COCH}_2\text{O}$, $\text{HO}-\text{COCH}(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{O}$, $\text{CH}\equiv\text{CCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{O}$, $\text{CH}_2=\text{CHCH}_2\text{S}$, $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{S}$, $\text{CH}\equiv\text{CCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{OC}_2\text{H}_4\text{S}$, $\text{HO}-\text{COCH}_2\text{S}$, $\text{CH}_3\text{O}-\text{COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}_2\text{S}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}_2\text{S}(\text{O})$, $n\text{-C}_4\text{H}_9\text{O}-\text{COCH}_2\text{S}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}_2\text{S}$, $\text{CH}_3\text{SCH}_2\text{O}-\text{COCH}_2\text{S}$,

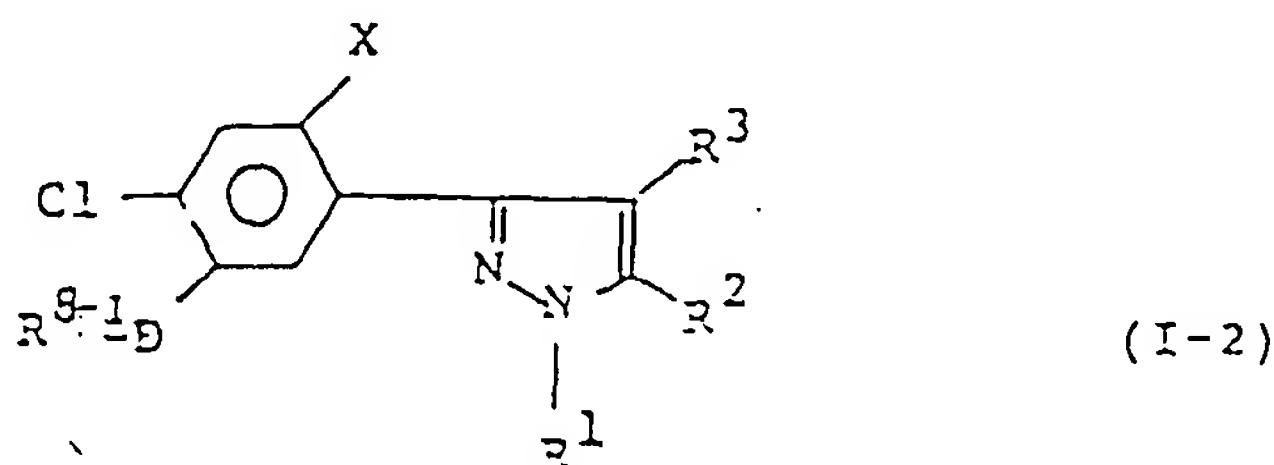


$\text{CH}_3\text{NHCOCH}_2\text{S}$, $(\text{CH}_3)_2\text{NCOCH}_2\text{S}$, $\text{HO}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $n\text{-C}_4\text{H}_9\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{CH}\equiv\text{CCH}_2\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{OC}_2\text{H}_4\text{OC}_2\text{H}_4\text{O}-\text{COCH}(\text{CH}_3)\text{S}$, $\text{CH}_3\text{SO}_2\text{NH}$, $\text{CH}_3\text{NHSO}_2\text{NH}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}_2\text{NH}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}(\text{CH}_3)\text{NH}$, $\text{CH}_2=\text{CHCH}_2\text{O}-\text{CO}$, $\text{CH}\equiv\text{CCH}_2\text{O}-\text{CO}$, $\text{CH}_3\text{O}-\text{COCH}(\text{CH}_3)\text{O}-\text{CO}$, $\text{C}_2\text{H}_5\text{O}-\text{COCH}(\text{CH}_3)\text{O}-\text{CO}$, $(\text{CH}_3)_2\text{NCO}$ et

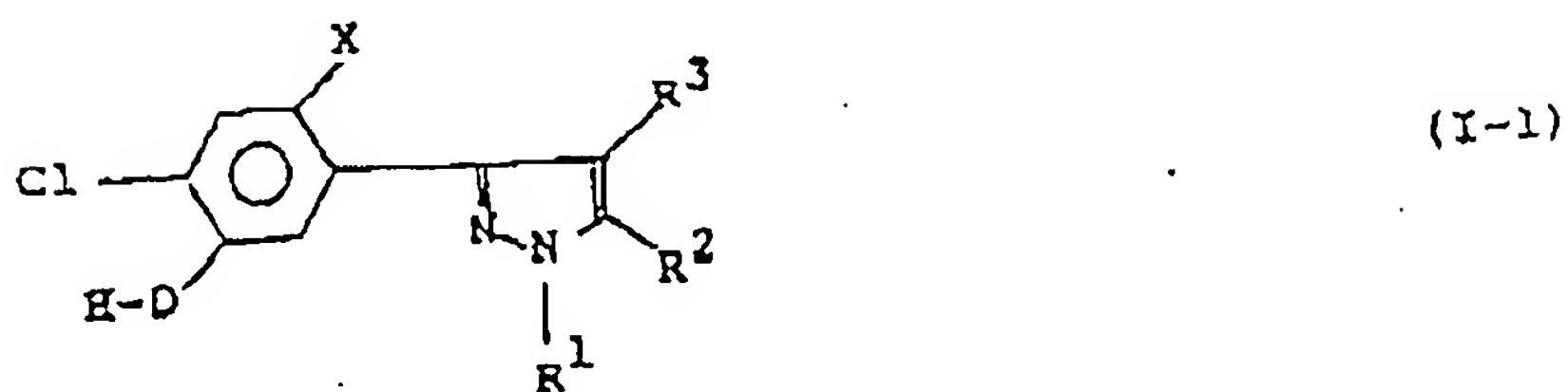


j) X est Cl, R^1 est CH_3 , R^2 est CHF_2O , R^3 est Br et R^4 est $\text{BrCH}_2\text{CH}(\text{Br})\text{CH}_2\text{O}$ ou $\text{BrC}\equiv\text{CCH}_2\text{O}$.

3. Procédé de production de dérivés de 3-(phényl substitué) pyrazole représentés par la formule générale



dans laquelle X, R^1 , R^2 et R^3 sont tels que définis dans la revendication 1, et D-R^{8-1} est défini comme D-R^8 en revendication 1 sauf que R^{8-1} n'est pas l'hydrogène, le procédé comprenant la réaction d'un dérivé de pyrazole représenté par la formule générale

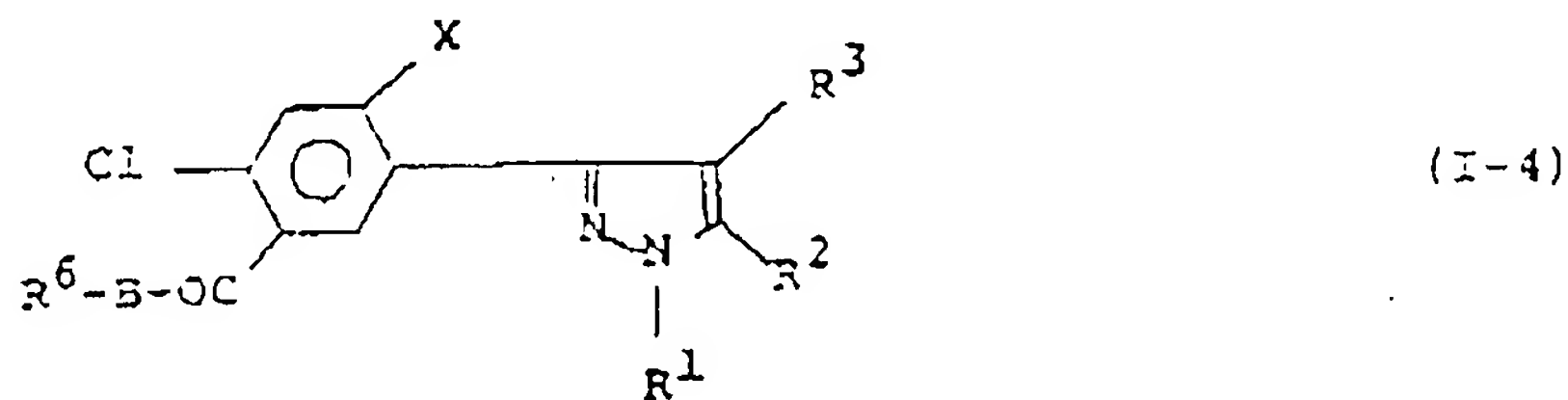


10 dans laquelle X, R¹, R², R³ et D sont tels que définis ci-dessus, avec un halogénure représenté par la formule générale

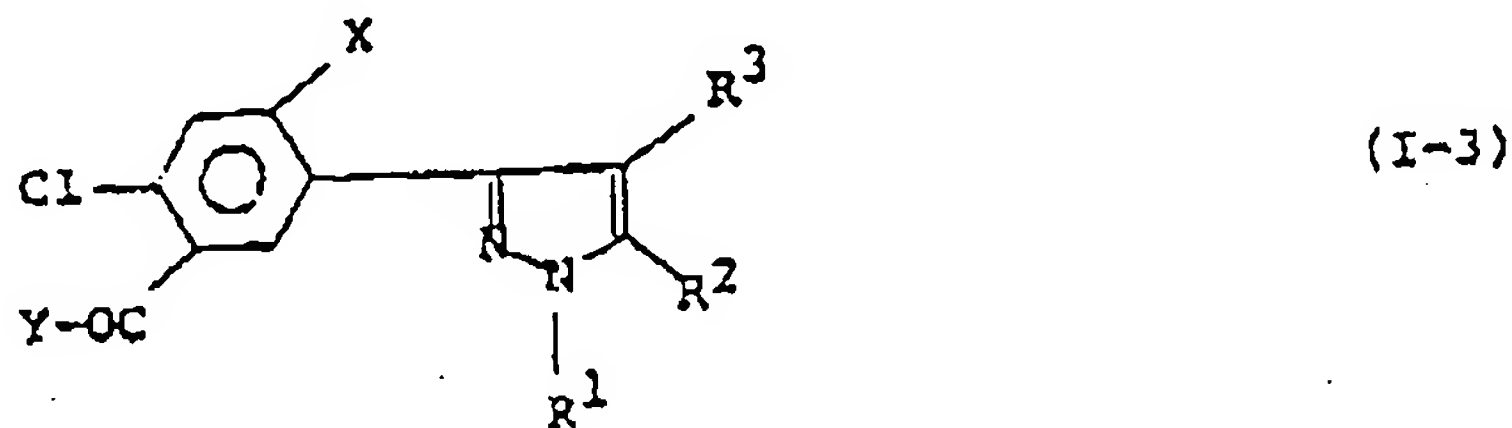


dans laquelle R⁸⁻¹ est défini ci-dessus et Z est un halogène.

- 20 4. Procédé de production de dérivés de 3-(phényl substitué) pyrazole représentés par la formule générale



30 dans laquelle X, R¹, R², R³ et -CO-B-R⁶ sont tels que définis dans la revendication 1, le procédé comprenant la réaction d'un dérivé de pyrazole représenté par la formule générale



45 dans laquelle X, R¹, R² et R³ sont tels que définis ci-dessus et Y signifie un halogène ou OH, avec un composé représenté par la formule générale



50 dans laquelle R⁶ et B sont tels que définis ci-dessus.

- 55 5. Composition herbicide qui contient un dérivé de 3-(phényl substitué) pyrazole ou un sel de celui-ci selon la revendication 1 ou la revendication 2 comme ingrédient actif.
6. Utilisation de la composition herbicide de la revendication 5 pour l'application à des rizières, des champs de hautes terres ou des zones non agricoles.

7. Procédé pour réguler des plantes indésirables qui comprend l'application d'une composition herbicide contenant un dérivé de 3-(phényl substitué)pyrazole ou un sel de celui-ci selon la revendication 1 ou la revendication 2 comme ingrédient actif, selon une dose de 1,0 g à 10 kg en termes de quantité d'ingrédient actif par hectare.

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